

Evaluation of Antioxidant Activities and Phenolic Compounds of *Scorzonera latifolia* (Fisch. & Mey.) DC. Collected from Different Geographic Origins in Turkey

Türkiye'nin Farklı Coğrafik Bölgelerinden Toplanmış *Scorzonera latifolia* (Fisch. & Mey.) DC. Örneklerinin Fenolik Bileşiklerinin ve Antioksidan Aktivitelerinin Değerlendirilmesi

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ABSTRACT

Objectives: The chemical composition of plants is considered to be affected by many parameters. Therefore, the region where the samples are collected is likely to have an influence on the composition of phenolic compounds, so that their biological activities. In the present study, evaluation of antioxidant activity potentials of *Scorzonera latifolia* (Fisch. & Mey.) DC. aerial parts and roots, which were collected from different regions of Turkey, was aimed.

Materials and Methods: 1,1-diphenyl-2-picrylhydrazyl (DPPH) radical scavenging method and measurement of malondialdehyde (MDA) levels were used for determining antioxidant capacities of the tested extracts. In order to observe variations in the chemical composition of the investigated samples qualitatively as well as quantitatively, high performance liquid chromatography analyses were performed.

Results: Quantitative analysis showed that the amounts of chlorogenic acid and hyperoside in plants vary according to the regions where the samples were collected. As a result aerial parts of the *S. latifolia* collected from the Kars region have found to contain higher amount of chlorogenic acid (1246.78±3.20 μ g/g) as well as hyperoside (652.32±2.48 μ g/g) than other samples. The highest DPPH radical scavenging activity was determined with the IC₅₀ value of 1.036 mg/mL for *S. latifolia* aerial parts of Kayseri sample. MDA level was detected as the lowest with treatment of *S. latifolia* Bayburt root sample (4.41 nmol/mL).

Conclusion: According to the antioxidant activity results, no significant difference was observed in the antioxidant potential between the samples collected from different locations except for *S. latifolia* collected from the Kars region.

Key words: DPPH, free radical scavenging activity, high performance liquid chromatography, MDA, Scorzonera latifolia

ÖΖ

Amaç: Bitkilerin kimyasal içeriği pek çok parametreden etkilenmektedir. Bitkinin toplandığı bölge içerdiği fenolik bileşiklerin kompozisyonunu ve dolayısıyla aktivitelerini etkilemektedir. Bu çalışmada, Türkiye'nin farklı bölgelerinden toplanan *Scorzonera latifolia* (Fisch & Mey.) DC. kök ve toprak üstü kısımlarının antioksidan etki potansiyallerinin ölçülmesi amaçlanmıştır.

Gereç ve Yöntemler: Test edilen ekstrelerin antioksidan kapasiteleri 1,1-difenil-2-pikrilhidrazil (DPPH) radikal süpürücü etkileri ve malondialdehit (MDA) seviyesi üzerine etkilerinin ölçülmesiyle tespit edilmiştir. Araştırma konusu olan örneklerin kimyasal içeriğindeki değişiklikleri incelemek için kalitatif ve kantitatif yüksek performanslı sıvı kromatografisi analizleri yapılmıştır.

Bulgular: Kantitatif analiz sonuçları bitkilerde tespit edilen klorojenik asit ve hiperozit miktarlarının toplandığı bölgeye göre değiştiğini göstermiştir. *S. latifolia* Kars örneğinin diğer örneklere göre daha yüksek hiperozit (652.32±2.48 µg/g) ve klorojenik asit (1246.78±3.20 µg/g) içeriğine sahip olduğunu ortaya koymuştur. En yüksek DPPH radikal süpürücü etki 1.036 mg/mL IC₅₀ değeri ile *S. latifolia* Kayseri örneğinde gözlenmiştir. En düşük MDA seviyesi ise *S. latifolia* Bayburt kök örneğinde (4.41 nmol/mL) belirlenmiştir.

Sonuç: Antioksidan aktivite sonuçları değerlendirildiğinde, farklı lokasyonlardan toplanan örnekler arasında, *S. latifolia* Kars kök örneği haricinde belirgin bir farklılık tespit edilememiştir.

Anahtar kelimeler: DPPH, serbest radikal süpürücü aktivite, yüksek performanslı sıvı kromatografisi, MDA, Scorzonera latifolia

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INTRODUCTION

It has been reported that the risk of various chronic diseases such as cancer, diabetes, cardiovascular diseases, aging, chronic inflammation, cataracts, atherosclerosis, Alzheimer's disease could be reduced by consumption of foods and beverages rich in natural antioxidants.¹⁻⁴ Although free radicals are generated by normal physiological processes and they are involved in phagocytosis, energy production and regulation of cell growth,⁵ exogenous factors such as smoking, pollutants, ionising radiations, alcohol, synthetic pesticides, and solvent may increase the production of free radicals.⁶ The overproduction of free radicals provokes oxidative damage on nucleic acids, proteins, lipids or DNA and causes degenerative diseases.⁷ Natural antioxidants provide protection by their inhibitor and scavenger activities on free radicals.⁴ Therefore, the plants containing natural antioxidants are considered to have therapeutic potential in treatment of many diseases and these findings have led to increased interest in the antioxidant activity studies of plants.¹⁻³

Vitamin C, Vitamin E, carotenoids and phenolic compounds are recognized as the natural antioxidants found in plants.8 The phenolic compounds consisting of flavonoids, isoflavones, flavones, anthocyanins, coumarins, lignans, catechins and isocatechins, etc.⁹ have been shown to possess strong antioxidant activity.⁵ Phenolic compounds exert their antioxidant activities mainly due to their redox properties that led to act them as reducing agents, hydrogen donors, singlet oxygen quenchers or metal chelators.^{10,11} Different parameters such as time and period of sample collection, geographical origin and climatic conditions are considered as notable factors influencing the phenolic composition of plants.^{11,12} The plants, cultivated with different conditions, exhibit an alteration in the quantity of phytochemicals and therefore display varied therapeutic effects.^{13,14} The quality, uniformity and quantity of chemical constituents in plants are important parameters for the safety and effectiveness of herbal drugs.¹⁴ The genus Scorzonera L. which consists of about 180 species belongs to Asteraceae family and is distributed in arid regions of Asia, Europe and Africa. It is represented by 50 species in Turkey.¹⁵ The members of this genus are used as folk remedies against several diseases such as arteriosclerosis, kidney problems, hypertension and also for wound healing.^{16,17} Scorzonera latifolia Fisc. & Mey. (Asteraceae) naturally grows in Central and East Anatolian Region and it is a widely known species from which a mastic named "yakı sakızı" is obtained. Yakı sakızı is known to be prepared from the latex of the plant roots and used as a folk medicine for its analgesic activity and against infertility externally as well as antihelmintic internally.¹⁷ Antinociceptive, anti-inflammatory, wound healing, antioxidant and antimicrobial activities of S. latifolia have been reported.¹⁸⁻²¹ S. latifolia roots contain terpenes; taraxasteryl myristate, taraxasteryl acetate, motiol, 3-β- hydroxy fern-8 en-7-one acetate, urs-12-en-11-one-3-acetyl, 3-\u03c3-hydroxy-fern-7-en-6-one-acetate, olean-12-en-11-one-3-acetyl, fern-7-en-3-β-one, leucodin, β-sitosterol²²⁻²⁴, phenolic compounds; chlorogenic acid, chlorogenic acid methyl ester, 1,5-dicaffeoyl quinic acid, 3,5-dicaffeoyl quinic acid,

methylester of 3,5-dicaffeoyl quinic acid, hydrangenol-8-*O*-glucoside, hydrangenol-4'-*O*-glucoside, scorzotomentosin-4'-*O*-glucoside and a new isocoumarine derivative²⁵ as well as scorzoveratrin 4'-*O*- β -glucoside, scorzoveratrin, scorzoveratrozit, 4,5-dicaffeoylquinic acid, 4,5-dicaffeoylquinic acid methyl ester and caffeic acid.²⁶ Aerial parts of the plant have been reported to contain phenolics which were identified as quercetin-3-*O*- β -glucoside, hydrangenol-8-*O*-glucoside, swertisin, 7-methylisoorientin, 4,5-*O*-dicaffeoyl-quinic acid, 3,5-di-*O*-caffeoyl-quinic acid and chlorogenic acid.²⁷

In the current study, three samples of *S. latifolia* Fisc. & Mey. which were collected randomly from different locations of Turkey were evaluated for their antioxidant capacities and we also observed qualitative and quantitative variations in the chemical composition of samples by HPLC analysis.

EXPERIMENTAL

Plant materials

S. latifolia samples were collected from different locations of Turkey in flowering period. Identification of these plants was confirmed by Prof. Dr. H. Duman, in the Department of Biological Sciences, Faculty of Art and Sciences, Gazi University. Voucher specimens were stored in the herbarium of Ankara University, Faculty of Pharmacy (AEF) (Table 1).

Preparation of the extracts

Dried and powdered roots and aerial parts of the plants (0.25 g, each) were extracted in 10 mL of methanol:water (80:20) mixture by continuous stirring at room temperature for 8 hours. Obtained extracts were filtered and then concentrated to dryness under reduced pressure and low temperature (40-50°C) on a rotary evaporator to give crude extracts. The crude extracts were dissolved in methanol-water mixture and adjusted to 10 mL in volumetric flask.

Phytochemical screening

HPLC analysis

For the HPLC analyses of the extracts of the roots and aerial parts of *Scorzonera* species, the method which was developed and validated by Küpeli Akkol et al.¹⁹ was used. In accordance with this method, the analyses were carried out using Agilent LC 1100 model cromatograph (Agilent Technologies, California,

Table 1. Collection sites and herbarium numbers of the plants				
Plant species	Locality	AEF no		
<i>S. latifolia</i> (Kars)	Kars, Arpaçay, between Telekköy and Melekköy	23830		
<i>S. latifolia</i> (Kayseri)	Kayseri, Erciyes Mountain, 1800-2000 m	25941		
<i>S. latifolia</i> (Bayburt)	Bayburt, Kop Passage, 2000- 2409 m	23827		

USA) at the wavelength of 254 nm and peak areas were integrated by Agilent Software. A Supelcosil (250 mm x 4.6 mm; 5 μ m) column was used with a gradient elution of acetonitrile (A) and water: 0 min, A-B (8:92, v/v); 0-10 min, linear change from A-B (8:92, v/v) to A-B (18:82); 10-20 min, a linear change from A-B (18:82) to A-B (20:80); 20-30 min, a linear gradient elution from A-B (20:80) to A-B (30:70). Phenolic acids including *p*-coumaric acid, ferulic acid, rosmarinic acid, caffeic acid, chlorogenic acid and flavonoids such as apigenin, luteolin, quercetin, rutin, hyperoside, hesperidin were used as standards.

DPPH radical scavenging activity

DPPH radical scavenging activity assays of the samples were carried out according to the method of Williams et al.²⁸ 0.01 g of sample was dissolved in 10 mL dimethylsulfoxide and seven different concentrations (1 mg/mL to 0.015 mg/mL) were prepared with ½ dilutions. 2.9 mL DPPH solution dissolved in 10⁻⁴ M ethanol was added into 0.1 mL of sample solutions. The mixture was shaken vigorously and incubated 30 min in a water bath adjusted to 30°C. Absorbance of the resulting solution was measured against the blank (the same mixture without the sample) at 517 nm UV-visible spectrophotometer (Shimadzu). All assays were carried out in triplicates and propylgallate is used as positive control. Percentage of inhibition determined as follows:

% DPPH radical-scavenging = [(Absorbance of control - Absorbance of sample) / Absorbance of control] x 100

According to this method, decrease at the absorbance of the reaction mixture indicates stronger DPPH radical-scavenging activity. For each sample, the IC_{50} values were calculated by linear regression analysis using % inhibition and concentration values.

Thiobarbituric acid reactive substances assay

The measurement of malondialdehyde (MDA) levels was performed spectrophotometrically according to the modified method of Puhl et al.²⁹ and Zhang et al.³⁰ 2.0 mg of plant extracts solved in dH₂O were incubated with 8.125 mM CuSO₄ solution. Following incubation, trichloroacetic acid (0.1 g/ dL, final volume) and thiobarbituric acid (TBA) (0.67 g/dL, final volume) solutions were added and the absorbance was measured at 532 nm. Quantitation of TBA reactive substances (TBARS) was performed by comparison with a standard curve of MDA equivalents generated by acid-catalyzed hydrolysis of 1,1,3,3-tetraethoxypropane and the results were expressed as nmol/mL.

RESULTS AND DISCUSSION

Natural antioxidants from plant sources have been gaining more interest in recent years due to their important roles in the maintenance of health and protection from ageing-related degenerative diseases.^{31,32} Plants serve as potential sources of compounds with antioxidant activity such as vitamin C,

vitamin E, carotenoids, phenolic compounds, etc.³³ Phenolic compounds, one of the largest group of secondary metabolites, exhibit strong antioxidant activity as free radical scavengers, hydrogen donating sources or as singlet oxygen quenchers and metal ion chelators.^{10,31}

In the current study, antioxidant activities of *S. latifolia* samples were evaluated by using two different methods. The DPPH method is the most common, cost-effective and quick method to evaluate antioxidant activities of natural products.^{33,34} Scavenging activity against DPPH radical is considered as hydrogen donating ability of samples. DPPH is a stable free radical and accepts an electron on hydrogen radical to become a stable diamagnetic molecule.³³ The DPPH is decolorized by accepting an electron donated by an antioxidant.¹² DPPH radical scavenging activities of extracts are presented in Table 2. Among the root samples, *S. latifolia* Kars sample showed the highest DPPH radical scavenging activity with the IC₅₀ value of 1.906 mg/mL as the highest activity among the tested aerial parts was determined for *S. latifolia* Kayseri sample (IC₅₀ = 1.036 mg/mL).

Lipid oxidation plays an important role in the generation of reactive oxygen species and is considered as an important factor in the initiation and progression of several diseases.³⁵ TBARS assay is used to measure the degree of lipid peroxidation. TBA reacts with MDA which is a secondary product of lipid peroxidation, to give a red chromogen, than can be determined spectrophotometrically.³⁶ In the TBARS assay, the MDA levels for the root of *S. latifolia* samples from Bayburt, Kayseri, and Kars were determined as 4.41 nmol/mL, 4.66 nmol/mL, and 5.41 nmol/mL, respectively (Table 3).

The extracts of the aerial parts of all *S. latifolia* samples inhibited lipid peroxidation significantly and the lowest MDA level was detected in Kayseri sample as 5.15 nmol/mL (Table 3).

Table 2. IC ₅₀ values (mg/mL) o scavenging assay	of plant extracts in	DPPH radical
Plant material	Root	Aerial parts
<i>S. latifolia</i> (Kars)	1.906	1.122
<i>S. latifolia</i> (Kayseri)	4.102	1.036
<i>S. latifolia</i> (Bayburt)	4.628	1.143
	Control	
Propylgallate	0.742	

DPPH: 1,1-difenil-2-pikrilhidrazil

Table 3. MDA levels of plant extracts, the results are given as nmol/mL concentrations				
Plant material	Root	Aerial parts		
<i>S. latifolia</i> (Kars)	5.41	6.14		
<i>S. latifolia</i> (Kayseri)	4.66	5.15		
<i>S. latifolia</i> (Bayburt)	4.41	5.90		

MDA: Malondialdehit

The radical scavenging activities of the extracts were not found to be quite different except *S. latifolia* root Kars sample. Aerial part samples exhibited higher scavenger activities on DPPH radical compared to root samples (Table 2). The observed scavenger activities of aerial part samples were also close to the effect of propylgallate which is the positive control of the assay.

According to the HPLC results, root samples contain chlorogenic acid as one of the major constituent both in roots (Figure 1) and aerial parts (Figure 2) of *S. latifolia*. The highest amount was detected in the root of Kars sample as 1246.78±3.20 µg/g. None of the tested flavonoid standards were detected in root samples (Figure 3). However all aerial part samples contain chlorogenic acid as well as hyperoside as major compounds. Aerial part of *S. latifolia* was also found to contain higher amount of chlorogenic acid (652.32±2.48 µg/g) as well as hyperoside (305.71±1.70 µg/g) (Table 4).

Differences between the scavenging activity of root and aerial part extracts on DPPH radical may be explained by the flavonoid content of aerial part samples. It has been reported that, flavonoids which are widely distributed in plants have the ability to scavenge active oxygen radicals, superoxide and hydroxyl radicals by single electron transfer.³⁷ On the other hand in the TBARS assay, root samples were found to be more active when

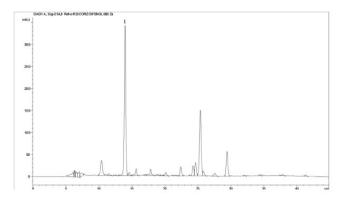


Figure 1. Chromatogram of the *S. latifolia* root (Kars sample) extract 1: Chlorogenic acid

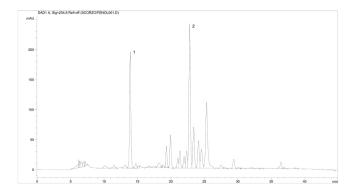


Figure 2. Chromatogram of the *S. latifolia* aerial part (Kars sample) extract 1: Chlorogenic acid, 2: Hyperoside

compared to aerial part samples. These results may be related to high amount of chlorogenic acid content of root extracts. Chlorogenic acid exhibits scavenging activity against reactive oxygen and nitrogen species³⁸ and inhibits the initiation of chain lipid peroxidations by free radicals.³⁹ Antioxidant activity of chlorogenic acid (5-CQA) against human LDL has also been reported.³⁶ Thus, chlorogenic acid may be considered to be one of the responsible compounds for the antioxidant activities of *S. latifolia* samples.

Qualitative and quantitative changes in the phytochemical composition of the S. latifolia root and aerial part samples which were collected from different localities of Anatolia were investigated by HPLC in the current study and the differences between the amounts of chlorogenic acid and hyperoside were observed. Results of present study have revealed that different samples of S. latifolia root and aerial parts have similar chemical composition however the quantity of the chlorogenic acid and hyperoside were different. A number of factors such as temperature, season, stages of maturity, geographical origin, climatic conditions and soil may affect the phytochemical content of plants.^{11,12,40,41} Changes in the phytochemical composition may result variation in therapeutic effects of the medicinal plants.^{13,14} According to the antioxidant activity results; we didn't find a significant correlation between the chlorogenic acid and/or hyperoside content and antioxidant capacities of samples.

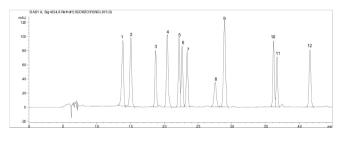


Figure 3. Chromatogram of the standart mixture

1: Chlorogenic acid, 2: Caffeic acid, 3: *p*-Coumaric acid, 4: Ferulic acid, 5: Rutin, 6: Hyperoside, 7: Luteolin-7-glucoside, 8: Hesperidin, 9: Rosmarinic acid, 10: Quercetin, 11: Luteolin, 12: Apigenin

Table 4. HPLC qualitative and quantitative data for roots and aerial parts of Scorzonera samples ($\mu g/g$)					
Plant material		Chlorogenic acid (µg/g ± standart error)	Hyperoside (µg/g ± standart error)		
<i>S. latifolia</i> (Kars)	R*	1246.78±3.20	-		
	AE*	652.32±2.48	305.71±1.70		
<i>S. latifolia</i> (Kayseri)	R*	988.070±3.78	-		
	AE*	290.64±0.84	70.99±0.29		
<i>S. latifolia</i> (Bayburt)	R*	541.96±2.56	-		
	AE*	295.27±0.53	36.24±0.42		

*R: Root, *AE: Aerial part

CONCLUSION

In conclusion, quantitative analyses have shown that same species collected from different localities may have different chemical composition. Therefore, the region chosen for the collection of a plant which is used as an herbal remedy is a very important parameter with respect to the effectiveness of an herbal drug.

Conflict of Interest: No conflict of interest was declared by the authors.

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