

Phytochemical Screening for Various Secondary Metabolites, Antioxidant, and Anthelmintic Activity of *Coscinium fenestratum* Fruit Pulp: A New Biosource for Novel Drug Discovery

Coscinium fenestratum Meyvelerinin Farklı Sekonder Metabolitleri Üzerinde Fitokimyasal Analiz Çalışmaları, Antioksidan ve Antihelmintik Aktiviteleri: Yeni İlaçların Keşfinde Yeni Bir Doğal Kaynak

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ABSTRACT

Objectives: Coscinium fenestratum (Gaertn.) Colebr. (CF, Family: Menispermaceae) is an important endangered woody climber in India. CF contains various major secondary metabolites for the treatment of various disease conditions. The present study aimed to establish the antioxidant and anthelmintic activity of Coscinium fenestratum fruit pulp.

Materials and Methods: The dried fruit pulp was subjected to aqueous, methanol, and mixed aqueous and methanol (1:1) solvent extraction followed by phytochemical investigations, estimations of alkaloids, phenolics, flavonoids, antioxidant potentiality (DPPH and hydrogen peroxide scavenging methods), and anthelmintic activity tests were carried out.

Results: Preliminary phytochemical screening of CF fruit extracts revealed the presence of alkaloids phenols, flavonoids, tannins, steroids, and resins, which are responsible for biologic properties. The combined aqueous and methanol extract resulted in significant anthelmintic and antioxidant properties in a dose-dependent manner. The DPPH free radical scavenging assay and H_2O_2 assay exhibited IC₅₀ values of 42.38±0.012 µg/mL and 46.80±0.011 µg/mL, respectively. Thereafter, the anthelmintic activity test was carried out against *Pheretima posthuma* and *Taenia solium* with the extract at varying concentrations of 25, 50, 100 and 150 mg/mL and compared with standard albendazole (25 and 50 mg/mL) and saline (0.9%) as a control. All the extracts exhibited concentration-dependent paralytic effect, followed by death on the test organism, but significant activity was observed with the combined methanol and aqueous extract.

Conclusion: The study was conducted in order to find possible isolated compounds as a biosources for future novel antioxidants in food and pharmaceutical formulations. Our findings indicate for the first time that the CF fruit pulp has therapeutic values with prominent antioxidant and anthelmintic properties.

Key words: Antioxidant study, anthelmintic activity, Coscinium fenestratum, extracts, phytochemical study

ÖΖ

Amaç: Coscinium fenestratum (Gaertn.) Colebr. (CF, Familya: Menispermaceae) Hindistan'da önemli bir odunsu bitkidir. Bitki çeşitli hastalıkların tedavisinde etkili olan çeşitli majör sekonder metabolitler içermektedir. Bu çalışmada, Coscinium fenestratum meyvelerinin antioksidan ve antihelmintik aktiviteleri değerlendirilmiştir.

Gereç ve Yöntemler: Bu çalışmada, kurutulmuş meyvelerin sulu, metanollü ve sulu metanollü (1:1) ekstreleri hazırlanmış ve ekstrelerin alkaloit, fenolik, flavonoid içerikleri ile antioksidan (DPPH ve hidrojen peroksit süpürücü etki tayin yöntemi) ve antihelmintik aktiviteleri değerlendirilmiştir.

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Bulgular: Ön fitokimyasal tarama çalışmaları, CF meyve ekstrelerinin biyolojik etkiden sorumlu alkaloitler, fenoller, flavonoidler, tanenler, steroidler ve reçine içerdiğini ortaya koymuştur. Sulu metanollü ekstrenin doz bağımlı olarak anlamlı derecede antihelmintik ve antioksidan etkiye sahip olduğu belirlenmiştir. DPPH serbest radikal süpürücü aktivite tayini ve H_2O_2 deneyinde IC_{50} değerleri sırasıyla 42.38±0.012 µg/mL ve 46.80±0.011 µg/mL olarak belirlenmiştir. Antihelmintik aktivite, *Pheretima posthuma* ve *Taenia solium*'a karşı ekstrenin 25, 50, 100 ve 150 mg/mL gibi değişen konsantrasyonlarında, standart albendazol (25 and 50 mg/mL) ve kontrol olarak kullanılan tuzlu suya (%0.9) kıyasla denemiştir. Tüm ekstreler doz bağımlı paralitik etki göstermiş, organizmalar üzerinde anlamlı derecede öldürücü etki, sulu metanollü ekstre için tespit edilmiştir.

Sonuç: Bu çalışmanın bulguları, CF meyve ekstresinin antioksidan ve antihelmintik etkileri ile tedavi edici etkiye sahip olduğunu ilk defa göstermiş, izole edilen bileşiklerin gıda ve farmasötik formülasyonlarda yeni antioksidan doğal kaynaklar olarak kullanılabileceğini ortaya koymuştur.

Anahtar kelimeler: Antioksidan çalışma, antihelmintik etki, Coscinium fenestratum, ekstreler, fitokimyasal çalışma

INTRODUCTION

Secondary metabolites are important plant constituents for effective therapeutic activities. It was reported that the presence of this specific group of compounds showed specific medicinal actions and sometimes traditionally reported, but there is little by way of scientific validation. Among these activities, antioxidant and anthelmintic activities are very important. Rapid production of free radicals leads to oxidative damage to biomolecules and results in serious disorders viz. degenerative disorders, cancer, diabetes, neural disorders, and ageing, and hence antioxidant plays a vital role to block free radical production.^{1,2} Moreover, infections with parasitic worms are serious problems for humans, producing various diseases worldwide, and helminthes is one of them. There are various types of worms viz. round worms or nematodes including intestinal worms, filarial worms that cause lymphatic filariasis, chocerciasis, platyhelminths or flatworms including flukes, and tapeworms.³ These worms cause lymphatic filariasis, onchocerciasis, cysticercosis, malnutrition, anemia, eosinophilia, and pneumonia, which are life-threatening. As per the report of the World Health Organization, more than two billion people have parasitic worm infections globally.⁴ Treatment with synthetic drugs causes many adverse effects and helminthes become resistant. Hence, there are demands for natural plant secondary metabolites in the treatment and prevention of this chronic problem.

Coscinium fenestratum (Gaertn.) Colebr. (CF) is a woody climber that belongs to the family Menispermaceae. The plant is commonly known as tree turmeric or false Calumbadue due to its yellow stem. CF is found in Asian countries such as India, Malaysia, Vietnam, Myanmar, Singapore, Thailand, and Sri Lanka.^{5,6} In India, the plant is endangered and located in the Western Ghats areas, especially in high rainfall evergreen forests of Karnataka, Kerala, and Tamil Nadu at altitudes of 500-750 m.^{7,8} The tree requires long seed germination times and takes 14-15 years to mature and flower. Hence, the fruits and seeds are very rare and this leads to endangered red- labeled species due to over exploitation from natural habitats, zero cultivation planning, and trees being uprooted before their reproduction stage for their medicinal importance.⁴ The leaves and roots are traditionally used for the treatment of ulcers, skin diseases, eye disorders, inflammation, hypertension, jaundice, diabetes, and snake bites.9-11 Multiple beneficial pharmacologic-related properties with various solvent extracts of the leaves and roots of CF have been reported viz. hepatoprotective, immune protective, hypoglycemic, anti-tumor

activities, dressing wounds, ulcer treatment, and for cutaneous leishmaniasis, and it is non-toxic to mammals.^{12,13} Stem and root extracts have also shown antioxidant and antimicrobial potential.¹⁴ Traditionally CF is used as one of the ingredients in several ayurvedic preparations such as soap, bath gels, face wash and bath oil, and in the cosmetic industry as facial masks. fairness creams, and body lotions.⁴ Furthermore, stem extract of CF was reported to have a significant effect on stimulating insulin secretion.¹⁵ These activities are due to the presence of the important alkaloid-containing phytoconstituents such as berlambine, dihydroberlambine, noroxyhydrastine, berberine(an isoquinoline alkaloid) and other constituents such asceryl-alcohol, saponin, hentriacontane, sitosterol glucoside, palmitic acid, and oleic acid, which are isolated from the stem and roots of the plants.¹⁶ Recently ecdysterone was identified and isolated from the stem and leaves of CF and evaluated using high-performance liquid chromatography and liquid chromatography-mass spectrometry.¹⁷ The availability of various phytoconstituents in fruits is unknown and hence it is required to select the various solvent extractions for the fruits for further processing. It was reported that the constituents varied with solvent extraction and the zone of collection of the raw materials.¹⁸⁻²⁰ Till now, no literature has revealed the medicinal importance of the fruit and the pulp constituents, perhaps due to improper collection or the low availability of the fruits and seeds. It is also essential to understand the cause of delayed germination. Based on the availability of secondary metabolites in the plant, the present study was conducted to determine and evaluate the phytoconstituents present and novel antioxidant and anthelmintic activities were assessed for the first time from various extracts of the dried fruit pulp.

MATERIALS AND METHODS

Collection and identification of fruits

One hundred CF fruits were collected from Dr. Gokul S, CIMAP Research Centre, Allalasandra, GKVK Post, Bangalore -65 (Latitude: 12° 58' N and Longitude: 77° 38' E), and authenticated by Dr. P.E. Rajasekharan, Principal Scientist, IIHR, Bangalore. The fruits are stored as herbarium in Pharmacognosy Department of Krupanidhi College of Pharmacy, Bangalore (Herbarium No: CF-317/KCP/2016-17).

Morphological study of fruits and seeds

Fifty fruits were randomly selected and measured for diameter using *Vernier calipers* with the measurement readings in centimeters (cm), with precision up to 2 decimal places.

Thereafter, seed diameters were also measured after removal of pulp and resinous matter to understand the probability for late germination. In addition, a complete morphologic study was performed, observing color, odor, size, shape, and extra features.

Preparation of extracts

Fresh fruits of the CF plant were shade-dried for several days and in between observed for fungal infections. The dried pulps were ground to a course powder and 250 g of the same underwent soxhlet extraction with light petroleum ether for 4 hrs and defatting the materials. The pulps were successively extracted with four solvents viz. chloroform, methanol (80%), aqueous, and an equal ratio mixture of aqueous and methanol (80%) (1:1). The reflux method was used for all extracts separately for 7-8 hrs (after drying after each extraction) preparation and finally the yield was calculated after removal of the solvents by rotary evaporation (at 45°C) and the dried extract was stored in a refrigerator (at 4-5°C) for further investigations.

Phytochemical screening

The preliminary phytochemical analysis of the plant extracts was performed using the standard protocol as describe by Khandelwal²¹, and Kokate²², to identify the presence of alkaloids, flavonoids, steroids, glycosides, cardiac glycosides, anthraquinones, tannins, and saponins.

Based on the presence of phytoconstituents, the following estimations of secondary metabolites were carried out:

Determination of alkaloid

Onemilligram of each plant extract was dissolved separately in dimethyl sulphoxide and 1 mL of 2 N HCl was added and filtered. The solutions were transferred to a separating funnel with the addition of 5 mL of Bromocresol green solution and 5 mL of phosphate buffer. The mixture was then shaken thoroughly with 1, 2, 3, and 4 mL of chloroform and collected in a 10-mL volumetric flask and diluted to volume with chloroform. A set of reference standard solutions of atropine (10, 20, 30, 40, 50 and $60 \mu g/mL$) were prepared in the same manner. The absorbances for the test and standard solutions were determined using an ultraviolet (UV) spectrophotometer at 470 nm. A blank sample was prepared for error correction. Finally, the total alkaloid content was calculated as mg of atropine equivalent (AE)/g of each extract.²³

Total phenolic content

The total phenolic compounds in all three fruit pulp extracts of CF were determined using Folin-Ciocalteu's method. A blue color is formed during the reaction, which is measured spectrophotometrically.²⁴ One milliliter of sample (1 mg/mL) was mixed with 1 mL of Folin-Ciocalteu's phenol reagent. After 5 min, 10 mL of a 7% Na₂CO₃ solution was added to the mixture, then 13 mL of deionized distilled water was added and mixed thoroughly. The mixture was kept in the dark for 90 min at 23°C for the reaction (blue color formation). Gallic acid was used as a standard and the standard solution was prepared as per the same method followed for the sample (10, 20, 30, 40, 50

and 60 µg/mL). Then, absorbance was taken at 765 nm. The total phenolic content was determined from extrapolation of a calibration curve, which was made by preparing a gallic acid solution, and was expressed as mg of gallic acid equivalent (GAEs) per g of extract (GA mg/g). The estimation of the phenolic compounds was carried out in triplicate. The following formula was used for the calculation:

$T = (C \times V)/M$

Where, T = total content of phenolic compounds, mg/g plant extract, in GAE; C = concentration of gallic acid established from the calibration curve (μ g/mL); V = volume of extract (mL); M = weight of water extract of the plant (g).

Total flavonoid content

The total flavonoid content was measured using an aluminum chloride colorimetric assay as described by Park et al. ²⁵ In a 10-mL test tube, 0.3 mL of extracts, 3.4 mL of 30% methanol, 0.15 mL of NaNO₂ (0.5 M), and 0.15 mL of AlCl₃.6H₂O (0.3 M) were mixed thoroughly. After a few min, 1 mL of NaOH (1 M) was added. In the same way, the standard solution was also prepared using rutin (Ru) as a standard. The standard curve for total flavonoids was made using a Ru standard solution (10, 20, 30, 40, 50 and 60 µg/mL). The solution was mixed well and the absorbance was measured against the reagent blank at 506 nm. The total flavonoids were expressed as mg of Ru equivalent/g of dried extract.

Antioxidant assays

Each sample was dissolved in 80% methanol to make a concentration of 1 mg/mL and then diluted to prepare the series concentrations for antioxidant assays.

1,1-diphenyl-2-picryl-hydrazyl (DPPH) assay

All the fruit pulp extracts were tested for their free radical scavenging activity against the stable free radical DPPH. The ability to bleach DPPH by the extracts was quantified using a spectrophotometer. The method used was as described by Brand-Williams et al.²⁶ One milliliter of 0.1 mM DPPH solution in methanol was mixed with 1 mL of plant extract solution of varying concentrations (25, 50, 100, 150, and 200 µg/mL). Corresponding blank samples were prepared using a mixed 1 mL methanol and 1 mL DPPH solution, methanol and L-ascorbic acid was used as reference standard (1-100 µg/mL) and the experiment was performed in triplicate. The decrease in absorbance was measured at 517 nm after 30 min in the dark (for the reaction) using a UV-Vis spectrophotometer. The percentage scavenging was calculated using the following formula:

DPPH scavenging effect (%)=[(A_{control}-A_{sample}/A_{control}) x 100]

The IC_{50} value of the sample i.e. the concentration of sample required to inhibit 50% of the DPPH free radical, was calculated using the calibration curve by linear regression.

Hydrogen peroxide-scavenging activity:

The hydrogen peroxide assay was as described by Nabavi et al.²⁷ Hydrogen peroxide solution (2 mmol/L) was prepared in phosphate buffer (pH 7.4). Extract (0.6 mL) at various

concentrations (25, 50, 100, 150 and 200 μ g/mL) was added to hydrogen peroxide solution. For each concentration, a separate blank sample was prepared. The absorbance of hydrogen peroxide with a UV visible spectrometer at 230 nm was determined after 10 min, and then readings were calculated for the blank solution containing phosphate buffer without hydrogen peroxide. The percentage inhibition of H₂O₂ scavenging activity was calculated using the formula below:

% Scavenging activity = [1- (Absorbance of test / Absorbance of control)] x 100

Anthelmintic property

Chemicals and drugs

All chemicals and drugs were obtained commercially and were of analytical grade. Albendazole was purchased from the local market of Bangalore. Dimethyl formamide (DMF) were purchased from Merck, Germany.

Selection of experimental organisms

The preliminary assay was performed on adult earthworm, Pheretima posthuma, belong to class Oligochaeta. Due to its easy availability and its anatomic and physiologic resemblance with intestinal round worm parasites of humans, they have been widely used for preliminary evaluations of anthelmintic activity. The experiment was conducted after ethics clearance was obtained from the Institutional Animal Ethics Committee of Krupanidhi College, Bangalore (Approval no: KCP/ PCOL/5/2017/). Thereafter, tapeworm (Taenia solium, Family: Taeniidae) was selected for assurance of anthelmintic activity. Earth worms were collected from moist soil of the medicinal garden of Krupanidhi College of Pharmacy, Bangalore, and tape worms were collected from a local slaughter house (infested intestines of pigs), Yeshwanthpur, Bangalore. Both were separately washed with normal saline to remove all foreign matter from the body and later used for the anthelmintic study. Earthworms of 3-5 cm in length and 0.1-0.2 cm in width. and tapeworms of 6-8 cm in length were used for the entire experimental protocol. Albendazole (25 mg/mL and 50 mg/mL) was used as a standard solution (prepared by dissolved in DMF) and each test solution of the CF fruit extracts (25, 50, 100 and 150 mg/mL) was evaluated for anthelmintic activity.

Methods

For the evaluation of each plant extract, four worms were placed in separate Petri dishes containing 20 mL solution of crude extracts in the said concentrations and then the worms were introduced to the solutions. The same method was used for each case.

Observations

Observations were made for the time taken for paralysis and death of individual worms during the completion of the investigation. When there was no movement of any part of the body, the time was noted for the paralysis condition, followed by the death time, which was noted when no movement of any part of the body even after being shaken vigorously, and also followed by fading of the body colors of the worms. Death was also ascertained when the worms were dipped in warm water at 50°C.²⁸ The experiment was carried out as per the guidelines of the Institutional Biosafety and Ethics Committee.²⁹

Statistical analysis

Data are expressed as mean \pm SD from three replications. For antioxidant assays and anthelmintic activity, the one-way ANOVA test followed by Tukey's test (p<0.05) was used to analyze the differences among IC₅₀ of the various extracts for different antioxidant assays. The IC₅₀ values were determined using the Graph Pad Prism 5 software. Correlation coefficient (r) was calculated for the extract and the activities. P values less than 0.05 were considered statistically significant.

RESULT AND DISCUSSION

Morphologic study of the fruits and seeds

Vernier calipers were used to measure the diameter (50 measurements) of CF fruits and seeds separately (Figures 1 and 2) and the diameters were recorded (Table 1 and 2, respectively). The color of the fruits (Figure 3) was greyish ash and there was no odor. The sizes ranged from 10-14 cm in diameter and the shape was globulus, tapering towards the embryonic site. Each fruit contained a single seed. The fruit pulp was made up of resinous matter (Figure 1). In contrast, the color of the seeds was off-white to gray; the seeds also had no odor. The seeds were 4-6 cm in diameter and sub globular-



Figure 1. Measurement of CF fruit



Figure 2. Measurement of CF seed

shaped, divaricate. The seeds were very hard to break (Figure 4). The high content of alkaloids i.e. berberine in the seed and fact that the embryonic opening part was closed by resinous matter were considered to cause the delayed germination. The average diameter of fruit was 12.35 cm; the minimum diameter of fruit was 10.6 cm; and the maximum diameter of fruit was 13.7 cm. The average seed diameter was 5.03 cm; the minimum



Figure 3. Fruits of CF



Figure 4. Seeds of CF a) Seeds without resinous mass, b) Broken seed with yellow berberine, c) Seeds surrounded by resinous mass

seed diameter was 4.1 cm; and the maximum seed diameter was 6.3 cm.

Extract yield:

The yield and color of the crude extract obtained from the extracted fruit pulp of CF are depicted in Table 3 and Figures 5 and 6. The yield of the extracts was found at a higher percentage (5.76%) in the combined extract of methanol and aqueous solvents, followed by the methanol (5.12%) and aqueous (4.12%) extracts. Earlier literature are also reported that combined extracts increased the yield of the crude extracts more than the individual extract in particular plant species.^{19,30}



Figure 5. Various CF fruit pulp extracts



Figure 6. Yield of crude extracts of CF fruits in various solvents

Table 1.	Table 1. CF Fruit diameter (50 measurements) determination								
Sl No.	Diameter (cm)	Sl No.	Diameter (cm)	Sl No.	Diameter (cm)	Sl No.	Diameter (cm)	Sl No.	Diameter (cm)
1	12.5	11	12.4	21	12.4	31	12.6	41	11.7
2	11.7	12	12.7	22	12.6	32	13.2	42	11.4
3	13.2	13	12.3	23	13.7	33	13.2	43	10.9
4	12.7	14	11.5	24	13.2	34	11.5	44	10.9
5	12.6	15	11.3	25	13.5	35	11.5	45	12.4
6	13.1	16	13.2	26	11.6	36	11.4	46	12.5
7	13.4	17	11.8	27	11.5	37	12.4	47	10.6
8	11.6	18	13.6	28	11.8	38	12.5	48	11.8
9	11.4	19	13.2	29	12.1	39	12.5	49	11.6
10	12.1	20	13.6	30	13.6	40	13.4	50	13.7

Each fruit diameter was determined using Vernier calipers

Table 2	Table 2. CF Seed diameter (50 measurements) determination								
Sl No.	Diameter (cm)	Sl No.	Diameter (cm)	Sl No.	Diameter (cm)	Sl No.	Diameter (cm)	Sl No.	Diameter (cm)
1	4.1	11	5.3	21	5.3	31	4.6	41	4.8
2	4.7	12	5.5	22	5.2	32	4.5	42	6.1
3	5.6	13	4.7	23	5.1	33	4.5	43	6.1
4	4.3	14	4.5	24	4.4	34	4.5	44	5.6
5	4.1	15	4.6	25	4.5	35	6.2	45	5.6
6	5.3	16	4.2	26	4.3	36	5.5	46	5.3
7	5.4	17	5.3	27	4.7	37	5.4	47	4.9
8	5.6	18	6.1	28	4.7	38	5.3	48	4.7
9	4.2	19	6.3	29	4.8	39	5.2	49	4.4
10	5.3	20	5.6	30	5.2	40	5.4	50	4.3

Each seed diameter was determined using Vernier calipers

Table 3. Yield and color of extracts of CF fruit pulps						
Different extracts	% Yield	Color of extract				
Chloroform	1.84	Pale greenish Ash				
Methanol (80%)	5.12	Medium to dark brown				
Aqueous	4.12	Dark brownish grey				
Aqueous + methanol (80%) (1:1)	5.76	Light to medium dark brown				

Table 4. Various chemical tests for CF fruit pulp extracts

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Tests	Different extracts						
	Chloroform	Methanol	Aqueous	Aqueous + methanol (1:1)			
Protein	+						
Carbohydrate							
Lipid	+						
Alkaloids		+	+	++			
Glycosides		+	+	++			
Tannins		+	+	++			
Flavonoids		+		++			
Saponin		+	+	+			
Resin	+	+		+			
Steroids							
Phenols		+	+	++			

(--) = Negative test; (+) = Positive test

The present investigation also resulted similarly. This indicated that the CF fruit pulp extracts was also dependent on the type of solvent used. A literature survey revealed that the extraction yield increases with the increasing polarity of the solvent used in extraction. Hence, the combined use of water and organic

Table 5. Content of total alkaloids in CF fruit pulp extracts					
Extracts	Alkaloid content (mg of AE/g)				
Methanol extract	68.20±0.025**				
Aqueous extract	41.01±0.015**				
Aqueous + methanol extract	77.02±0.020**				

Mean \pm SD (n=3); One-way ANOVA study followed by Tukey's post test. Significance level, **p<0.05. AE: Atropine equivalent

solvent may facilitate the extraction of soluble chemicals in water and/or organic solvent, and accordingly, the yield of secondary metabolites is higher than with individual solvent extraction.³¹ The results of this study are in agreement with the extraction yields of some medicinal plants.^{32,33}

Phytochemical screening

Various chemical tests were performed to detect the presence of secondary metabolites. The results are tabulated in Table 4. Based on the availability of the secondary metabolites, further estimation of the plant's vital constituents viz. total alkaloids, phenols, and total flavonoid were determined for methanol, aqueous, and combined aqueous and methanol extracts.

Total alkaloids

The alkaloid content was determined in CF fruit pulp extracts and expressed in terms of AE as mg of AE/g of extract (standard curve equation: y = 0.014x + 0.106, $R^2 = 0.994$). The highest concentration of alkaloid was measured as 77.02 mg/g in the combined extract. This is because of the high solubility of the alkaloids in the combined extract rather than in individual extracts (Table 5).^{31,33}

Total phenolic content

The total phenolic content in the entire extracted CF fruit pulp was determined using Folin-Ciocalteu's reagent and is expressed in terms of GAEs (mg of GA/g of extract, standard curve equation: y = 0.012x + 0.166, $R^2 = 0.991$). The highest

concentration of phenols was measured in the combined aqueous and methanol extract, followed by the methanol and aqueous extract. It was reported that high solubility of phenols in polar solvents provides high concentrations in extracts,^{34,35} and the same trend followed in the present investigation where combined aqueous and methanol solvents increased the solubility of phenolic compounds more than with individual solvents (Table 6).

Total flavonoid content

The concentration of flavonoids was determined using spectrophotometry for all three extracts and the content of flavonoids is expressed in terms of Ru equivalent (mg of Ru/g, the standard curve equation: y = 0.011x + 0.041, $R^2 = 0.993$). In this case, the same trend also followed as above. The highest flavonoid concentration (98.03 mg Ru/g) was recorded for the combined extract, followed by the methanol extract. This result was due to the solubility. It was reported that the concentration of flavonoid in plant extracts depends on the polarity of solvents used in the extract preparation.³⁶ A similar result was obtained in this study (Table 7).

Based on the estimation of total alkaloids, phenols, and flavonoids, a further investigation was carried out to reveal an antioxidant study. Phenolic and flavonoid compounds are known to have a correlation with antioxidant activities.^{37,38} The present study revealed the higher content of these compounds, and due to their presence, the CF fruit pulp may have scavenging activity i.e. mainly due to their redox properties, which play an important role in adsorbing and neutralizing free radicals and decomposing peroxides.³⁹ In view of this, two different methods were used to reveal antioxidant activity.

Antioxidant assays

An antioxidant is defined as any substance that inhibits oxidative damage to a target molecule.⁴⁰ Antioxidant compounds such as phenolic acids, polyphenols, and flavonoids reduce free radicals such as peroxide, hydrogen peroxide or lipid peroxyl,

Table 6. Content of total Phenolics in CF fruit pulp extracts						
Extracts	Phenolic content (mg of GA/g)					
Methanol extract	56.12±0.015**					
Aqueous extract	43.22±0.020**					
Aqueous + Methanol extract	69.13±0.188**					

Mean \pm SD (n=3); One-way ANOVA study followed by Tukey's post test. Significance level, **p<0.05. GA: Gallic acid

Table 7. Content of total Flavonoids in CF fruit pulp extracts						
Extracts	Flavonoids content (mg of RuE/g)					
Methanol extract	36.02±0.021**					
Aqueous extract	24.12±0.020**					
Aqueous + methanol extract	48.22±0.015**					

Mean \pm SD (n=3); One-way ANOVA study followed by Tukey's post test Significance level, **p<0.05. RuE: Rutin equivalent

and thus inhibit oxidative mechanisms that lead to degenerative diseases.⁴¹ Based on that, the following few methods were studied for the CF fruit pulp extract.

DPPH: The free radical scavenging activity of all crude extracts of the CF fruit pulp was quantitatively determined using a DPPH assay along with IC_{50} values. IC_{50} values represent the particular concentration of a test extract that inhibit activity by 50%. The results are tabulated in Figure 7. The effect of antioxidants on DPPH is thought to be due to their hydrogen-donating ability.42 DPPH is a purple-colored dye with absorption maxima of 517 nm, and upon reaction with a hydrogen donor, the purple color fades or disappears due to its conversion to 2, 2-diphenyl-1-picryl hydrazine, and hence absorbance is decreased. The combined aqueous and methanol extracts showed the maximum percentage inhibition (68.43%), followed by the methanol extract (58.24%) at 200 µg/mL concentration, whereas L-ascorbic acid showed 92.17% inhibition. The scavenging activity potency for all the extracts was determined using IC₅₀ values. The combined extract (aqueous + methanol) showed a lower IC₅₀ value of 42.38 μ g/mL, followed by the methanol extract (52.43 μ g/mL, Table 8) when compared with standard L-ascorbic acid (4.87 μ g/mL).

Hydrogen peroxide radical scavenging assay: Hydrogen peroxide is a weak oxidizing agent, and through oxidation of essential thiol (-SH) groups, it can inactivate a few enzymes. The generation of H_2O_2 in minimum quantities in biologic systems is significant to determine. Naturally-occurring iron complexes inside the cell react with H_2O_2 in vivo and highly reactive hydroxyl radicals are generated, which cause the origins of toxic effects.⁴³



Figure 7. DPPH activity of L-ascorbic acid and various CF fruit extracts Results were triplicate, each values represent mean \pm SD (n=3)

Table 8 IC values of different extracts of CE fruits in DPPH and

H_2O_2 scavenging assay								
	IC ₅₀ μg/mL							
Extracts	DPPH scavenging assay	Hydrogen peroxide- scavenging activity						
Methanol	52.43±0.024	54.22±0.014						
Aqueous	66.21±0.021	68.28±0.031						
Aqueous + methanol	42.38±0.012	46.80±0.011						
Ascorbic acid (standard)	4.87±0.030	9.18±0.020						

Results of triplicate tests, Each value represents mean ± SD (n=3)

The scavenging activity of CF fruit extracts was evaluated and compared with ascorbic acid and the results are tabulated in Figure 8. It was reported that H_2O_2 scavenging activity of extracts depends on the phenolic content, which can donate electrons to H_2O_2 ;⁴⁴ the present study revealed the content of phenolics showing scavenging activity of H_2O_2 . Among the three extracts, the combined extract (aqueous + methanol) followed by the methanol extract showed good activity in depleting H_2O_2 with IC₅₀ values of 46.80 and 54.22 µg/mL, respectively (Table 8). The percentage of H_2O_2 scavenging activity of the combined extract was found as 61.07%, followed by the methanol extract (59.23%) at 200 µg/mL concentration as compared with standard L-ascorbic acid (90.13%).

Correlation matrix: There is a direct correlation between the percentage yield and the content of secondary metabolites; the antioxidant study was observed in the present investigation.



Figure 8. Hydrogen peroxide scavenging activity of L-ascorbic acid and various CF fruit extracts

Table 9. Correlation matrix among the percentage yield, content of secondary metabolites, and antioxidant parameters							
Parameters	% Yield	Total phenolic content	Total alkaloid content	Total flavonoid content	% inhibition of DPPH	% inhibition of H_2O_2	
% Yield	1.00						
Total phenolic content	0.992**	1.00					
Total alkaloid content	0.987*	0.957	1.00				
Total flavonoid content	0.991	0.999**	0.957	1.00			
% inhibition of DPPH	0.988*	0.999**	0.951	0.999**	1.00		
% inhibition of H ₂ O ₂	0.992*	0.968	0.999**	0.967	0.961	1.00	

**Significant at 1%; *Significant at 5%

Extracts	Concentration (mg/ mL)	Earth worms		Tape worms	
		Time taken for paralysis (min)	Time taken for death (min)	Time taken for paralysis (min)	Time taken for death (min)
Control (0.9% normal saline)					
	25	34.03±0.04*	41.01±0.21*	36.22±0.11*	42.23±0.01*
	50	30.11±0.04*	38.19±0.14*	33.23±0.12*	40.02±0.34*
Methanol extract	100	26.06±0.04*	36.03±0.11*	28.13±0.11*	38.01±0.01*
	150	22.07±0.04*	28.04±0.02*	24.15±0.12*	29.02±0.02*
	25	39.21±0.01*	48.13±0.11*	40.31±0.20*	45.02±0.21*
	50	35.31±0.05*	43.06±0.04*	37.42±0.01*	42.10±0.24*
Aqueous extract	100	29.01±0.05*	37.07±0.11*	32.33±0.12*	39.11±0.21*
	150	26.20±0.11*	30.02±0.05*	28.31±0.11*	35.22±0.15*
	25	33.10±0.07*	42.20±0.11*	34.20±0.03*	45.10±0.21*
	50	29.11±0.04*	38.21±0.01*	30.13±0.01*	41.16±0.01*
Aqueous + methanol extract (1:1)	100	24.20±0.12*	32.12±0.04*	26.21±0.10*	33.11±0.10*
	150	19.12±0.24*	26.01±0.01*	21.14±0.22*	27.32±0.01*
	25	23.13±0.10	29.03±0.02	22.01±0.31	26.01±0.21
Albendazole (standard)	50	18.30±0.11	25.20±0.20	19.20±0.10	28.11±0.01

Values are expressed as mean ± SD. Values were find out by using one-way ANOVA followed by Dunnett's t-test. *Values are significantly different from control at (p<0.05)

The results are depicted in Table 9.

The results from Table 9 indicate that the percentage yield of extract in particular solvents (CF fruit pulp) has a direct correlation with the content of secondary metabolites and even antioxidant activities. Furthermore, Table 8 indicates that antioxidant activity is dependent on IC₅₀ values, which are inversely correlated; this result is in agreement with earlier research findings.⁴⁵ Based on the presence of various phytochemicals viz. alkaloids, phenols, tannins, flavonoids and due to the strong antioxidant activity of CF fruit extracts, further anthelmintic activity was tested for the first time. Many scientific studies have already revealed that the presence of phenols, tannins, and flavonoids leads to anthelmintic activity.^{46,47}

Anthelmintic activity: Preliminary anthelmintic activity was tested using various extracts of the CF plant on adult earthworms (Pheretima posthuma), followed by tape worms (Taenia solium), at doses of 25, 50, 100, and 150 mg/mL and compared with acontrol (0.9% normal saline) and albendazole (25 mg/mL and 50 mg/mL) as a standard. The results revealed that the combined aqueous and methanol extract of CF fruit pulp showed significant anthelmintic activity (26.01 min for the death of the worms) compared with the others with respect to paralysis followed by death for earth worms at 150 mg/mL concentration, and the same combined extract produced death in tape worms at 37.32 min, which was near to the standard drug (28.11 min) (Table 10). Table 10 reveals that albendazole at 50 mg/mL concentration showed 18.30 and 19.20 min for paralysis and 25.20 and 28.11 min for the death of the earth worms and tape worms, respectively.

A literature review reported that tannins and phenolics were known to interfere with energy generation in parasites with the mechanism of uncoupled oxidative phosphorylation,⁴⁶ and causes death by binding with free proteins in the gastrointestinal tract of the host animal or glycoprotein on the cuticle of the parasite. The estimation of alkaloids, phenolics, and flavonoids in CF fruit extracts resulted with a high content in the combined methanol and aqueous extract, which supported the strong anthelmintic activity.³¹ The result was reported similarly in earlier scientific research.^{48,49} In the present study, earthworms were selected for the preliminary study because they are more sensitive than tape worms and round worms.⁵⁰ Earlier literature established the anthelmintic activity of Coscinium fenestratum stem aqueous extract against round worms and earthworms⁵¹ and death resulted after more than 63 min, whereas in our study, the fruit pulp extract produced death at around 37.32 min, less than stem extract. The result must be due to the effect of the solvent used in extraction.

CONCLUSION

The present investigation has shown that CF fruit pulp has therapeutic activities due to the presence of secondary metabolites and has significant anthelmintic activity due to estimated total alkaloid, phenol, and flavonoid content. The solvent system played a vital role for the activities in which combined aqueous and methanol extract showed the most significant antioxidant and anthelmintic activity as compared with individual methanol and aqueous extracts. This is the first report on CF fruit pulp extract, which may be further explored for its phytochemical profile to recognize the active constituent responsible for anthelmintic activity.

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