

# TURKISH JOURNAL OF PHARMACEUTICAL SCIENCES



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## CONTENTS

### Original articles

- Wound Healing Potential of Selected Liverworts Growing in Turkey  
Alev TOSUN, İpek SÜNTAR, Hikmet KELEŞ, Hatice ÖZENOĞLU KİREMİT, Yoshinori ASAKAWA, Esra KÜPELİ AKKOL... 285
- Development and Validation of HPLC-UV Method for the Determination of Diclofenac in Human Plasma with Application to a Pharmacokinetic Study  
Gülsüm Gül ARISOY, Emrah DURAL, Görkem MERGEN, Mustafa ARISOY, Gülin GÜVENDİK, Tülin SÖYLEMEZOĞLU ..... 292
- Composition of the Essential Oils of *Juniperus oxycedrus* L. subsp. *oxycedrus* Growing in Turkey  
Sevim ALAN, Mine KÜRKCÜOĞLU, Görkem ŞENER .....300
- Formulation and Evaluation of Lansoprazole Loaded Nanosponges  
Subhash Chandra Bose PENJURI, Nagaraju RAVOURU, Saritha DAMINENI, Sailakshmi BNS, Srikanth Reddy POREDDY ...304
- Development and Characterization of Voriconazole Loaded *In Situ* Gel Formulations for Ophthalmic Application  
Neslihan ÜSTÜNDAĞ OKUR, Ayşegül YOLTAŞ, Vildan YOZGATLI .....311
- Determination of Cytotoxic and Anticandidal Activities of Three *Verbascum* L. Species from Turkey: *V. cheiranthifolium* Boiss. var. *asperulum* (Boiss.) Murb. Monorg., *V. pycnostachyum* Boiss. & Heldr and *V. orgyale* Boiss. & Heldr  
Sevim KÜÇÜK, Filiz ÖZDEMİR, Gökalep İŞCAN, Zerrin İNCESU.....318
- The Importance of *Asphodeline* Species on Enzyme Inhibition: Anti-Elastase, Anti-Hyaluronidase and Anti-Collagenase Potential  
Mert İLHAN, Gökhan ZENGİN, Esra KÜPELİ AKKOL, Abdurrahman AKTÜMSEK, İpek SÜNTAR .....323
- Gastroprotective Effects of Various *Scrophularia striata* Extracts on Ethanol-Induced Gastric Ulcer in Rats  
Maryam RAFIEI, Fatemeh NAMAZI, Hamid RAJAIAN, Saeed NAZIFI .....328
- Voltammetric Determination of Nimesulide Using Multiwalled Carbon Nanotubes Modified Carbon Paste Electrode  
Fatma AĞIN, Vildan SERDAROĞLU .....335
- Cytotoxic Effect of *Conyza canadensis* (L.) Cronquist on Human Lung Cancer Cell Lines  
Fatma AYAZ, Mehmet SARIMAHMUT, Nurgün KÜÇÜKBOYACI, Engin ULUKAYA .....342
- ### Short communication
- Antioxidant and Anticholinesterase Activities of Essential Oil of *Alseodaphne peduncularis* Meisn  
Wan Mohd Nuzul Hakimi Wan SALLEH, Farediah AHMAD.....347
- ### Review article
- Nano Chemotherapeutics in Bone Metastasis and Targeting  
Müge KILIÇARSLAN, Miray GÖRGÖZ .....351

# CONTENTS 13(3) 2016

## **Original articles**

- Wound Healing Potential of Selected Liverworts Growing in Turkey  
Alev TOSUN, İpek SÜNTAR, Hikmet KELEŞ, Hatice ÖZENOĞLU KİREMİT, Yoshinori ASAKAWA, Esra KÜPELİ AKKOL.. 285
- Development and Validation of HPLC-UV Method for the Determination of Diclofenac in Human Plasma with Application to a Pharmacokinetic Study  
Gülsüm Gül ARISOY, Emrah DURAL, Görkem MERGEN, Mustafa ARISOY, Gülin GÜVENDİK, Tülin SÖYLEMEZOĞLU ..... 292
- Composition of the Essential Oils of *Juniperus oxycedrus* L. subsp. *oxycedrus* Growing in Turkey  
Sevim ALAN, Mine KÜRKÇÜOĞLU, Görkem ŞENER .....300
- Formulation and Evaluation of Lansoprazole Loaded Nanosponges  
Subhash Chandra Bose PENJURI, Nagaraju RAVOURU, Saritha DAMINENI, Sailakshmi BNS, Srikanth Reddy POREDDY ...304
- Development and Characterization of Voriconazole Loaded *In Situ* Gel Formulations for Ophthalmic Application  
Neslihan ÜSTÜNDAĞ OKUR, Ayşegül YOLTAŞ, Vildan YOZGATLI .....311
- Determination of Cytotoxic and Anticandidal Activities of Three *Verbascum* L. Species from Turkey: *V. cheiranthifolium* Boiss. var. *asperulum* (Boiss.) Murb. Monorg., *V. pycnostachyum* Boiss. & Heldr and *V. orgyale* Boiss. & Heldr  
Sevim KÜÇÜK, Filiz ÖZDEMİR, Gökalep İŞCAN, Zerrin İNCESU.....318
- The Importance of *Asphodeline* Species on Enzyme Inhibition: Anti-Elastase, Anti-Hyaluronidase and Anti-Collagenase Potential  
Mert İLHAN, Gökhan ZENGİN, Esra KÜPELİ AKKOL, Abdurrahman AKTÜMSEK, İpek SÜNTAR .....323
- Gastroprotective Effects of Various *Scrophularia striata* Extracts on Ethanol-Induced Gastric Ulcer in Rats  
Maryam RAFIEI, Fatemeh NAMAZI, Hamid RAJAIAN, Saeed NAZIFI .....328
- Voltammetric Determination of Nimesulide Using Multiwalled Carbon Nanotubes Modified Carbon Paste Electrode  
Fatma AĞIN, Vildan SERDAROĞLU .....335
- Cytotoxic Effect of *Conyza canadensis* (L.) Cronquist on Human Lung Cancer Cell Lines  
Fatma AYAZ, Mehmet SARIMAHMUT, Nurgün KÜÇÜKBOYACI, Engin ULUKAYA .....342
- ## **Short communication**
- Antioxidant and Anticholinesterase Activities of Essential Oil of *Alseodaphne peduncularis* Meisn  
Wan Mohd Nuzul Hakimi Wan SALLEH, Farediah AHMAD.....347
- ## **Review article**
- Nano Chemotherapeutics in Bone Metastasis and Targeting  
Müge KILIÇARSLAN, Miray GÖRGÖZ .....351

# Wound Healing Potential of Selected Liverworts Growing in Turkey

## Türkiye’de Yetişen Bazı Ciğer Otlarının Yara İyileştirici Etkileri

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### ABSTRACT

The Bryophytes have been traditionally used to cure cuts, burns, external wounds, bacteriosis, pulmonary tuberculosis, neurasthenia, fractures, convulsions, scalds, uropathy, inflammation, fever and pneumonia. Selected eight liverworts growing in Turkey were tested for their wound healing potential. In the present study, *in vivo* wound-healing activities of the ethereal extracts of some liverworts collected from Southern Anatolia were investigated by using linear incision and circular excision experimental patterns subsequently histopathological analysis. For the bioassays, 1% ointment formulations were prepared from the crude extracts. The ethereal extracts from *Reboulia hemisphaerica* (L.) Raddi (Aytoniaceae), *Plagiochasma rupestre* (J.R. Forst et G. Forst) Steph. (Aytoniaceae), and *Targionia hypophylla* L. (Targioniaceae) showed 43.1, 38.3, and 24.1% increase in tensile strength on the linear incision wound test, respectively. The extracts of *Reboulia hemisphaerica*, *Plagiochasma rupestre* and *Targionia hypophylla* also exhibited significant contraction effect with the values of 62.1, 58.0, and 39.8%, respectively, on the circular excision wound model, quite comparable to reference sample. The liverworts have many important biological activities, which are investigated previously. However, the wound healing effects of Turkish liverworts have not been expounded yet. Thus, the results obtained from the present study will be important for the future researches. Moreover, the present study supported the usages of some Bryophytes in some native societies of the world against skin problems, including wound, bruises, burns, etc.

**Key words:** Wound healing, Liverworts, Bryophytes, Marchantiophyta

### ÖZ

Bryofitler geleneksel olarak kesik ve yanık yaralarında, bakteriozis, pulmoner tüberküloz, nevrasteni, kırık, konvülsiyon, üropati, enflamasyon, ateş ve pnömoni tedavisinde kullanılmaktadır. Bu çalışmada, Türkiye’de yetişen ve Anadolu’nun güney kesimlerinden toplanan sekiz ciğer otunun eterli ekstraktlarının *in vivo* yara iyileştirici etkileri çizgisel insizyon ve dairesel eksizyon yara modelleri kullanılarak araştırıldı. Takibinde dokular üzerinde histopatolojik analizler yapıldı. Deneylerde, ham ekstreden hazırlanan %1’lik merhem formülasyonları kullanıldı. İnsizyon yara modelinde, *Reboulia hemisphaerica* (L.) Raddi (Aytoniaceae), *Plagiochasma rupestre* (J.R. Forst et G. Forst) Steph. (Aytoniaceae) ve *Targionia hypophylla* L. (Targioniaceae) eterli ekstraktları sırasıyla %43.1, %38.3 ve %24.1 oranlarında yara gerilme kuvvetinde artış oluşturdu. *Reboulia hemisphaerica*, *Plagiochasma rupestre* ve *Targionia hypophylla* ekstraktları sırasıyla %62.1, %58.0 ve %39.8 kontraksiyon değerleriyle dairesel eksizyon yara modelinde de referans örnek ile karşılaştırılabilir düzeyde ve anlamlı derecede etkili bulundu. Ciğer otlarının daha önceden belirlenmiş birçok biyolojik etkisi bulunmaktadır. Ancak, Türkiye’de yetişen ciğer otlarının yara iyileştirici etkileri daha önce araştırılmamıştır. Bu nedenle, bu çalışmada elde edilen sonuçlar gelecek araştırmalar için önem taşımaktadır. Bununla beraber, bu çalışma bryofitlerin halk arasında yara ve yanık gibi deri problemlerinde kullanımını doğrulamaktadır.

**Anahtar kelimeler:** Yara iyileşmesi, Ciğer otları, Bryofitler, Marchantiophyta

## INTRODUCTION

The medicinal plants and their derivatives have been used for providing relief and cure of illness in traditional forms in all cultures for many years (1-4). Many articles related the biological activity of the plants have been released contemporarily, some part of these studies had been reported for wound-healing activity (5).

The use of wound-healing plants to treat external injuries as well as internal injury is a popular ethnomedicinal practice in many countries such as China, Peru, India and Nigeria (5-9).

The phytochemistry of the Bryophytes has been ignored for a long time because of their morphological small size, unobtrusive position in the nature and difficult taxonomy. Besides of all, they do not have nutritional usages for humans due to their unpleasant taste. However, some Bryophytic plants are used by Native Americans in soup, and as a flavor agent in wine (10). The Bryophytes have also been traditionally used for their medicinal properties in China, Europe and North America, to cure cuts, burns, external wounds, bacteriosis, pulmonary tuberculosis, neurasthenia, fractures, convulsions, scalds, uropathy, inflammation, fever, pneumonia, etc. (10-13).

The Bryophytes are taxonomically placed between algae and pteridophytes. They are classified into three relevant phyla known as Bryophyta (Moss), Marchantiophyta (Liverwort) and Anthocerotophyta (Hornwort). The important group as of the liverworts belongs to Marchantiophyta group are represented by approximately 8000 species in 74 families, 380 genera, and considered to be the oldest aquatic-terrestrial primitive plants in the world. The liverworts are used for remedies of cuts, fractures, burns, bruises, wounds and snake bite as mentioned in the literature. Especially during the Renaissance, the liverworts were used in herbal medicines. In China, it is still used to treat the hepatitis and as an external cure to reduce the inflammation. In fact, the most important sides of the liverworts are their structurally and biologically interesting active terpenoid compounds, which are located in oil bodies known as unique organelles of the liverworts. The components isolated from liverworts have significant biological activities such as antimicrobial, antifungal, insect antifeedant, cytotoxic, apoptosis inducing activity, some enzyme inhibitory, neurotic sprouting activity, muscle relaxing, cardiotoxic activity and so on (11-13).

*Plagiochasma appendiculatum* Lehm. et Lind. (Aytoniaceae) distributed in India, is widely used ethnomedicinally by Gaddi tribe in Kangra valley for treating skin diseases in the form of paste, and applied externally for the treatment of burns, boils and blisters on the body. The species is also used for the treatment of skin eruption caused due to bright sun in the summer time (14). Himalayan Indians have used *Marchantia polymorpha* L. (Marchantiaceae) or *M. palmate* Nees to treat boils and abscesses because the young archegoniophore resembles an abscess as it emerges from the thallus. Moreover, in China, a mixture of the thallus of *Conocephalum*

*conicum* L. (Dum.) (Conocephalaceae) and *Marchantia polymorpha* with vegetable oils is used on bites, boils, burns, cuts, wounds and eczema (10).

The most investigated taxa are from Angiosperms whereas limited data is currently available about Bryophytes. Particularly, there is not enough information about Turkish liverworts in *Flora of Turkey* (15). Moreover, chemical and biological studies are scarcely limited on these species except for a few studies. Thus, the aim of this study was to evaluate the wound healing activity of the Turkish liverworts by using *in vivo* models for the preliminary screening. In this frame, the ethereal extracts (Table 1) of the *Riccia fluitans* L. (Ricciaceae), *Porella cordaeana* (Huebener) Moore (Porellaceae), *Porella platyphylla* (L.) Pfeiff. (Porellaceae), *Corsinia coriandrina* (Spreng.) Lindb. (Corsiaceae), *Mannia androgyna* (L.) A. Evans (Aytoniaceae), *Reboulia hemisphaerica* (L.) Raddi (Aytoniaceae), *Plagiochasma rupestre* (J.R. Forst et G. Forst) Steph. (Aytoniaceae), *Targionia hypophylla* L. (Targioniaceae) were debated for healing efficiency on incision and excision wounds models *in vivo*. In our ongoing research, their chemical constituents will be examined in future studies to find promising agents.

## MATERIALS AND METHODS

### Plant materials

Plant materials were collected from İzmir, Aydın and Muğla provinces, Turkey in 2009. Voucher specimens were identified by H. Ozenoglu Kiremit (Adnan Menderes University-Department of Biology) and deposited at the Herbarium of Adnan Menderes University in Aydın, Turkey. The list of the species with their collection places and herbarium numbers are presented in Table 1.

### Preparation of plant extracts

The plant extracts were prepared according to Tosun et al., 2013. First of all, dried plant materials were macerated with ether in a dark and cool place for 3-4 weeks. The extracts were filtrated, and the ether was evaporated *in vacuo* less than 30°C. The amount of the crude extracts was exactly weighed. The extracts were kept in a freezer up to use in the activity tests (16).

### Biological activity tests

#### Animals

Male, Sprague-Dawley rats (160-180 g) and Swiss albino mice (20-25 g) purchased from the animal breeding laboratories of from the animal breeding laboratory of Saki Yenilli (Ankara, Turkey), were used in the experiments. Before the experiments, animals were left for three days at room conditions for acclimatization. They were maintained on standard pellet diet and water *ad libitum* throughout the experiment. A minimum of six animals was used in each group.

### Preparation of test samples for bioassay

Test samples of *in vivo* experimental wound models were prepared in an ointment base (vehicle) consisting of glycol stearate, 1, 2-propylene glycol, liquid paraffin (3:6:1) in 1% concentration. Immediately, after the wound was created by a surgical blade, animals of the experimental group were topically treated with 0.5 g of each test ointment. Ointment base was applied to the animals of the vehicle group. Reference group animals were treated with 0.5 g of Madecassol® (Bayer, 00001199), whereas negative control group animals were not treated with any material.

### Wound healing activity

#### Linear incision wound model

All the animals were anaesthetized with 0.05 cc Xylazine (2% Alfazine®) and 0.15 cc Ketamine (10% Ketazol®), and the back hairs were shaved with a shaving machine. Two linear-paravertebral full-thickness skin incisions of 5 cm long were made with a sterile surgical blade at the distance of 1.5 cm from the midline of each side of the vertebral column (17). The incised wounds were sutured by three interrupted sutures of 1 cm apart. The extracts, the reference drug (Madecassol®) and the vehicle were topically applied once in a day throughout nine days. Animals of the negative control group were not treated with any material. On the 9<sup>th</sup> post wound day, all the sutures were removed. Animals were killed under anesthesia on day 10<sup>th</sup>. For the determination of tensile strength of the wound, a linear-paravertebral incised skin sample was measured using a tensiometer (Zwick/Roell Z0.5, Germany). Another skin sample underwent histopathological examination (18-20). A tensiometer measures the breaking strength in N (Newton), which is called tensile strength.

#### Circular excision wound model

The circular excision wound model was used for the evaluation of the wound contraction. The mice were anaesthetized with 0.02 cc Xylazine (2% Alfazine®) and 0.08 cc Ketamine (10% Ketazol®) and the back hairs were depilated by shaving. A circular wound was created on the dorsal interscapular region of each animal by excising the skin with a 5 mm biopsy punch; wounds were left open. The extracts, the reference

drug (Madecassol® Bayer) and the vehicle ointments were applied topically once a day until the wounds of one group were completely healed. The progressive changes in the wound area were monitored by a camera (Fuji, S20 Pro, Japan) every other day. Later, wound area was measured by using AutoCAD program. Wound contraction was calculated as the percentage of the reduction in the wounded area. For the histopathological examination, a specimen sample of tissue was removed from the healed skin of each group of mice (20-22).

### Histopathology

The cross-sectional full-thickness skin specimens from each group were collected at the end of the experiment to evaluate for the histopathological alterations. Samples were fixed in 10% buffered formalin, processed and blocked with paraffin and then sectioned into five micrometer sections and stained with hematoxylin & eosin (HE) and Van Gieson (VG) stains. The tissues were examined by the light microscope (Olympus CX41 attached Kameram® Digital Image Analyze System) and graded subjectively as mild (+), moderate (++) and severe (+++) for epidermal or dermal re-modeling. Re-epithelization or ulcer in epidermis; fibroblast proliferation, mononuclear and/or polymorph nuclear cells, neovascularization and collagen depositions in the dermis were analyzed to score the epidermal or dermal re-modeling. At the end of the examination, obtained results were combined and staged for wound healing phases as inflammation, proliferation, and re-modeling in all groups (20).

### Statistical analysis of the data

The data on percentage wound healing was statistically analyzed using one-way analysis of variance (ANOVA). A *p*-value  $\leq 0.05$  was considered statistically significant. Histopathologic data were considered to be nonparametric (20).

## RESULTS AND DISCUSSION

In the present study, the diethyl ether extracts of eight different species of the liverworts growing in Turkey (Table 1) were investigated for their *in vivo* wound healing activity. The wound healing activity was evaluated by using linear incision

**Table 1. The herbarium numbers and collection sites of the liverworts growing in Turkey**

| Family        | Plant name  | Herbarium number | Collection sites  |
|---------------|---|------------------|---|
|               | <i>Mannia androgyna</i> (L.) A. Evans                       | C11/237          | Muğla; Milas, Kapıkırı Village, Heraclea Archaic City Ruins, on the rocks and soil bank near the roadside, 35 m |
| Aytoniaceae   | <i>Plagiochasma rupestre</i> (J.R. Forst et G. Forst) Steph | C11/226          | Muğla; Milas, Kapıkırı Village, Heraclea Archaic City Ruins, on the rocks and soil ground, 70 m                 |
|               | <i>Reboulia hemisphaerica</i> (L.) Raddi                    | C11/227          | Aydın; Koçarlı, Mersin Belen road 5 <sup>th</sup> km, on the road in stream bank, 696 m                         |
| Porellaceae   | <i>Porella cordaeana</i> (Huebener) Moore                   | C11/223          | Aydın; Koçarlı, Mersin Belen road 5 <sup>th</sup> km, on the road in stream bank, 696 m                         |
|               | <i>Porella platyphylla</i> (L.) Pfeiff                      | C11/222          | Aydın; Koçarlı, Mersin Belen road 5 <sup>th</sup> km, on the road in stream bank, 696 m                         |
| Ricciaceae    | <i>Riccia fluitans</i> L.                                   | C11/225          | İzmir; Selçuk, Zeytinköy Village, Kazangöl Lake, 3 m  |
| Targioniaceae | <i>Targionia hypophylla</i> L.                              | C11/236          | Muğla; Milas, Kapıkırı Village, Heraclea Archaic City Ruins, on the rocks and soil bank near the roadside, 30 m |

and circular excision wound models. The skin samples were also evaluated histopathologically.

As shown in Table 2, topical application of the ointment prepared with the ethereal extracts of the *Reboulia hemisphaerica*, *Plagiochasma rupestre* and *Targionia*

**Table 2. Effects of the Liverworts extracts on linear incision wound model**

| Material                      | Statistical Mean ± S.E.M. | (Tensile strength %) |
|-------------------------------|---------------------------|----------------------|
| Vehicle                       | 9.60±2.13                 | 5.7                  |
| Negative control              | 10.18±2.44                | -                    |
| <i>Corsinia coriandrina</i>   | 10.88±2.02                | 13.3                 |
| <i>Mannia androgyna</i>       | 9.15±2.82                 | -                    |
| <i>Porella cordaeana</i>      | 10.09±1.68                | 5.1                  |
| <i>Porella platyphylla</i>    | 11.39±1.79                | 18.7                 |
| <i>Plagiochasma rupestre</i>  | 13.28±1.64                | <b>38.3**</b>        |
| <i>Riccia fluitans</i>        | 9.47±2.17                 | -                    |
| <i>Reboulia hemisphaerica</i> | 13.74±1.10                | <b>43.1***</b>       |
| <i>Targionia hypophylla</i>   | 11.93±1.29                | <b>24.3*</b>         |
| Madecassol®                   | 15.24±1.06                | <b>58.8***</b>       |

\*, p<0.05, \*\*, p<0.01, \*\*\*, p<0.001, S.E.M.: Standard error of the mean  
Percentage of tensile strength values: Vehicle group was compared to Negative control group; Extracts were compared to Vehicle group

*hypophylla* onto the incised wounds demonstrated the best wound tensile strength by the highest values of 43.1, 38.3, and 24.3%, respectively.

The contraction values of the wounds on the circular excision wound model for vehicle, negative control, the ethereal extracts and reference drug treated group are shown in Table 3. The ethereal extracts of *Reboulia hemisphaerica*, *Plagiochasma rupestre* and *Targionia hypophylla* were found to have wound healing potential with the contraction values of 35.65 and 62.13%; 27.94 and 58.04%; 31.48 and 39.78%; respectively, on day 10 and 12, while the vehicle and negative control groups and the extracts of the liverworts showed no statistically significant wound healing activity onto the incised wounds demonstrated the best wound tensile strength by the highest values of 43.1, 38.3, and 24.3%, respectively.

Histopathological analysis showed that the ethereal extracts of *Reboulia hemisphaerica*, *Plagiochasma rupestre* and *Targionia hypophylla* treated groups demonstrated better healing with rapid re-epithelization. The data obtained in histopathological evaluation were in accord with the outcome of wound models used in the present study. Delay in wound healing processes were observed especially in the vehicle and negative control groups compared to others (Table 4, Figure 1).

Skin sections show the hematoxylin & eosin (HE) stained epidermis and dermis in A, and the dermis stained with

**Table 3. Effects of the Liverworts on circular excision wound model**

| Material                      | Wound area ± S.E.M. (Contraction %) |                       |                       |                              |                                |                                |                                 |
|-------------------------------|-------------------------------------|-----------------------|-----------------------|------------------------------|--------------------------------|--------------------------------|---------------------------------|
|                               | 0                                   | 2                     | 4                     | 6                            | 8                              | 10                             | 12                              |
| Vehicle                       | 20.17±2.21                          | 17.22±1.86<br>(1.37)  | 16.65±2.01<br>-       | 15.03±2.15<br>-              | 11.08±1.53<br>(11.29)          | 7.91±1.42<br>(10.01)           | 3.67±0.32<br>(8.48)             |
| Negative control              | 19.67±2.18                          | 17.46±1.99            | 16.59±1.95            | 14.88±1.83                   | 12.49±1.85                     | 8.79±1.42                      | 4.01±1.19                       |
| <i>Corsinia coriandrina</i>   | 19.44±1.80                          | 16.03±1.86<br>(6.91)  | 14.90±2.10<br>(10.51) | 13.13±2.14<br>(12.64)        | 9.49±1.31<br>(14.35)           | 6.22±1.37<br>(21.37)           | 2.93±0.25<br>(20.16)            |
| <i>Mannia androgyna</i>       | 19.40±2.77                          | 17.03±1.73<br>(1.10)  | 17.01±1.86<br>-       | 14.19±1.92<br>(5.59)         | 10.20±1.34<br>(7.94)           | 7.77±1.39<br>(1.77)            | 3.51±0.88<br>(4.36)             |
| <i>Porella cordaeana</i>      | 19.59±1.96                          | 17.39±1.25<br>-       | 16.72±2.16<br>-       | 14.61±1.89<br>(2.79)         | 10.51±1.39<br>(5.14)           | 7.41±1.26<br>(6.32)            | 3.71±0.59<br>-                  |
| <i>Porella platyphylla</i>    | 19.27±1.53                          | 15.64±1.07<br>(9.18)  | 14.05±1.48<br>(15.62) | 13.26±2.36<br>(11.78)        | 9.70±1.47<br>(12.45)           | 6.60±1.11<br>(16.56)           | 2.96±0.34<br>(19.35)            |
| <i>Plagiochasma rupestre</i>  | 20.68±2.41                          | 15.02±1.40<br>(12.78) | 13.44±1.96<br>(19.28) | 11.89±2.18<br>(20.89)        | 9.79±1.15<br>(11.64)           | 5.70±1.10<br>(27.94)           | 1.54±0.09<br><b>(58.04)**</b>   |
| <i>Riccia fluitans</i>        | 19.90±2.55                          | 16.81±1.56<br>(2.38)  | 16.08±2.03<br>(3.42)  | 14.80±1.90<br>(1.53)         | 10.76±1.60<br>(2.89)           | 8.05±1.56<br>-                 | 3.84±0.29<br>-                  |
| <i>Reboulia hemisphaerica</i> | 20.56±2.06                          | 15.58±1.44<br>(9.52)  | 14.07±1.65<br>(15.49) | 12.21±2.23<br>(18.76)        | 8.84±1.17<br>(20.22)           | 5.09±1.16<br><b>(35.65)*</b>   | 1.39±0.08<br><b>(62.13)**</b>   |
| <i>Targionia hypophylla</i>   | 19.78±2.23                          | 15.41±1.85<br>(10.51) | 14.62±1.29<br>(12.19) | 12.19±1.76<br>(18.89)        | 8.36±1.69<br>(24.55)           | 5.42±1.31<br><b>(31.48)*</b>   | 2.21±0.14<br><b>(39.78)*</b>    |
| Madecassol®                   | 19.31±1.65                          | 13.89±1.14<br>(19.34) | 13.24±1.79<br>(20.48) | 9.51±1.06<br><b>(36.73)*</b> | 3.19±0.75<br><b>(71.21)***</b> | 1.17±0.43<br><b>(85.21)***</b> | 0.00±0.00<br><b>(100.00)***</b> |

\*, p<0.05, \*\*, p<0.01, \*\*\*, p<0.001, S.E.M.: Standard error of the mean  
Percentage of contraction values: Vehicle group was compared to Negative control group, Extracts were compared to Vehicle group.

Van Gieson (VG) in B. The original magnification was x100 and the scale bars represent 120  $\mu\text{m}$  for figures in A, and the original magnification was x400 and the scale bars represent 40  $\mu\text{m}$  for B. Data are representative of 6 animal per group. 1. Vehicle; 2. Negative Control; 3. *Corsinia coriandrina*; 4. *Mannia androgyna*; 5. *Porella cordaeana*; 6. *Porella platyphylla*; 7. *Plagiochasma rupestre*, 8. *Riccia fluitans*, 9. *Reboulia hemisphaerica*, 10. *Targionia hypophylla*, 11. Madecassol®. Arrows pointing events during wound healing; s: scab, u: ulcer, re: re-epithelization, f: fibroblast, c: collagen, mnc: mononuclear cells, pmn: polymorphonuclear cells, nv: neovascularization.

A large number of people are suffered from wounds especially from chronic wounds. In spite of advances in medicine, the availability of drugs which is curative in the wound process is still limited (7). Nowadays, the natural products are potential sources for the development of the new drugs as well in the past. Wounds are physical injuries that result in an opening or breaking of the skin. Wound infections are the most common end, especially in developing countries because of poor hygienic conditions and also seriously reduce their quality of life. Thus, developing new wound healing agents are important in biological activity studies. The plant natural products can act as potent anti-inflammatory, antioxidant or anticancer agents. Antioxidant, antimicrobial and anti-inflammatory activities help to promote wound healing and contribute skin regeneration. The mechanism of antioxidant compounds is the inhibition of lipid peroxidation and by way of radical scavenging activity, which prevent cell damage.

Antimicrobial activity is also important for the wound healing period, since the wound is more inclined to microbial attacks such as *Staphylococcus aureus*, *Streptococcus pneumonia*, *Klebsiella pneumonia*, which are caused infections and latency in wound area (5,23).

Wound-healing is a complex multifactorial process that results in the contraction and closure of the wound and restoration of a functional barrier (24,25). Thus, wound-healing process consists of several phases, including homeostasis, inflammation, cellular proliferation and migration of different cell types as resulting remodeling phase. The first response is inflammation as a defense mechanism of the tissue, which provides a resistance to the microbial contaminations. However, a long duration in the inflammatory phase causes a delay in healing process. Skin fibroblast proliferation is important in tissue repair as fibroblasts are involved in migration, proliferation, contractions and collagen production. The process of re-epithelialization occurs in the proliferatives phase. The final phase of the wound healing is called as maturation and remodeling phase. The completion of healing is ended by collagen coverage (3,5,6,23,26).

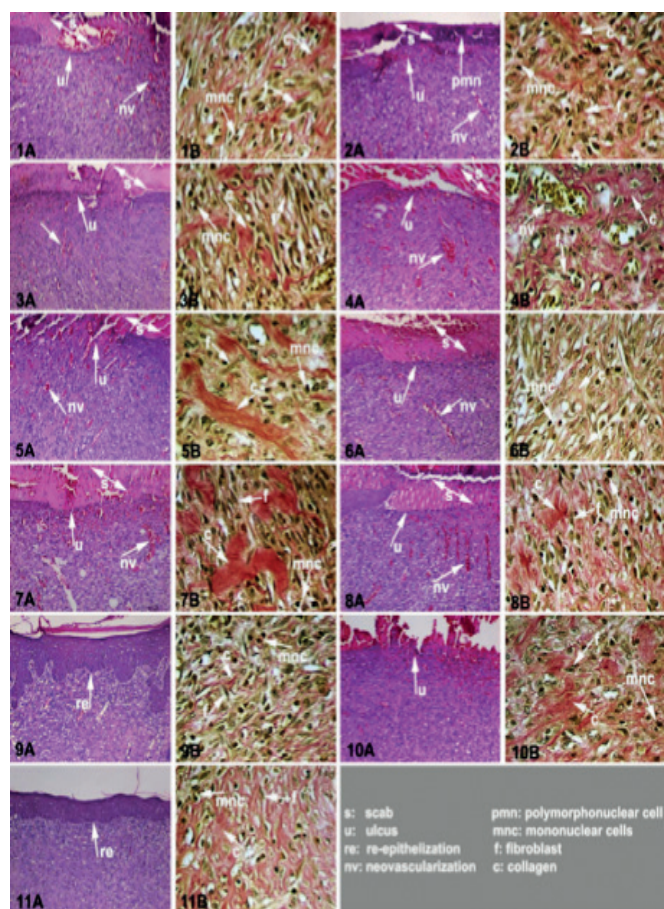
The liverworts have the wide range of endemic species, which include many interesting constituents with important biological activities. Principally, the liverworts contain a large amount of mono-, sesqui- and diterpenoids, and aromatic compounds with unique structures. However, the reason of the increasing interest of the liverworts is for their structurally impressive constituents exhibiting high therapeutic effects as cytotoxic, antitumor, antimicrobial, antifungal, antifeedant,

**Table 4. Wound healing processes and healing phases of the experimental group animals**

| Groups                        | Wound Healing Processes |       |      |       |       |      |      | Healing Phases |      |       |      |
|-------------------------------|-------------------------|-------|------|-------|-------|------|------|----------------|------|-------|------|
|                               | S                       | U     | RE   | FP    | CD    | MNC  | PMN  | NV             | I    | P     | R    |
| Vehicle                       | +++                     | +++   | -    | ++    | ++    | +/** | +/** | ++             | ++   | ++    | -    |
| Negative control              | ++                      | +++   | -    | +/+++ | +/+   | ++   | ++   | ++             | ++   | +/+++ | -    |
| <i>Corsinia coriandrina</i>   | ++                      | +/+++ | -    | ++    | ++    | +/** | +/** | +/**           | +/** | ++    | -    |
| <i>Mannia androgyna</i>       | ++                      | +++   | -    | ++    | +/+++ | +/** | +/** | +/+++          | +/** | +/+++ | -    |
| <i>Porella cordaeana</i>      | +/+++                   | +++   | -    | +/+++ | ++    | ++   | +/** | +/+++          | +/** | +/+++ | -    |
| <i>Porella platyphylla</i>    | +/+++                   | +/+++ | -    | +/+++ | +/**  | +/** | +/** | ++             | +/** | +/+++ | -    |
| <i>Plagiochasma rupestre</i>  | +/+++                   | +/+++ | -    | +/+++ | ++    | ++   | +/** | ++             | +/** | +/+++ | -    |
| <i>Riccia fluitans</i>        | +/+++                   | +++   | -    | +/+++ | ++    | ++   | +/** | ++             | +/** | +/+++ | -    |
| <i>Reboulia hemisphaerica</i> | ++                      | -     | +/** | +/+++ | +/**  | +/** | +/** | ++             | +/** | ++    | +    |
| <i>Targionia hypophylla</i>   | ++                      | +/+++ | -    | ++    | ++    | +/** | +/** | ++             | +/** | ++    | -    |
| Madecassol®                   | +/**                    | -     | +/** | ++    | ++    | +    | +    | +/**           | +/** | ++    | +/** |

\*HE and VG stained sections were scored as mild (+), moderate (++) and severe (+++) for epidermal and/or dermal re-modeling. S: Scab, U: Ulcer, RE: Re-epithelization, FP: Fibroblast proliferation, CD: Collagen depositions, MNC: Mononuclear cells, PMN: Polymorphonuclear cells, NV: Neovascularization, I: Inflammation phase, P: Proliferation phase, R: Re-modeling phase





**Figure 1.** Histopathological view of wound healing and epidermal/dermal re-modeling in the vehicle, negative control, extracts and reference ointment Madecassol® administered animals

insecticidal, muscle relaxing, some enzyme inhibitory and apoptosis inducing activities (11-13,27).

There is not much investigation as *in vivo* biological activity studies on the most active species of *Plagiochasma rupestre* and *Reboulia haemispherica* and *Targionia hypophylla* as well as ethnomedicinal usages of these species (11). However, another species of the *Plagiochasma*, *Plagiochasma appendiculatum* (Aytoniaceae) used by Gaddi tribe in India ethnomedicinally for treating skin diseases has been found potent wound healing activity as evident from the wound contraction and increased tensile strength (14,16). It was also found that *P. appendiculatum* extract possesses potent antioxidant activity by inhibiting lipid peroxidation and increase in the super oxide dismutase (SOD) and catalase (14). Moreover, according to the records, the liverworts are used for the cure of cuts, fractures, burns, bruises, open wounds and snakebites. Especially in China, the mixture of *Marchantia polymorpha* and *Conocephalum conicum* with vegetable oils is used for bites, burns, cuts, eczema, and wounds (10).

According to the phytochemical studies on the liverworts, a wide variety of the terpenoid derivatives such as mono, sesqui- and diterpenes which are mainly accumulated

in nonpolar fractions as well as aromatic and phenolic compounds have been isolated from too many liverworts collected different places of the world (11). In thin layer chromatography and qualitative analysis by using chemical reagent, these type components are detected in the ethereal extracts of the Turkish liverworts. In our previous study GC analysis was conducted on the ethereal extracts of *Reboulia hemisphaerica* and  $\beta$ -microbiotene (12.43%), grimaldone (8.62%),  $\beta$ -caryophyllene (4.91%) and cis- $\beta$ -elemene (4.14%) were determined as the major components (16). In previous studies it was demonstrated that terpenic compounds possess antimicrobial, anticarcinogenic, antioxidant, anti-inflammatory, analgesic and wound healing activities (28-30). Eventually, the effect might be attributed to the mainly terpenic compounds. Moreover, it is obvious that phenolics have potent antioxidant and anti-inflammatory effects, which are the main principles in the wound healing effect (11).

## CONCLUSION

The ethereal extracts of the liverwort species growing in southern part of Turkey such as *Riccia fluitans*, *Porella cordaeana*, *Porella platyphylla*, *Corsinia coriandrina*, *Mannia androgyna*, *Reboulia hemisphaerica*, *Plagiochasma rupestre*, *Targionia hypophylla* were evaluated comparatively for the wound-healing activity. Experimental results indicated that the ethereal extracts of *Reboulia hemisphaerica*, *Plagiochasma rupestre* and *Targionia hypophylla* has been potent wound healing capacity in this test system. The rest of the species did not show any remarkable wound healing effect. In the present study, wound-healing activity of the liverwort extracts supported the traditional use of these species. However, further studies are need to isolate the pharmacologically active compounds contributing to the wound-healing properties of the liverwort species. Moreover, wild species of the plants that have no information about their usages are also waiting to be investigated for new agents to heal the illnesses.

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# Development and Validation of HPLC-UV Method for the Determination of Diclofenac in Human Plasma with Application to a Pharmacokinetic Study

## Diklofenak'ın İnsan Plazmasından HPLC-UV ile Tayin Yöntemi Geliştirilmesi ve Validasyonu ile Farmakokinetik Çalışmaya Uygulanması

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### ABSTRACT

A simple, rapid and reliable high performance liquid chromatography method (HPLC) with ultraviolet detection (UV) was developed and validated according to ICH guidelines, for quantitative analysis and therapeutic drug monitoring of diclofenac sodium (DS) in human plasma. Plasma samples (0.7 mL) were acid hydrolysis by 100 µL, 1 M hydrochloric acid. Analytes were concentrated from plasma by liquid-liquid extraction with 2 mL ethyl acetate by repeated twice, which allows to obtain good extraction yields (98.75%-99.32%). The separation was achieved by employing C18 analytical column (3.5 µm particle size, 150 mmx3.9 mm I.D.) under isocratic conditions using acetonitrile and NaH<sub>2</sub>PO<sub>4</sub> mixture (42.5:57.5, v/v) as mobile phase (pH: 3.16) flow rate of 1.5 mL/min. Naproxen (3 µg/mL) was used as an internal standard (IS). The DS and IS were detected at 281 nm and eluted at 2.6 and 6.2 min, respectively. Total run time was 7 min. Method showed linearity with very good determination coefficients ( $r^2=0.999$ ), over the concentration range of 50 - 1600 ng/mL. Limits of detection (LOD) and quantification (LOQ) were 8.95 ng/mL and 27.12 ng/mL, respectively. Intra-day precision and accuracy were between 0.93-5.27; 1.74-9.81, respectively. Inter-day precision and accuracy were between 2.71-6.64; 2.03-9.16, respectively. This method was successfully applied for determination of DS plasma concentrations during a pharmacokinetic study in healthy volunteers (n=12) after an oral administration of Voltaren® 75 mg/tablet and remarkable variations in DS levels were observed. In our study, on the contrary to equivalent doses of DS, the observed significant differences in plasma levels of DS, on 2<sup>nd</sup>, 4<sup>th</sup> and 6<sup>th</sup> hours, can be explained by pharmacokinetic differences, that arise from mainly polymorphisms of CYP2C9 and CYP3A4, which are major enzymes responsible for DS metabolism.

**Key words:** Diclofenac sodium, Validation, Plasma, High performance liquid chromatography, Ultraviolet detection

### ÖZ

Diklofenak sodyumun (DS) insan plazmasında kantitatif analizleri ve terapötik ilaç izlemi için, basit, hızlı ve güvenilir bir ultraviyole dedektörlü yüksek performanslı sıvı kromatografisi (YPSK-UV) metodu geliştirildi ve ICH kurallarına göre valide edildi. Plazma örnekleri (0.7 mL) 100 µL, 1 M hidroklorik asitle hidroliz edildi. Analitler plazmadan iyi ekstraksiyon verimi (%98.75-%99.32) sağlayan sıvı-sıvı ekstraksiyonu metodu ile 2 mL etil asetatla iki tekrar ile elde edildi. Ayrım izokratik şartlar altında C18 analitik kolon (3.5 µm partikül büyüklüğü, 150 mmx3.9 mm I.D.), mobil faz asetonitril ve NaH<sub>2</sub>PO<sub>4</sub> (42.5:57.5, v/v) karışımı (pH 3.16), akış hızı 1.5 mL/dk ile gerçekleştirildi. Naproksen (3 µg/mL) iç standart (İS) olarak kullanıldı. DS ve İS 281 nm'de ve sırasıyla 2.6 ve 6.2 dakikalarda tespit edildi. Toplam analiz süresi 7 dakika idi. Metot 50 - 1600 ng/mL konsantrasyon aralığında çok iyi bir belirleme katsayısı ile ( $r^2=0.999$ ) doğrusallık gösterdi. Gözlenebilir sınırları ve tayin sınırları sırasıyla 8.95 ng/mL ve 27.12 ng/mL idi. Gün içi kesinlik ve doğruluk sırasıyla 0.93-5.27; 1.74-9.81 aralığında idi. Günler arası kesinlik ve doğruluk ise sırasıyla 2.71-6.64; 2.03-9.16 aralığında idi. Bu metot sağlıklı gönüllülerin (n=12) tek doz oral Voltaren® 75 mg/tablet alımı ile farmakokinetik çalışma süresince DS plazma konsantrasyonlarının belirlenmesi için başarıyla uygulandı ve DS düzeylerinde dikkate değer varyasyonlar gözlemlendi. Çalışmamızda DS'nin eşit dozlarına karşın, 2., 4. ve 6. saatlerdeki plazma düzeylerinde gözlenen anlamlı farklılıklar, özellikle DS metabolizmasından sorumlu majör enzimler olan CYP2C9 ve CYP3A4 polimorfizmlerinden kaynaklanabilecek farmakokinetik farklılıklarla açıklanabilir.

**Anahtar kelimeler:** Diklofenak sodyum, Validasyon, Plazma, Yüksek performanslı sıvı kromatografisi, Ultraviyole dedeksiyon

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## INTRODUCTION

Diclofenac sodium (DS), 2-[(2,6-dichlorophenyl)amino]phenylacetic acid, (Figure 1a) is non-steroidal anti-inflammatory analgesic (NSAID) with potent cyclooxygenase (COX) inhibition activity (1,2). DS has a well documented safety profile, which is comparable to those of other NSAIDs. It inhibits prostaglandin synthesis by inhibition of enzymatic transformation of arachidonic acid into prostaglandins (3,4). This drug is widely used in clinical medicine for the pain control and treatment of rheumatoid arthritis, osteoarthritis, ankylosing spondylitis also acute injury (1). In addition to, it is used to treat chronic pain associated with cancer (2,5). DS has anti-uricosuric feature. The development of Alzheimer disease may be prevent if use of DS which low-dose in long-term (5).

DS is well-absorbed after oral administration with expensive hepatic metabolism. It is extensively (>99.7%) protein bound (albumin) in plasma and serum at therapeutic concentrations. Terminal half life is 1-2 hours. Food can cause a delay in the onset of absorption and a reduction in plasma levels of approximately 30%. After absorption, approximately half of the absorbed dose is metabolized immediately by the liver, due to first pass metabolism. 35% of absorbed DS enters enterohepatic circulation. The distribution volume ( $V_d$ ) of DS is 1.4 L/Kg.  $C_{max}$  is reached at approximately 4 hours.  $T_{max}$  is approximately 0.5-1 h. Elimination is rapid with 90% of the drug clearance taking between 3 to 4 hours. The DS metabolism products, which are mainly 4'-hydroxy (OH) diclofenac and minor monohydroxy metabolites are 3'-OH diclofenac and 5'-OH diclofenac, are excreted by the urine (65%) and biliary (35%) (2,5-7).

The use of DS has been associated with occasional hepatic toxicity (8,9). Although the etiology of this toxicity is not known, clinical evidence suggests that it may be due to an immune (10,11) or a non-immune mechanism (9,12,13). In both cases, covalent modification of liver proteins may play an important role in the etiology of DS hepatotoxicity (14). The fact that, CYP2C9 was known to have a major role in the oxidative metabolism of DS (15,16). It seemed possible that CYP2C9 might metabolically activate DS into a reactive metabolite(s), which may have a role in DS hepatitis in humans. Naproxen which was used as an internal standart in our study, is a propionic acid derivate related to the arylacetic acid group of nonsteroidal anti-inflammatory drugs. Naproxen is chemically, (*S*)-6-methoxy- $\alpha$ -methyl-2-naphthaleneacetic acid (Figure 1b) is commonly used for the reduction of moderate

to severe pain, fever, inflammation, and stiffness. It works by inhibiting both the COX-1 and COX-2 enzymes. Its antipyretic and analgesic effects were related to inhibition of both the COX-1 and COX-2 (17).

Bioavailability, bioequivalence and therapeutic drug monitoring studies have received major attention from the pharmaceutical industry, health authorities and clinic. These studies are performed to evaluate the safety and efficacy of a genetic structure. Which studies as well as drug product development studies require rapid, simple, sensitive and reliable bioanalytical methods to monitoring the target drug in human plasma sample. Also for clinical studies, it is essential to establish accurate, sensitive and selective analytical techniques that permit detection and quantitative measurement of drug entities in biological and pharmaceutical samples (1). Several methods have been reported for determination of DS including spectrophotometric (18,19), spectrofluorimetric (20,21), polarographic (22), conductometric (23), high-performance liquid chromatography (HPLC) (1,24-29), gas chromatography mass spectrometry (GC-MS) (30,31) and capillary electrophoresis (32) in human plasma and other biological fluids. Some of these methods are not suitable for routine analysis because they need sophisticated instruments, not yet available in many routine control laboratories. Since, HPLC-UV assays are reliable, inexpensive and widely utilized; it has appeared to fit best for performing simultaneous separation, quantification and clinical monitoring of DS as a primary concern of this paper.

Compared with other studies the advantages of present method are short run time (total run time 7 min) (26,28), using low biological sample volume (0.7 mL plasma) (26,27), high sensitivity (LOD 8,95  $\mu\text{g/mL}$ ) with small volume samples (1,2), include simple, efficient and inexpensive extraction procedure, high recovery value (98.75-99.32%) (26-28) and using inexpensive chemicals in analytical processes (27). Also this method is used to determine the DS in human plasma samples obtained from twelve healthy volunteers. The originality of this DS pharmacokinetic study is being the only research has ever been done with the contribution of the largest volunteers in Turkey. In addition to this, the method is efficient in analyzing large numbers of plasma obtained for pharmacokinetic study after therapeutic dose of diclofenac. In the present study, our objective is to develop and validate a reliable, simple, fast, and inexpensive HPLC method using UV detection for determination of DS in human plasma with the lower volume sample preparation according to ICH guidelines (33,34) and also to emphasize interpersonal differences in metabolization of DS for correct dosage of by simultaneously monitoring its levels in plasma. The developed method is validated by using linearity, precision, accuracy and sensitivity parameters according to ICH guidelines. For this purpose a group of volunteers, who are the employees of Gülhane Military Medical Academy, were contributed and DS levels in the plasma samples from volunteers were investigated.

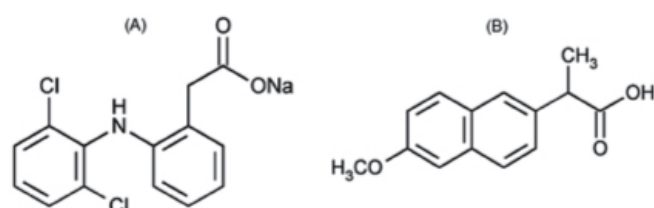


Figure 1. The chemical structure of diclofenac sodium (A) and naproxen (B)

## EXPERIMENTAL

### Chemicals and reagents

Ethyl acetate (EtOAc), acetonitrile (ACN), sodium dihydrogen phosphate dehydrate ( $\text{NaH}_2\text{PO}_4$ ), hydrochloric acid (HCl), orthophosphoric acid ( $\text{H}_3\text{PO}_4$ ) and methanol (MeOH) were purchased from Merck (Darmstadt, Germany). The standards of pharmaceuticals, diclofenac sodium (DS) and naproxen, which is internal standard (IS), were obtained from Sigma-Aldrich (St. Louis, MO, USA). Ultrapure water, which is prepared daily fresh, was used. Prescribed DS tablets, for volunteers, were obtained from pharmaceutical firm Novartis Medical Company (Istanbul, Turkey). All chemicals used in this study were of analytical-grade in the highest purity available, except MeOH and ACN that were HPLC grade.

### Instrumentation and optimized chromatographic conditions

The separation and quantification were performed by HP Agilent 1100 (Santa Clara, CA, USA) high-performance liquid chromatography (HPLC) system, equipped with a UV detector.

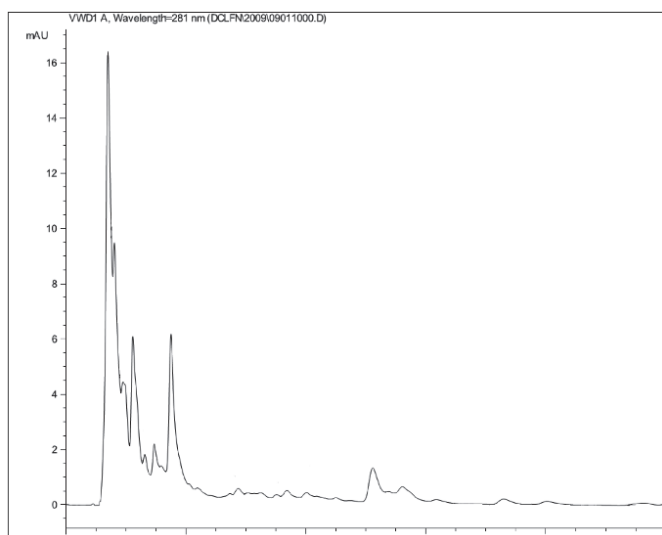


Figure 2. HPLC chromatogram of the blank human plasma

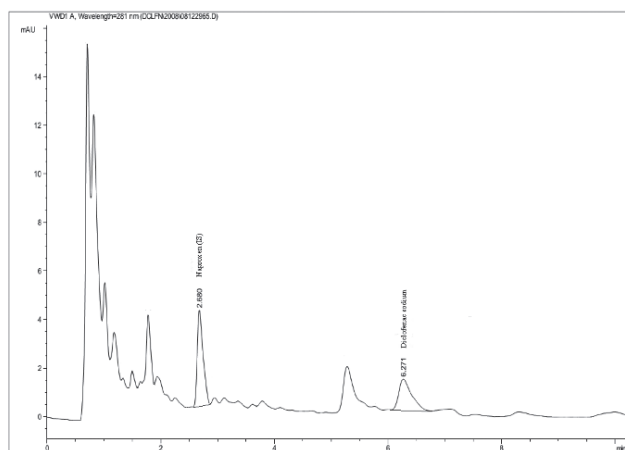


Figure 3. The HPLC chromatogram of DS (and internal standard) in human plasma spiked with  $0.8 \mu\text{g/mL}$  DS ( $t_R=2.680$ ) and  $3 \mu\text{g/mL}$  naproxen ( $t_R=6.271$ ) as an internal standard

Optimum analytic conditions were set after an optimization procedure was performed for column selection, content of mobile phase (MP) and wavelength. Prior to optimization, a standard assay for DS determination offered by the manufacturer was used and each parameter was adjusted while others were fixed.

The system consisted of isocratic pump, manual injector with a loop volume of  $20 \mu\text{L}$ , and a Waters C18 column ( $3.5 \mu\text{m}$ ,  $150 \times 4.6 \text{ mm}$  I.D.) The wavelength at  $281 \text{ nm}$  was chosen for UV detection. The mobile phase, which consists of acetonitrile and  $\text{Na}_2\text{HPO}_4$  buffer ( $42.5:57.5, \text{v/v}$ ) adjusted to  $\text{pH}$  3.16 by  $0.1 \text{ M}$  orthophosphoric acid. It was filtered through a  $0.45 \mu\text{m}$  membrane (Alltech, IL, USA) and degassed in ultrasonic bath for 30 min. An isocratic solution was performed at a flow rate of  $1.5 \text{ mL/min}$  and at room temperature. Peak areas were measured and calculations were carried out considering the internal standard (IS) peak ratios.

### Preparation of stock and working standard solutions

Stock solutions of DS and naproxen (IS) were prepared in methanol solution at the concentrations of  $200 \mu\text{g/mL}$  and  $30 \mu\text{g/mL}$  respectively. Working standards of DS, used for spiked plasma samples, were prepared weekly in the concentrations of  $0.5, 1, 2, 4, 8$  and  $16 \mu\text{g/mL}$  and made by dilution of the stock solutions with methanol. Blank human blood samples, which were obtained from Ankara University Serpil Akdağ Blood Center, centrifuged at  $3000 \text{ g}$  for 5 min to separate the plasma. Human plasma samples and all working solutions were stored at  $-20^\circ\text{C}$  until the analysis were carried out. These were checked chromatographically for purity before experiments and utilized (Figure 2) as quality control specimens for validation and optimization process. Their stability was checked before and after the injections of every sample set.

### Preparation of sample

Blank plasma samples ( $0.7 \text{ mL}$ ) were placed into clean glass tubes which is containing  $1 \text{ mL}$  HCl ( $1 \text{ M}$ ). After that each of these tubes,  $100 \mu\text{L}$  IS ( $30 \mu\text{g/mL}$ ) and  $100 \mu\text{L}$  of diclofenac sodium solutions, which were the concentrations of  $0.5, 1, 2, 4, 8$  and  $16 \mu\text{g/mL}$ , were spiked. Reaching the total volume to  $1.1 \text{ mL}$  and achieving 10 times diluted mixtures; yielding analyte concentrations as calibration samples namely  $50, 100, 200, 400, 800, 1600 \text{ ng/mL}$  diclofenac sodium and  $3000 \text{ ng/mL}$  naproxen. Different from the calibration solutions,  $100 \mu\text{L}$  MeOH was added instead of DS solution for the samples from patient's plasma.

### Extraction procedure

Liquid-liquid extraction was applied for human plasma samples with acidic hydrolysis (35,36). A volume of  $0.1 \text{ mL}$  of  $1 \text{ M}$  hydrochloric acid was added to the samples and mixed at  $900 \text{ g}$  for 30 second. A volume of  $2 \text{ mL}$  of ethyl acetate was added to sample tubes and mixed at  $900 \text{ g}$  for 2 min and then centrifuged at  $3000 \text{ g}$  for 5 min. This extraction procedure

was repeated twice. Supernatant was collected to another sample tube and dried under nitrogen at 40 °C. Residue was dissolved in 50 µL volume of mobile phase, in order to be ready for injection, and then loaded into HPLC, by 20 µL which is loop volume.

#### Volunteers

Volunteers, who were consist of 12 healthy Gülhane Military Medical Academy employees, participated in this study. The average age of participants was 32±8.12. After ingestion of tablets, which including 75 mg DS, bloodletting was performed at 2<sup>nd</sup>, 4<sup>th</sup> and 6<sup>th</sup> hours. The plasma samples were collected severally and samples were stored at -20°C. This study was approved by *Local Ethical Committee of Gülhane Military Medical Academy* in 12/11/2008 and decision number is 90.

## RESULTS AND DISCUSSION

#### Selection of internal standard

Naproxen, which was developed in the 1970s, is a non-selective COX inhibitor is a nonsteroidal anti-inflammatory drug (NSAID) of the propionic acid class and is commonly used for relief of a wide variety of pain, fever, swelling and stiffness (17). Naproxen was chosen as internal standard due to its similar physico-chemical properties with diclofenac sodium (DS). On the other hand, the chromatographic pre-study have demonstrated that, naproxen might used as a reliable internal standard (Figure 1-3).

#### Validation of method

The analytical method was validated to demonstrate the selectivity, specificity, linearity, recovery, limit of

**Table 1. Recovery of DS in human plasma (n=3)**

| Theoretical concentration ng/mL | Mean areas of diclofenac sodium (DS) and internal standard (IS) |                            |               |                           |                            |               |                   |       |
|---------------------------------|---|----------------------------|---------------|---------------------------|----------------------------|---------------|-------------------|-------|
|                                 | First group extraction  |                            |               | Second group extraction   |                            |               | Mean Recovery (%) |       |
|                                 | DS added before extraction                                      | IS added before extraction | Areas (DS/IS) | DS added after extraction | IS added before extraction | Areas (DS/IS) |                   |       |
| 200                             | 1   | 4.59                       | 33.78         |                           | 4.87                       | 33.34         | 0.146             |       |
|                                 | 2   | 4.64                       | 30.40         |                           | 4.99                       | 33.35         | 0.150             |       |
|                                 | 3   | 4.83                       | 31.92         |                           | 4.72                       | 32.29         | 0.146             |       |
|                                 | Mean  | 4.69                       | 32.03         | 0.146                     | 4.86                       | 32.99         | 0.147             | 99.32 |
| 400                             | 1   | 10.3                       | 31.7          |                           | 10.99                      | 33.33         | 0.330             |       |
|                                 | 2   | 10.02                      | 33.47         |                           | 10.01                      | 32.35         | 0.309             |       |
|                                 | 3   | 10.72                      | 33.25         |                           | 10.65                      | 33.65         | 0.317             |       |
|                                 | Mean  | 10.35                      | 32.81         | 0.315                     | 10.55                      | 33.11         | 0.319             | 98.75 |
| 800                             | 1   | 21.12                      | 31.92         |                           | 20.4                       | 32.28         | 0.632             |       |
|                                 | 2   | 20.11                      | 31.53         |                           | 21.78                      | 32.23         | 0.676             |       |
|                                 | 3   | 20.58                      | 32.36         |                           | 20.62                      | 32.11         | 0.642             |       |
|                                 | Mean  | 20.60                      | 31.94         | 0.645                     | 20.93                      | 32.21         | 0.650             | 99.23 |

DS: Diclofenac sodium, IS: Internal standard

**Table 2. Precision and accuracy of the assay for DS (n=5)**

| Theoretical Conc. ng/mL | Intra-day          |       |       |      | Inter-day       |       |      |      |
|-------------------------|--------------------|-------|-------|------|-----------------|-------|------|------|
|                         | DS/IS              |       |       |      | DS/IS           |       |      |      |
|                         | Estimated DS Conc. | SD    | RSD%  | RE%  | Estimated Conc. | SD    | RSD% | RE%  |
| 50                      | 54.91              | 2.20  | 4.01  | 9.81 | 54.58           | 1.82  | 3.33 | 9.16 |
| 100                     | 97.29              | 0.91  | 0.93  | 2.71 | 93.86           | 6.13  | 6.53 | 6.14 |
| 200                     | 184.66             | 6.90  | 3.74  | 7.67 | 182.18          | 5.10  | 2.80 | 8.91 |
| 400                     | 429.06             | 13.02 | 3.035 | 7.27 | 430.96          | 28.60 | 6.64 | 7.74 |
| 800                     | 755.60             | 21.68 | 2.87  | 5.55 | 775.96          | 21.59 | 2.78 | 3.01 |
| 1600                    | 1572.20            | 82.90 | 5.27  | 1.74 | 1567.60         | 42.44 | 2.71 | 2.03 |

DS: Diclofenac sodium, IS: Internal standard, SD: Standard deviation, RSD%: Percent relative standard deviation, RE%: Percent relative error

detection (LOD) and limit of quantification (LOQ). Intra- and inter-day validation protocol was applied considering reproducibility of method and instrument to obtain accurate and precise measurements in agreement with Conference on Harmonization Guidelines (33,34).

#### Recovery

The recovery of extraction procedures from human plasma is determined by comparing areas of DS and IS. DS and the IS first joined together in the plasma samples before extraction was performed three different extractions (200, 400, 800 ng/mL diclofenac sodium and 3000 ng/mL naproxen). Also DS and the IS joined together in the plasma samples after extraction in three different plasma samples. The areas of DS and IS are calculated and compared then % efficiency is obtained. Three individual replicates of spiked samples at mid-concentrations (200, 400, 800 ng/mL DS) were prepared with internal standard (n=3). The extraction procedure was conducted as described previously. Peak area ratios were compared and recoveries were calculated as 98.75%-99.32% for plasma (Table 1).

#### Precision

Precision, defined as relative standard deviation (RSD%), was determined by five individual replicates at six different concentrations (50, 100, 200, 400, 800 and 1600 ng/mL) (n=5). Interday and intraday precisions of method and the instrument were calculated. Results were showed that, RSD% were less than 6.7% both for intra- and interday precisions (Table 2). Analytical instrument precision was also displayed higher precisions namely less than 6.3% considering intra- and interday performances (Table 3).

#### Accuracy

Accuracy defines as the measure of how close the experimental value is to the true value. Accuracy is the degree of veracity while precision is the degree of reproducibility and accuracy expressed as the relative error (RE%) of the estimated concentrations. Table 2 shows RE% of estimated concentrations for intra- and interday accuracy of assays. The instrument was as accurate as assay which is displayed in Table 3. The accurate and precise measurements pointed at good reproducibility for the method and the instrument.

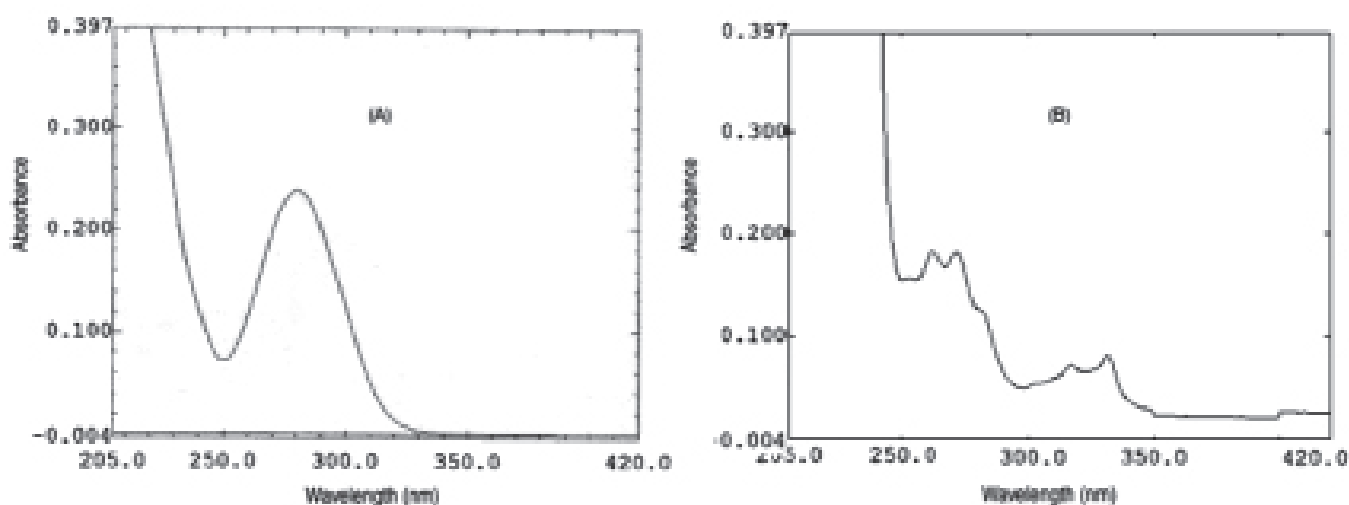


Figure 4. The obtained UV spectrums and maximum absorption wavelengths of DS (A) and Naproxen (B)

Table 3. Precision and accuracy of analytical instrument, HPLC – UV, for determination of DS (n=5)

| Theoretical Conc.<br>ng/mL | Intra-day<br>DS/IS |        |      |       | Inter-day<br>DS/IS |       |      |       |
|----------------------------|--------------------|--------|------|-------|--------------------|-------|------|-------|
|                            | Estimated Conc.    | SD     | RSD% | RE%   | Estimated Conc.    | SD    | RSD% | RE%   |
| 50                         | 60.16              | 1.12   | 1.86 | 20.32 | 55.65              | 2.71  | 4.87 | 11.30 |
| 100                        | 112.55             | 5.68   | 5.05 | 12.55 | 95.17              | 4.81  | 5.06 | 4.83  |
| 200                        | 223.23             | 11.16  | 5.00 | 11.61 | 184.99             | 11.52 | 6.23 | 7.51  |
| 400                        | 443.71             | 12.066 | 2.72 | 10.93 | 418.57             | 14.90 | 3.56 | 4.64  |
| 800                        | 915.36             | 15.39  | 1.68 | 14.42 | 756.65             | 11.27 | 1.49 | 5.42  |
| 1600                       | 1829.62            | 78.19  | 4.27 | 14.35 | 1577.49            | 40.15 | 2.55 | 1.41  |

DS: Diclofenac sodium, IS: Internal standard, SD: Standard deviation, RSD%: Percent relative standard deviation, RE%: Percent relative error

### Specificity and selectivity

Method for plasma demonstrated excellent chromatographic specificity with no endogenous interference at the retention times of the IS and DS (2.6 and 6.2 min, respectively). Representative chromatograms for human plasma spiked with DS (800 ng/mL) and the IS (3000 ng/mL) are shown in Figure 2 and Figure 3. The most appropriate wavelength at 281 nm was chosen for UV detection of DS and IS (Figure 4).

### Limit of detection and quantification

Limit of detection (LOD) and limit of quantification (LOQ) were determined based on the standard deviation of the response and the slope of the calibration curve, according to ICH guidelines (LOD= 3.3  $\sigma$ /S, LOQ= 10  $\sigma$ /S where  $\sigma$  is the standard deviation of the response and S is the slope of the calibration curve) (33,34). LOD and LOQ values were

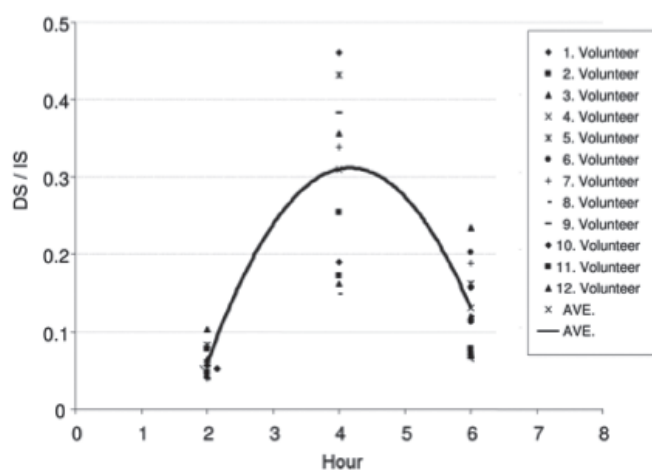


Figure 5. DS levels in plasma of volunteers after administration of Voltaren® SR 75 mg single oral dose

**Table 4. Diclofenac sodium levels in 2<sup>nd</sup>, 4<sup>th</sup> and 6<sup>th</sup> times of volunteers plasma (ng/mL)**

| Number of volunteers | 2 <sup>nd</sup> hour | 4 <sup>th</sup> hour | 6 <sup>th</sup> hour |
|----------------------|----------------------|----------------------|----------------------|
| 1                    | 82.28                | 248.82               | 153.40               |
| 2                    | 70.95                | 329.18               | 104.95               |
| 3                    | 92.27                | 213.86               | 160.60               |
| 4                    | 86.10                | 398.09               | 92.49                |
| 5                    | 114.78               | 550.89               | 214.11               |
| 6                    | 63.22                | 658.37               | 264.44               |
| 7                    | 61.74                | 435.47               | 246.66               |
| 8                    | 60.67                | 196.24               | 161.02               |
| 9                    | 80.14                | 490.15               | 92.74                |
| 10                   | 73.99                | 586.69               | 208.64               |
| 11                   | 109.60               | 227.44               | 109.07               |
| 12                   | 139.68               | 456.46               | 295.23               |
| Mean                 | 86.29                | 399.31               | 175.28               |

calculated from plasma samples and found as 8.95 ng/mL and 27.12 ng/mL respectively.

### Linearity

After establishing the chromatographic conditions, separate calibration curves were prepared for plasma over a DS concentration range of 50-1600 ng/mL. For each concentration 5 individual replicates were injected and linearity was obtained for both methods with the determination coefficients ( $r^2$ ) over 0.999.

## CONCLUSIONS

In this study, simultaneous procedure of HPLC-UV method was proposed with simple extraction of sample yielding good recovery, selective chromatographic separation, and sensitive UV detection with enhanced sensitivity and accuracy of determination for analysis of DS. Therefore, to achieve quality separation of analyte in a reasonable analysis time, acceptable chromatographic factors were adjusted. The mobile phase composition and the pH were optimized. The mobile phase was a phosphate buffer adjusted to acidic pH and containing ACN as the organic modifier. Baseline separation of the analyte (and the IS) was achieved in less than 7 min. The method was validated in terms of linearity, accuracy, precision, reproducibility, quantification and detection limits in accordance with internationally accepted guidelines which are ICH. Analysis for all analytes demonstrates very precise and accurate results, even for inter-day assays which allow determining therapeutic and toxic concentration levels.

The study results showed that, established HPLC-UV method is applicable for the therapeutic drug monitoring bioavailability, bioequivalence study and applicable as a reference method in routine monitoring for toxicological and/or analytic purposes in research laboratories.

### Diclofenac sodium levels of volunteers

Although considerable individual differences were observed, representative curves plotted by drug levels of each volunteer were similar. Variations in the DS levels detected in the volunteers were graphed and tabled (Table 4, Figure 5). After ingestion of DS, reaches the maximum level in plasma, at 4<sup>th</sup> hour. Even though, each volunteer was medicated with equivalent single oral dose, DS levels in plasma were quite dissimilar. Variations may arise from because of enzyme differences which are CYP2C9 and CYP3A4, mainly enzymes responsible for DS metabolism. In this respect, for an application of an effective dosage, the enzyme polymorphism should be taken under consideration in medical therapies.

According to monitored drug levels, differences in concentration rates of DS in human plasma can be expressed and the variation of plasma concentration can be based on hepatic enzyme polymorphism. DS is metabolized mainly by CYP2C9 and CYP3A4. The resulting data may also be considered as a pre-research for pharmacogenetic



polymorphism in this group of volunteers. The analytes were then quantified using HPLC technique, which provided good sensitivity and selectivity. The method has been shown to provide good reproducible recoveries and low limits of detection that allow the accurate quantification of the DS in plasma samples. As the DS level increases in biological material, the expected medical effects also increase. Since, the optimum drug use and maximum resultant effect is the main objective, this report offers clinical application of proposed method for monitorizations of DS in human plasma.

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# Composition of the Essential Oils of *Juniperus oxycedrus* L. subsp. *oxycedrus* Growing in Turkey

## Türkiye’de Yetişen *Juniperus oxycedrus* L. subsp. *oxycedrus*’un Uçucu Yağ Bileşikleri

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### ABSTRACT

In this study, the chemical compositions of the leaves, berries and twigs essential oils of *J. oxycedrus* L. subsp. *oxycedrus*, collected in Turkey, were determined. The oils were analyzed by GC and GC-MS. 15-21 volatile compounds were identified of the leaves, berries and twigs essential oils representing 82.4-98.0% of the total oils. The essential oils were obtained from leaves, berries and twigs by yielding 0.02%, 2.12% and 0.01%, resp. The major compounds were determined manoyl oxide (32.8%) and caryophyllene oxide (11.9%) in leaf oil, myrcene (44.6%),  $\alpha$ -pinene (19.9%) and germacrene D (15.5%) in berry oil, manoyl oxide (35.4%) and caryophyllene oxide (16.8%) in twig oil.

**Key words:** *Juniperus oxycedrus*, GC and GC/MS, Essential oil

### ÖZ

Bu çalışmada, Türkiye’den toplanan *J. oxycedrus* subsp. *oxycedrus* türünün yaprak, meyve ve ince dallarından elde edilen uçucu yağlarının kimyasal içerikleri belirlenmiştir. Uçucu yağlar GC ve GC/MS cihazları aracılığıyla analiz edilip, toplam yağın %82.4-98.0 içeriği 15-21 arasında değişen uçucu bileşen ile yaprak, meyve ve ince dal üzerinde tespit edilmiştir. Uçucu yağlar yaprak, meyve ve ince dalardan sırasıyla %0.02, %2.12 ve %0.01 verimleri ile elde edilmiştir. Ana bileşenler yaprak yağında %32.8 manoil oksit ve %11.9 karyofillen oksit, meyve yağında %44.6 mirsen, %19.9  $\alpha$ -pinen ve %15.5 germakren D, ince dal yağında %35.4 manoil oksit ve %16.8 karyofillen oksit belirlenmiştir.

**Anahtar kelimeler:** *Juniperus oxycedrus*, GC and GC/MS, Uçucu Yağ

### INTRODUCTION

The genus *Juniperus* L. belongs to the Cupressaceae family, representing about 70 species all over the world, and widely distributed throughout the forests of the temperate and cold regions of the northern Hemisphere, from the far north to the Mediterranean. *Juniperus* L. is represented in Turkey by 7 species and 14 taxa. *Juniperus oxycedrus* has two subspecies – subsp. *oxycedrus* and subsp. *macrocarpa* – in Turkey (1,2).

Several studies have reported the chemical composition of leaves, berries and twigs and their the essential oils (EOs) of *J. oxycedrus* L. subsp. *oxycedrus* (Table 1) (3-16).

In the present work, leaves, berries and twigs EOs of *J. oxycedrus* subsp. *oxycedrus* were investigated to chemical compositions of plants collected from Eskişehir: Hekimdağ in Turkey. In the study, the oils were obtained by hydrodistillation.

The oils were analyzed by gas chromatography (GC) and gas chromatography/ mass spectrometry (GC/MS).

### EXPERIMENTAL

#### *Plant material*

*J. oxycedrus* subsp. *oxycedrus* was collected from Eskişehir: Hekimdağ, in Turkey on 15 March 2015. Voucher specimens are kept at the Herbarium of Pharmacy Faculty, Anadolu University, Turkey (ESSE 14987).

#### *Isolation of essential oils*

The oils were obtained by hydrodistillation for 3 hours using Clevenger apparatus. The essential oils were stored at 4°C in the dark until analyzed. The oils were analyzed by capillary GC and GC/MS using a Agilent GC-MSD system.

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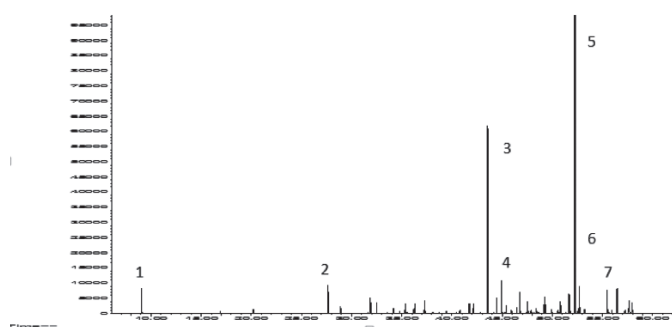
**Table 1. EO yield and chemical composition of *Juniperus oxycedrus* L. subsp. *oxycedrus* (3-16)**

| Part of plant | Yield %        | Major compounds %   | Ref. |
|---------------|----------------|---|------|
| Leaf          | 0.7<br>HD      | $\alpha$ -Pinene (42.9%)<br>Limonene (17.8%)<br>Caryophyllene oxide (5.1%)<br>$\beta$ -Myrcene (3.9%)<br>$\beta$ -Pinene (3.8%)                             | 3    |
| Leaf          | 0.01<br>HD     | $\alpha$ -Pinene (17.1%)<br>13- <i>epi</i> -Manoyl oxide (12.5%)<br>( <i>Z</i> )-6-pentadecen-2-one (11.5%)   | 4    |
| Leaf          | 0.1<br>HD      | <i>trans</i> -Pinocarveol (7.0%)<br><i>cis</i> -Verbenol (6.3%)<br>Manoyl oxide (6.0%)  | 5    |
| Leaf          | 1.66<br>HD     | $\alpha$ -Pinene (31.25%)<br>Sabinene (5.21%)<br>Limonene (5.02%)   | 6    |
| Leaf          | 0.2<br>HD      | Limonene (30.0%)<br>$\alpha$ -Pinene (26.3%)<br>( <i>Z</i> , <i>E</i> )-Farnesol (5.1%)<br>Salvial-4 (14)-en-1-one (4.9%)                                   | 7    |
| Leaf          | 0.4±0.14<br>HD | $\alpha$ -Pinene (49.46%)<br>Germacrene D (8.96%)<br>13- <i>epi</i> -Manoyl oxide (3.62%)   | 8    |
| Leaf          | 0.2-0.5<br>HD  | $\alpha$ -Pinene (41.3%)<br>$\alpha$ -Phellandrene (8.2%)<br><i>p</i> -Cymene (6.2%)<br>Manoyl oxide (5.3%)   | 9    |
| Berry         | -<br>SPME      | $\alpha$ -Pinene (88.44%)<br>$\beta$ -Myrcene (6.71%)<br>$\beta$ -Pinene (2.07%)  | 10   |
| Berry         | 0.97<br>HD     | $\alpha$ -Myrcene (23.4%)<br>$\alpha$ -Pinene (16.7%)<br>Citronellol (16.3%)<br>$\beta$ -Caryophyllene (6.3%)   | 11   |
| Berry         | 2.21<br>HD     | Citronellol (26.8%)<br>$\alpha$ -Myrcene (24.3%)<br>$\alpha$ -Pinene (14.4%)<br>Limonene (9.3%)   | 11   |
| Berry         | 0.7-1.2<br>HD  | $\beta$ -Myrcene (56.87±4.6%)<br>$\alpha$ -Pinene (14.84±2.9%)<br>DL-Limonene (5.96±0.6%)   | 12   |
| Berry         | 0.4-0.7<br>HD  | $\beta$ -Myrcene (54.06±6.1%)<br>$\alpha$ -Pinene (10.22±2.7%)<br>DL-Limonene (9.20±1.7%)<br>Germacrene D (8.56±1.7%)<br>( <i>E</i> )-Nerolidol (5.94±1.6%) | 12   |
| Berry         | 0.8-1.5<br>HD  | $\beta$ -Myrcene (49.75±3.8%)<br>$\alpha$ -Pinene (16.50±2.3%)<br>DL-limonene (13.82±2.7%)  | 12   |
| Berry         | 0.9-1.5<br>HD  | $\beta$ -Myrcene (56.97±3.7%)<br>$\alpha$ -Pinene (19.55±3.7%)<br>$\alpha$ -Cadinol (3.92±1.6%)<br>Germacrene D (3.74±0.2%)                                 | 12   |

**Table 1 continued**

|         |               |   |    |
|---------|---------------|---|----|
| Berry   | 1.2-1.8<br>HD | $\beta$ -Myrcene (45.50±3.0%)<br>$\alpha$ -Pinene (36.64±2.0%)<br>DL-Limonene (5.75±2.1%)<br>Germacrene D (3.65±0.6%)   | 12 |
| Berry   | 0.72<br>HD    | $\alpha$ -Pinene (27.4%)<br>$\beta$ -Myrcene (18.9%)<br>$\alpha$ -Phellandrene (7.1%)<br>Limonene (6.7%)  | 13 |
| Berry   | 2.5<br>HD     | $\alpha$ -Pinene (27.4%)<br>$\beta$ -Myrcene (18.9%)  | 14 |
| Berry   | 0.5<br>HD     | $\alpha$ -Pinene (60.60±3.33%)<br>$\beta$ -Myrcene (24.97±1.76%)<br>$\gamma$ -Murolene (5.19%)  | 15 |
| Berry   | 0.45<br>SFE   | Germacrene D (13.8%)<br>$\alpha$ -Pinene (10.5%)<br>$\beta$ -Myrcene (8.1%)   | 16 |
| Needles | -<br>HD       | $\alpha$ -Pinene (92.22%)<br>$\beta$ -Myrcene (2.46%)<br>$\beta$ -Pinene (1.79%)  | 10 |
| Wood    | 0.68<br>HD    | $\delta$ -Cadinene (14.5%)<br><i>cis</i> -Thujopsene (9.2%)<br>$\alpha$ -Murolene (4.9%)<br>Cadalene (3.7%)<br>Eremophilene (2.5%)<br>$\alpha$ -Cedrol (2.2%) | 13 |

HD: Hidrodistillation, SFE: Supercritical carbon dioxide extraction, SPME: Solid Phase Micro Extraction



**Figure 1.** Chromatogram of the twigs essential oil of *J. oxycedrus* L. subsp. *oxycedrus*

1.  $\alpha$ - Pinene, 2.  $\alpha$ - Cubebene, 3. Caryophyllene oxide, 4. Humulene epoxide-II, 5. Manoyl oxide, 6. Caryophyllenol II, 7. Dodecanoic acid

#### Gas Chromatography (GC) and Gas Chromatography - Mass Spectrometry (GC/MS) analysis

The GC/MS analysis was carried out with an Agilent 5975 GC-MSD system. Innowax FSC column (60 m x 0.25 mm, 0.25 mm film thickness) was used with helium as carrier gas (0.8 mL/min). GC oven temperature was kept at 60°C for 10 minutes and programmed to 220°C at a rate of 4°C/min, and

kept constant at 220°C for 10 minutes and then programmed to 240°C at a rate of 1°C/minutes. Split ratio was adjusted 40:1. The injector temperature was at 250°C. MS were taken at 70 eV. Mass range was from m/z 35 to 450.

**Table 2. Composition of the EOs of *Juniperus oxycedrus* L. subsp. *oxycedrus***

| RRI  | Compounds   | Twigs % | Berries % | Leaves % |
|------|---|---------|-----------|----------|
| 1032 | $\alpha$ -Pinene                                    | 2.4     | 19.9      | 1.4      |
| 1132 | Sabinene  | -       | 1.4       | -        |
| 1174 | Myrcene   | -       | 44.6      | -        |
| 1190 | Sylvestrene   | 0.7     | -         | -        |
| 1203 | Limonene  | -       | 2.7       | -        |
| 1213 | 1,8-Cineole   | -       | -         | -        |
| 1218 | $\beta$ -Phellandrene                               | -       | 0.7       | -        |
| 1280 | <i>p</i> -Cymene                                    | 0.5     | -         | -        |
| 1290 | Terpinolene   | -       | 0.5       | -        |
| 1466 | $\alpha$ -Cubebene                                  | 2.0     | 0.3       | -        |
| 1604 | Isobornyl acetate                                   | 1.6     | 0.3       | 1.2      |
| 1612 | $\beta$ -Caryophyllene                              | 1.5     | 4.6       | 3.7      |
| 1668 | ( <i>Z</i> )- $\beta$ -Farnesene                    | -       | 0.5       | -        |
| 1687 | $\alpha$ -Humulene                                  | -       | 3.1       | -        |
| 1704 | <i>g</i> -Muuroleone                                | -       | 0.3       | -        |
| 1706 | $\alpha$ -Terpineol                                 | 1.9     | -         | -        |
| 1726 | Germacrene D  | -       | 15.5      | 5.7      |
| 1740 | $\alpha$ -Muuroleone                                | 1.3     | 0.7       | 3.2      |
| 1773 | <i>d</i> -Cadinene                                  | 1.4     | 1.8       | 4.1      |
| 1776 | <i>g</i> -Cadinene                                  | -       | -         | 1.6      |
| 1941 | $\alpha$ -Calacorene                                | 0.7     | -         | 1.8      |
| 2008 | Caryophyllene oxide                                 | 16.8    | -         | 11.9     |
| 2050 | ( <i>E</i> )-Nerolidol                              | 1.4     | -         | -        |
| 2071 | Humulene epoxide-II                                 | 3.4     | -         | 4.7      |
| 2148 | Cedrol  | 2.9     | -         | -        |
| 2243 | Torilenol   | 1.2     | -         | 2.5      |
| 2255 | $\alpha$ -Cadinol                                   | -       | 1.1       | -        |
| 2256 | Cadalene  | 2.8     | -         | 2.2      |
| 2316 | Caryophylladienol I                                 | 1.8     | -         | -        |
| 2376 | Manoyl oxide  | 35.4    | -         | 32.8     |
| 2392 | Caryophyllenol II (=Caryophylla-2(12),6-dien-5b-ol) | 4.5     | -         | -        |
| 2503 | Dodecanoic acid (=lauric acid)                      | 4.2     | -         | 3.1      |
| 2524 | Abietatriene  | 2.5     | -         | 2.5      |

RRI: Relative Retention Indices calculated against n-alkanes

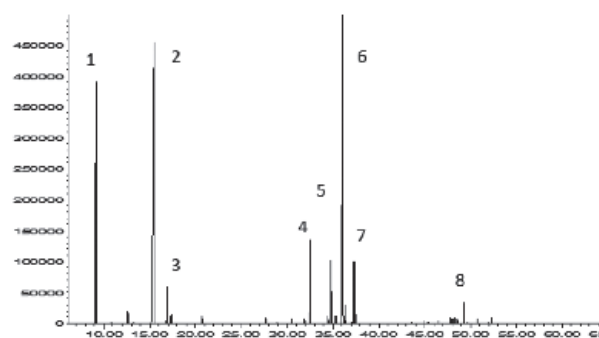
The GC analysis was carried out using an Agilent 6890 N GC system. In order to obtain the same elution order with GC/MS, simultaneous injection was done by using the same column and appropriate operational conditions. FID temperature was 300°C.

#### Identification of compounds

The components of essential oils were identified by comparison of their mass spectra with those in the Baser Library of Essential Oil Constituents, Wiley GC/MS Library, Adams Library, MassFinder Library and confirmed by comparison of their retention indices. Alkanes were used as reference points in the calculation of relative retention indices (RRI). Relative percentage amounts of the separated compounds were calculated from FID chromatograms.

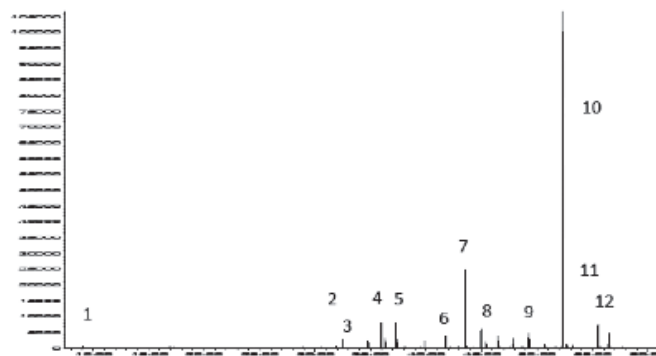
## RESULTS AND DISCUSSION

The EOs were obtained from leaves, berries and twigs by yielding 0.02%, 2.12% and 0.01%, respectively. The list of compounds identified in the hydrodistilled leaves, berries and twigs of *J. oxycedrus* L. subsp. *oxycedrus* with their relative percentages and retention indices are given in Table 2.



**Figure 2.** Chromatogram of the berries essential oil of *J. oxycedrus* L. subsp. *oxycedrus*

1.  $\alpha$ - Pinene, 2. Myrcene, 3. Limonene, 4.  $\beta$ - Caryophyllene, 5.  $\alpha$ - Humulene, 6. Germacrene D, 7. *d*- Cadinene, 8.  $\alpha$ - Cadinol



**Figure 3.** Chromatogram of the leaves essential oil of *J. oxycedrus* L. subsp. *oxycedrus*

1.  $\alpha$ - Pinene, 2. Bornyl acetate, 3.  $\beta$ - Caryophyllene, 4. Germacrene D, 5. *d*- Cadinene, 6.  $\alpha$ - Calacorene, 7. Caryophyllene oxide, 8. Humulene epoxide-II, 9. Cadalene, 10. Manoyl oxide, 11. Dodecanoic acid, 12. Abietatriene

In our study, 15-21 volatile compounds were identified of the leaves, berries and twigs EOs representing 82.4-98.0% of the total oils. Twig oil composition was not found in previous studies. Manoyl oxide (35.4%) and caryophyllene oxide (16.8%) were identified as major constituents in twig oil (Figure 1), myrcene (44.6%),  $\alpha$ -pinene (19.9%) and germacrene D (15.5%) in berry oil (Figure 2), manoyl oxide (32.8%) and caryophyllene oxide (11.9%) in leaf oil (Figure 3).

As seen in previous studies Table 1, the leaf oils were characterized by the presence of  $\alpha$ -pinene, t-pinocarveol and limonene as main constituents (3-10). But, in our study, the occurrence of manoyl oxide and caryophyllene oxide was interesting. While some constituents like  $\alpha$ -pinene and myrcene were found in berry oil samples (11-15), germacrene D was only in one sample (16).

Medini *et al.* have reported  $\alpha$ -pinene, germacrene D, myrcene, abietadiene and *cis*-calamenene as main constituents of the EOs of the berries of *Juniperus oxycedrus* L. subsp. *rufescens* (L.K.) and *Juniperus oxycedrus* L. subsp. *macrocarpa* (S. & M.) Ball. (17).

Sezik *et al.* have reported manoyl oxide (21.9%) and  $\alpha$ -pinene (11.3%) as main constituents in leaf oil of *J. oxycedrus* subsp. *macrocarpa* from Eskişehir (18).

Variability of the oil composition in different populations of the same plant species might be attributed mainly to genetic diversity (19). Chemical composition and the main components of EOs *J. oxycedrus* have differentiate chemotype due to containing different climatic conditions of a large geographical diversity.

## ACKNOWLEDGEMENTS

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# Formulation and Evaluation of Lansoprazole Loaded Nanosponges

## Lasoprazol Yüklü Nanosüngerlerin Formülasyonu ve Değerlendirilmesi

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### ABSTRACT

Lansoprazole is proton pump inhibitor which extensively degraded in acidic pH conditions. Lansoprazole loaded nanosponges were prepared by Emulsion solvent diffusion method using ethylcellulose, PVA and pluronic F-68 and dichloromethane as a solvent. The prepared nanosponges were evaluated for percentage yield, incorporation efficiency, particle size, drug polymer compatibility, scanning electron microscopy and *in-vitro* drug release. SEM studies confirmed their porous structure with number of nanochannels. The FTIR spectra showed stable character of lansoprazole in mixture of polymers and revealed the absence of drug polymer interactions. DSC study revealed that drug was involved in complexation with nanosponges. The average particle size of lansoprazole nanoparticles was found to be in the range of 83.4 nm to 190.69 nm. The negative zeta potential values were attained to ensure a good stability of nanosponges. The drug release from nanosponges was found to extended upto 12 h. The optimized nanosponges were formulated in to enteric coated tablet and evaluated for weight variation, hardness, friability and dissolution studies. *In-vitro* release of drug from enteric coated tablet follows zero order and showed controlled release behavior for a period of 24 h. The data obtained in this study suggests that nanosponges of lansoprazole are promising for controlled drug delivery, which can reduce dosing frequency.

**Key words:** Lansoprazole, Ethylcellulose, Pluronic F-68, Zero order, Nanosponges, Fickian release

### ÖZ

Lansoprazol asidik pH koşullarında yoğun şekilde bozulan proton pompası inhibitörüdür. Lansoprazol yüklü nanosüngerler etilselüloz, PVA ve pluronic F68 ve solvan olarak da diklorometan kullanılarak emülsiyon çözücü difüzyon metodu ile hazırlandı. Hazırlanan nanosüngerler yüzde verim, yükleme etkinliği, partikül büyüklüğü, etkin madde polimer geçimliliği, taramalı elektron mikroskobu ve *in vitro* etkin madde salımı açısından değerlendirildi. SEM çalışmaları çok sayıda nanokanallar ile poröz yapılarını doğruladı. FTIR spektrumu polimer karışımı içinde lansoprazolün stabilitesini ve etkin madde polimer etkileşimi olmadığını gösterdi. DSC çalışması etkin maddenin nanosüngerle kompleks halinde olduğu ortaya çıkardı. Lansoprazol nanopartiküllerin ortalama partikül büyüklüğü 83.4 nm ile 190.69 nm aralığında bulundu. Negatif zeta potansiyel değerlerine ulaşılması nanosüngerlerin iyi stabilite göstermelerini sağladı. Etkin maddenin nanosüngerlerden salımının 12 saate kadar uzatıldığı bulundu. Optimize edilmiş nanosüngerler enterik kaplı tabletler halinde formüle edildi ve ağırlık sapması, sertlik, friabilite ve çözünme çalışmaları değerlendirildi. Enterik kaplı tabletlerden etkin maddenin *in vitro* salımı sıfır dereceye uyumlu ve 24 saatlik periyotda kontrollü salım davranışı gösterdi. Bu çalışmadan elde edilen veriler gösterdi ki lansoprazol nanosüngerleri doz sıklığını azaltabilecek kontrollü salım için gelecek vaad etmektedir.

**Anahtar kelimeler:** Lansoprazol, Etilselüloz, Pluronic F68, Sıfır derece, Nanosünger, Fick salımı

### INTRODUCTION

The drug delivery technology has certainly a new interest for drugs by providing them new life through their therapeutic targets. Nowadays, targeting drug delivery is the major problem which is being faced by the researchers. Target oriented drug administration with improvements in therapeutic efficacy,

reduction in side effects and optimized dosing regimen, shall be the leading trends in the area of therapeutics. Targeted drug delivery implies for selective and effective localization of pharmacologically active moiety at pre identified target in therapeutic concentration, while restricting its access to non-target normal cellular linings and thus minimizing toxic effects and maximizing therapeutic index of the drug (1).

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Nanosponges have emerged as one of the most promising fields of science because of their perceived application in controlled drug delivery. Nanosponge delivery system can precisely control the release rates or target drugs to a specific body site and have an enormous impact on the health care system. This nanosized delivery system has definite advantages for the purpose of drug delivery because of its high stability, high carrier capacity and feasibility of incorporation of both hydrophilic and hydrophobic substances. The application of nanosponges for targeted and localized delivery of therapeutic agents is the driving force for the research in this area (2).

The sponge acts as a three-dimensional network or scaffold. The backbone is long-length polyester. It is mixed in solution with cross-linkers to form the polymer. The net effect is to form spherically shaped particles filled with cavities where drug molecules can be stored. The polyester is biodegradable, so it breaks down gradually in the body. As it breaks down, it releases its drug payload in a predictable fashion. The nanosponges can be synthesized to be specific size and to release drugs over time by varying proportions of cross-linker to polymer. The main limitation of nanosponges is their ability to include only small molecules (3).

Nanosponges are solid in nature and are small particles with porous surface can be formulated as oral, parenteral, topical or inhalational dosage forms. For oral administration, these may be dispersed in a matrix of excipients, diluents, lubricants and anti caking agents which is suitable for the preparation of tablets or capsules and the major benefits of these capsules or tablets are reduction of total dose, retention of dosage form, reduction in toxicity and improving patient compliance by prolonged release (3,4). For parenteral administration, these can be simply mixed with sterile water, saline or other aqueous solutions (4). For topical administration, they can be effectively incorporated into topical hydrogel (5).

Lansoprazole is a proton pump inhibitor commonly used in the treatment of gastric ulcer, gastro oesophageal reflux disease (GERD), duodenal ulcer, ulcers associated with usage of Nonsteroidal anti-inflammatory drug (NSAID) and long term management Zollinger-Ellison syndrome (6). Lansoprazole is primarily metabolized by liver. Hence it is a need to reduce the dose to the hepatic failure patients. But reduction of dose in conventional dosage systems may not show sufficient pharmacological effect (7). Regular usage of lansoprazole causes various adverse effects like abdominal pain, diarrhoea, skin rashes, thrombocytopenia, impotence etc. So, controlled delivery of lansoprazole at optimal concentration may be required (8). Oral route is preferable than other routes with respect to safety, comfort and reliability. Hence controlled delivery of lansoprazole by oral route is ideal. Controlled release of lansoprazole will reduce the frequency of dosing and

dose size and may increase patient convenience (7,8). More over lansoprazole is highly acid labile and represents many formulation challenges to protect it from acidic environment of the stomach (9). So the aim of the present investigation was to formulate enteric coated tablets of lansoprazole nanosponges to protect it from acidic environment and deliver at controlled rate to its absorptive site so that its oral bioavailability can be enhanced (8,9).

## MATERIAL AND METHOD

Lansoprazole was gift sample from Dr. Reddy's Labs limited, Hyderabad. Ethyl Cellulose, Polyvinyl Alcohol and Pluronic F68 were purchased from Qualigens Fine chemicals, New Delhi. All other ingredients used were analytical grade.

### *Methodology*

#### *Preparation of lansoprazole nanosponges (5,10)*

Lansoprazole nanosponges were prepared by different proportions of ethyl cellulose, polyvinyl alcohol and Pluronic F68 by emulsion solvent diffusion technique. The disperse phase consisting of 100 mg lansoprazole and specified quantity of ethylcellulose (Table 1) dissolved in 30 mL of dichloromethane was slowly added to a definite amount of PVA in 100 mL of aqueous continuous phase. The mixture was stirred at 1000 rpm on a magnetic stirrer for two hours. The formed lansoprazole nanosponges were collected by vacuum filtration and dried in an oven at 40°C for 24 h.

#### *Percentage yield*

The lansoprazole nanosponges obtained after drying was weighed. Percentage yield value was calculated as follows (11):  
% yield = Weight of nanosponges × 100 / Total solids weight

#### *Entrapment efficiency (11)*

UV spectrophotometric method was used to estimate entrapment efficiency of lansoprazole nanosponges. A calibration curve was plotted for lansoprazole in methanolic HCl in the range of 3-18 µg/mL (Beer's Lambert's range) at 293 nm. A good linear relationship was observed between the concentration of lansoprazole and its absorbance ( $r^2=0.9993$ ,  $m=0.0469$ ,  $n=3$ ). 100 mg of lansoprazole nanosponges of each batch were selected, powdered in a mortar and placed in 100 mL of methanolic HCl. Lansoprazole was extracted by centrifuging at 1000 rpm for 30 min, filtered and analyzed concentration from calibration curve data after necessary dilution. Percentage entrapment was calculated as follows:

% Entrapment efficiency = Actual drug content in the nanosponge × 100 / Theoretical drug content

#### *Particle size measurement (5,11)*

The average particle size of lansoprazole nanosponges were determined by photon correlation spectroscopy (PCS) using a Nano ZS-90 (Malvern Instruments limited, UK) at a fixed angle at 25°. Sample was diluted 10 times with distilled water and then it was analyzed for particle size.



### Zeta potential (5,11)

The zeta potential was measured for the determination of the movement velocity of the particles in an electric field and the particle charge. In the present work, the nanosponges was diluted 10 times with distilled water and analyzed by Zetasizer using Laser Doppler Micro electrophoresis (Zetasizer nano ZS, Malvern instruments Ltd., UK).

**Table 1. Composition of lansoprazole nanosponges**

|                        | F1  | F2  | F3  | F4   | F5   | F6   |
|------------------------|-----|-----|-----|------|------|------|
| Lansoprazole (mg)      | 100 | 100 | 100 | 100  | 100  | 100  |
| Polyvinyl alcohol (mg) | 600 | 800 | 900 | 1000 | 1100 | 1200 |
| Ethyl cellulose (mg)   | 400 | 600 | 800 | 1000 | 800  | 600  |
| Pluronic F68 (mg)      | 200 | 200 | 200 | 200  | 200  | 200  |
| Dichloromethane (mL)   | 30  | 30  | 30  | 30   | 30   | 30   |
| Distilled water (mL)   | 100 | 100 | 100 | 100  | 100  | 100  |

**Table 2. Formulation of lansoprazole tablets**

| Ingredient                 | Quantity (mg)                            |
|----------------------------|--|
| Nanosponges (F2)           | 35 (equivalent to 30 mg of lansoprazole) |
| Microcrystalline cellulose | 60                                       |
| Magnesium stearate         | 5  |

**Table 3. Evaluation parameters of lansoprazole nanosponges**

|    | Percentage Yield | Entrapment efficiency | Particle size (nm) | Zeta Potential (mV) |
|----|------------------|-----------------------|--------------------|---------------------|
| F1 | 38.35±1.27       | 50.71±0.73            | 190.69             | -4.9                |
| F2 | 59.57±1.09       | 86.93±0.65            | 83.4               | -5.2                |
| F3 | 34.68±1.17       | 79.57±1.01            | 103.26             | -5.6                |
| F4 | 28.24±0.97       | 78.04±1.62            | 114.91             | -6.1                |
| F5 | 33.31±2.1        | 70.31±0.94            | 135.33             | -5.3                |
| F6 | 24.8±1.73        | 69.47±1.2             | 173.27             | -5.2                |

(Mean ± SD, n=3)

**Table 4. Evaluation of lansoprazole tablets**

| Formulation | Weight variation | Thickness (mm) | Hardness (kg/cm <sup>2</sup> ) | Friability (%) | Assay (%)  |
|-------------|------------------|----------------|--------------------------------|----------------|------------|
| F1          | Complies         | 3.18±0.14      | 5.66±0.29                      | 0.886          | 99.93±1.16 |
| F2          | Complies         | 3.23±0.11      | 5.65±0.2                       | 0.752          | 99.47±1.81 |
| F3          | Complies         | 3.09±0.17      | 5.72±0.15                      | 0.892          | 98.18±1.43 |
| F4          | Complies         | 3.21±0.09      | 5.81±0.1                       | 0.836          | 99.97±1.97 |
| F5          | Complies         | 3.27±0.21      | 5.9±0.21                       | 0.811          | 99.01±2.13 |
| F6          | Complies         | 3.15±0.12      | 5.83±0.07                      | 0.798          | 98.43±1.73 |

### Particle shape and morphology (12)

The shape and morphology of nanosponges was examined using Scanning Electron Microscopy (LEO 440I). Sample was deposited on a glass slide, and was kept under vacuum. The samples were coated with a thin gold/palladium layer using a sputter coater unit. The scanning electron microscope was operated at an acceleration voltage of 15 kV.

### Fourier transform infrared spectroscopy studies (12)

The FTIR spectral measurements were taken at ambient temperature using a Perkin Elmer Model 1600 (USA). Samples were dispersed in KBr powder and the pellets were made by applying 5 ton pressure. FTIR spectra were obtained by powder diffuse reflectance on FTIR spectrophotometer.

### Differential scanning calorimetric studies (12)

Differential scanning calorimetry (DSC-60, Shimadzu Corporation, Japan) studies were carried out to check compatibility between drug and polymers. DSC was used after calibration with Indium and lead standards, samples (3-5 mg) were heated (range 50-400 °C, 10 °C/min) in crimped aluminium pans under a nitrogen atmosphere. The enthalpy of fusion and melting point were automatically calculated.

### Porosity

Bulk volume was obtained by pouring the nanosponges in to a grated cylinder and is noted. It is then under gone for 100 tappings and the volume is noted as true volume.

$$\% \text{ Porosity} = (\text{Bulk Volume} - \text{True Volume} / \text{Bulk volume}) \times 100$$

### Determination of residual solvents concentration (13)

Gas chromatography (Shimadzu GC-14B chromatograph, Japan) was used to estimate residual dichloromethane in lansoprazole nanosponges. Dichloromethane content in nanosponges was determined by gas chromatography on an Agilent 7890 Gas Chromatograph, USA fitted with a flame ionization detector. For estimation of residual solvents, 100 mg of nanosponges were dissolved in little amount of DMSO in a 10 mL volumetric flask and volume was made up to 10 mL with DMSO. The solution was filtered through 0.45 µm filter and degassed using sonicator. From the sample, 1 µl was injected into injection port, the chromatogram was recorded and the peak area of solvent was measured. A calibration curve was plotted for dichloromethane in the range of 10-50 ppm. A good linear relationship was observed between the concentration of dichloromethane and its peak area ( $r^2=0.9989$ ). The concentration of residual solvent was calculated from calibration curve data.

### Preparation of lansoprazole tablets

Lansoprazole tablets were prepared by direct compression method. The prescribed quantity of lansoprazole nanosponges, polymers and excipients (Table 2) were mixed homogeneously and the mixture was then compressed into tablets (100 mg) using an 8 mm, biconcave punches on a 'Rimek mini press 16 station rotary compression machine.

## Evaluation of lansoprazole tablets

### Weight variation

The weight variation test was performed according to specifications given in the Indian Pharmacopoeia on 20 tablets. The maximum acceptable limit is  $\pm 7.5\%$  deviation of an individual weight from average weight.

### Thickness

The thickness of 20 randomly selected tablets from each formulation was determined in mm using a vernier caliper (Pico India).

### Hardness

Twenty tablets were randomly selected from each formulation and measured hardness in  $\text{kg}/\text{cm}^2$  using Monsanto type hardness tester.

### Friability

Tablet friability was measured using the Roche Friabilator. Randomly selected twenty pre-weighed tablets were placed in the apparatus and operated for 100 revolutions and then the tablets were reweighed. The friability was determined as the mass loss in percent according to following to Equation

$$F = (WA - WB / WA) \times 100$$

Where F: Friability, WA: Initial weight (gm), WB: Final weight (gm); the acceptable limits of the weight loss should not be more than 1%.

### Assay

Ten tablets were randomly selected from each formulation and crushed to a fine powder in mortar with pestle. Weigh accurately equivalent to 10 mg of lansoprazole from fine powder then transfer in 100 mL volumetric flask, 100 mL of methanolic HCL was added to dissolve and sonicated for 20 minutes. Lansoprazole was extracted by centrifuging at 1000 rpm for 30 min. The samples were filtered, diluted and analyzed UV spectrophotometrically at 239 nm.

### Enteric coating of lansoprazole tablets (14)

Enteric coating of optimized lansoprazole tablets was done to protect the drug in acidic environment. Coating solution was prepared by dissolving 5% w/v of cellulose acetate phthalate and 1.5% w/v of propylene glycol 400 in acetone. Coating solution was applied by dip coating technique using pipette (10 mL) attached to vacuum pump. Vacuum pump produced suction force that allowed tablet to adhere to pipette mouth. This adhered tablet was then partially dipped in coating solution to allow coat formation at one side of tablet. The other side was coated when other side dried.

### In vitro release studies (15,16)

A calibration curve was plotted for lansoprazole in pH 1.2 and pH 6.8 buffers in the range of 3-18  $\mu\text{g}/\text{mL}$  (Beer's Lambert's range) at 306 nm and 285 nm respectively. A good linear relationship was observed between the concentration of lansoprazole and its absorbance in pH 1.2 buffer ( $r^2=0.9987$ ,

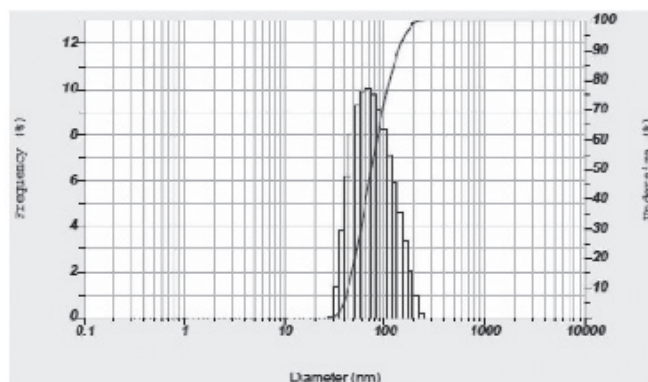


Figure 1. Particle size of lansoprazole nanosponges (F2)

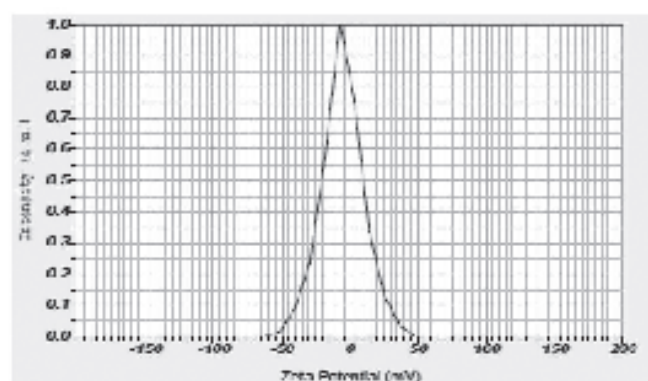


Figure 2. Zeta potential of lansoprazole nanosponges (F2)

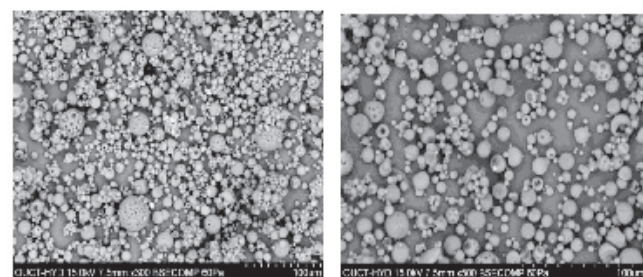


Figure 3. Scanning electron micrograph of lansoprazole nanosponges

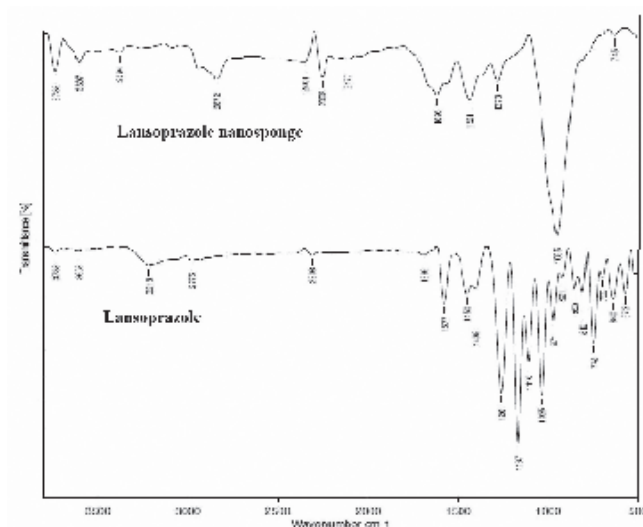


Figure 4. FTIR spectra of lansoprazole and lansoprazole nanosponges

$m=0.0089$ ,  $n=3$ ) and pH 6.8 buffer ( $r^2=0.9979$ ,  $m=0.0189$ ,  $n=3$ ). The dissolution test for optimized lansoprazole nanosponges and coated tablets was carried out according to USP 27 NF 22 by adapting the method B in pH 1.2 and pH 6.8 buffers.

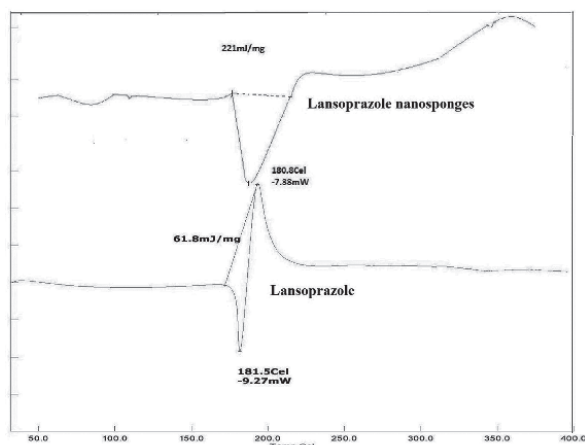


Figure 5. DSC thermograms of lansoprazole and lansoprazole nanosponges

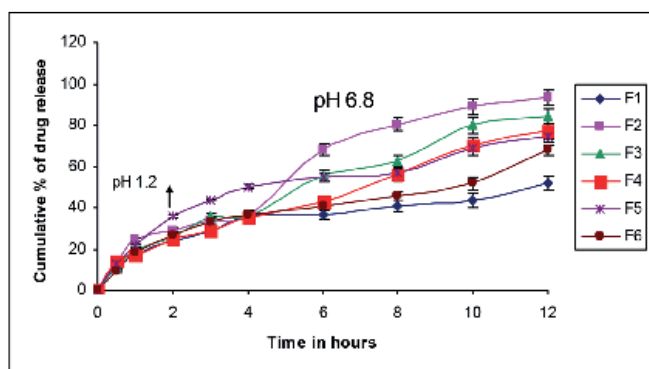


Figure 6. In vitro release profiles of lansoprazole nanosponges

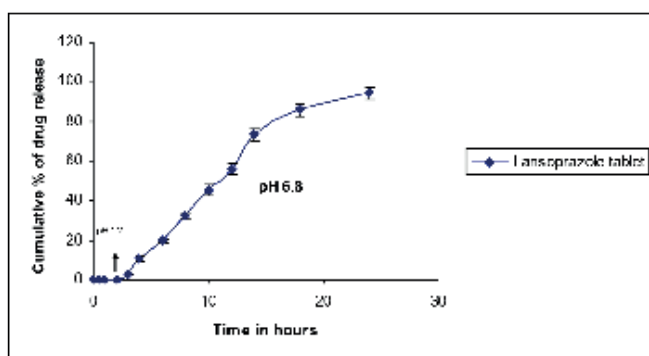


Figure 7. In vitro release profile of lansoprazole enteric coated tablet

Dissolution test was carried out using USP apparatus 1 (Model No TDT-08L, Electrolab, Mumbai) at 100 rpm. To reproduce digestive physiological phases, dissolution medium (900 mL) with different pH environments at  $37\pm0.5^\circ\text{C}$  was used. Six tablets were introduced into the apparatus and the apparatus was run for 2 h in pH 1.2 buffer and 5 mL sample was withdrawn at various time intervals and the same volume of fresh dissolution medium was replaced to maintain sink condition. The samples were filtered, diluted and analyzed UV spectrophotometrically (Shimadzu, Japan) at 306 nm. After 2 h the dissolution medium with the pH of 1.2 was replaced with 6.8 buffer and continued for up to 24 h. Five millilitre samples were withdrawn at regular intervals and the same volume of fresh dissolution medium was replaced to maintain sink condition. The samples were filtered, diluted and analyzed UV spectrophotometrically at 285 nm. Dissolution studies were performed and the mean cumulative percentage of lansoprazole was calculated and plotted against time.

*Evaluation of release kinetics (17-20)*

To investigate the mechanism of lansoprazole release from nanosponges and enteric coated tablets, the release data was analyzed for zero order, first order, Higuchi model and Korsmeyer-Peppas model. The data was presented in the following graphical representation and regression analysis was performed.

$M_t$  versus  $t$  (zero order)

Log cumulative % of drug remained versus  $t$  (first order)

$M_t$  versus square root of  $t$  (Higuchi)

Log  $M_t$  versus log  $t$  (Korsmeyer-Peppas)

$M_t$  is the cumulative % of drug released/permeated at time  $t$ .

Korsmeyer et al (20) derived a simple relationship which described drug release from a polymeric system.

$$M_t/M_\infty = k t^n$$

Where,  $M_t/M_\infty$  is the fraction of drug released at time  $t$ ,  $k$  is the rate constant and  $n$  is the release exponent. Release curve where  $M_t/M_\infty < 0.6$  was used to determine the exponent 'n' value. The  $n$  value was used to characterize different release mechanisms. For example,  $n = 0.45$  for Case I or Fickian diffusion,  $0.45 < n < 0.89$  for anomalous behaviour or non-Fickian transport,  $n=0.89$  for Case II transport, and  $n > 0.89$  for Super Case II transport. Fickian diffusional release occurs by the usual molecular diffusion of the drug due to a chemical potential gradient. Case II relaxational release is the drug transport mechanism associated with stresses

**Table 5. Comparison of correlation coefficient ( $r^2$ ) and rate constant of different kinetic models for F2 and enteric coated tablets**

|                       | Zero order |       | First order |       | Higuchi |       | Korsmeyer - Peppas |         |          |
|-----------------------|------------|-------|-------------|-------|---------|-------|--------------------|---------|----------|
|                       | $r^2$      | $k^0$ | $r^2$       | $k_1$ | $r^2$   | $k_H$ | $r^2$              | n value | $k_{kp}$ |
| F2                    | 0.9617     | 16.84 | 0.9502      | 0.227 | 0.9524  | 71.47 | 0.8528             | 0.581   | 1.404    |
| Enteric Coated tablet | 0.9815     | 10.83 | 0.9476      | 0.121 | 0.9418  | 59.81 | 0.7418             | 0.071   | 3.954    |

and state-transition in hydrophilic glassy polymers, which swell in water or biological fluids. This term also includes polymer disentanglement and erosion. The rate constant 'k', coefficients of correlation ( $r^2$ ) and 'n' of each model were calculated by linear regression analysis.

## RESULTS AND DISCUSSION

Percentage yield value, drug entrapment efficiency, particle size and zeta potential of lansoprazole nanosponges were shown in Table 2.

Percentage yield value of nanosponges was found to be best for F2. Further increasing the concentration of polymer the % yield was found to be decreased due to the sticky nature of the product which can not be filtered. The entrapment efficiency of nanosponges was found to be best for formulation F2. Further increasing the concentration of the polymer, entrapment efficiency was found to be decreased due to low solubility of polymer in aqueous phase (22,23). The size of the nanosponges was found to be in the range 83.4 nm to 190.69 nm (Table 3 and Figure 1). The zeta potential of the nanosponges was found to be in the range -4.9 mV to -5.6 mV (Table 3 and Figure 2). The negative sign indicates the stability of nanosponges.

The SEM images of the lansoprazole nanosponges were shown in Figure 3. SEM analysis revealed that Nanosized spherical particles with numerous pores on surface (lansoprazole nanosponges). The pores are tunneled inwards which may be due to diffusion of dichloromethane from the surface of the nanosponges (5).

The FTIR spectra of pure lansoprazole and lansoprazole nanosponges are shown in Figure 4. FTIR spectra of pure lansoprazole demonstrated the characteristic absorption peaks of 3608 cm for N-H stretching, at 2976 cm for aromatic C-H stretching, 2308 cm for aromatic C-N stretching, 1577 cm for C=C stretching and 1261 for S=O stretching. Lansoprazole nanosponges also showed almost similar absorption peaks indicates good compatibility with polymers (12).

DSC thermogram of pure lansoprazole shows sharp peak at 181.5°C corresponding to its melting point (Figure 5). Lansoprazole nanosponge showed a similar endothermic peak at 180.8°C which confirms no polymer drug interaction (12).

Porosity study is performed to check the extent of nanochannels and nanocavities formed. Porosity of the nanosponges can also be assessed by the use of density of nanosponges. Owing to their porous nature, nanosponges exhibit higher porosity compared to the parent polymer used to fabricate the system. Porosity of the nanosponges was found to be 60% and the bulk volume of the nanosponges was found to be 80 mL and true volume was found to be 32 mL.

The concentration of dichloromethane was found to be 298 ppm. According to Guidelines for residual solvents Q3C (ICH), dichloromethane is class II solvent (solvents to be limited) thus the limits of 600 ppm is acceptable without justification.

Lansoprazole tablets were evaluated for weight variation, thickness, hardness, friability and assay. The results of the evaluation are given in Table 4.

*In vitro* release studies of lansoprazole nanosponges were carried out in triplicate. After 12 h the release was found to be 51.94±3.26, 93.47±3.51, 84.38±3.53, 76.92±3.73, 74.26±2.96 and 67.73±2.49% for the formulations F1, F2, F3, F4, F5 and F6 respectively (Figure 6).

To study the release kinetics of optimized formulation, obtained *in vitro* release data was fitted in various kinetic models such as zero order, first order, Higuchi model and Korsmeyer-Peppas model. The *in vitro* release profile of F2 could be best expressed by zero order kinetic model, as the plot showed highest linearity ( $r^2=0.961$ ). The release exponent (n) value 0.581 (Table 5) indicates that the release from nanosponges followed non fickian release i.e., by swelling and erosion which is always associated with diffusion and dissolution mechanism.

Based on entrapment efficiency and % drug release profiles F2 was selected as optimized formulation and it was formulated in to tablet by direct compression and coated by dipping the tablets in coating solution (5% w/v cellulose acetate phthalate and 1.5% w/v polyethylene glycol 400 in acetone). After 10 min the tablets were removed and air dried.

The enteric coated tablets were subjected to weight variation, hardness, friability, thickness and *in vitro* dissolution studies. The average weight of all tablets was found to be 101.27±2.78. The deviation of all tablets was found to be within the limit. So, all formulations passed the test for uniformity of weight as per official requirements. Thickness of the tablets was found to be 3.37±0.21 mm. Hardness of tablets was found to be 5.7±0.10 kg/cm<sup>2</sup>. Percentage friability of tablets was found to be 0.89 i.e., less than 1%, indicating that the friability was within the prescribed limits. All the tablets showed acceptable properties and complied with the I.P specifications for weight variation, hardness, and friability.

Lansoprazole enteric coated tablet showed no release of drug in acidic medium which is desirable and 94.24±3.02% at the end of 24 h (Figure 7). The *in vitro* release profile of lansoprazole enteric coated tablets could be best expressed by zero order kinetic model, as the plot showed highest linearity ( $r^2=0.981$ ). The release exponent (n) value 0.071 (Table 5) indicates that the release from coated tablets followed fickian release i.e., release always associated with diffusion mechanism (20).

## CONCLUSION

The nanosponges containing lansoprazole exhibited most of the ideal characters required for an oral controlled release dosage forms. The nanosponges of lower particle size 83.4 nm aided with negatively charged surface charge has been achieved. The release profile indicated continuous controlled release up to 12 h. Lansoprazole enteric coated tablet showed no release

of drug in acidic medium which is desirable and controlled release behavior for a period of 24 h. The nanosponge systems have been found to have good potential for prolonged drug release and therefore can be beneficial such as dose reduction, reduced frequency of administration and avoiding related systemic side effects. Hence it can be concluded that the developed oral enteric coated tablet - nanosponges of lansoprazole is considered to be ideal and effective in the management of ulcer and related conditions.

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# Development and Characterization of Voriconazole Loaded *In Situ* Gel Formulations for Ophthalmic Application

## Oküler Uygulama için Vorikonazol Yüklü *In Situ* Jel Formülasyonlarının Geliştirilmesi ve Karakterizasyonu

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### ABSTRACT

The aim of this research was to prepare and evaluate the potential use of *in situ* gel formulations for ocular delivery of voriconazole for the treatment of fungal keratitis. An *in situ* gelling system was used to increase the residence time and thus the bioavailability of voriconazole in ocular mucosa. Temperature triggered *in situ* gel formulations were prepared by cold method using polymers like poloxamer 188, poloxamer 407 and sodium alginate. Finally, concentration of voriconazole in formulations was 0.1% (w/w). These formulations were evaluated for clarity, sol-gel transition temperature, gelling capacity, pH, viscosity and drug content. The gelation temperatures of all the formulations were within the range of 32-34°C. All the formulations exhibited fairly uniform drug content. Furthermore *in vitro* drug release and antifungal activity of these formulations were also evaluated. Drug release study of all the formulations showed sustained release properties. In conclusion, voriconazole loaded *in situ* gels could be offered as a promising strategy for ocular drug delivery for the treatment of fungal keratitis.

**Key words:** Voriconazole, *In situ* gel, Ocular drug delivery, Characterization, Microbiological study

### ÖZ

Bu araştırmanın amacı fungal keratit tedavisi için vorikonazolün göze hedeflendirilmiş potansiyel kullanımı olan *in situ* jel formülasyonlarını hazırlamak ve değerlendirmektir. *In situ* jelleştirici sistemi vorikonazolün oküler mukozada kalış süresini dolayısıyla biyoyararlanımını arttırmak için kullanılmıştır. Sıcaklıktan etkilenen *in situ* jel formülasyonları poloksamer 188, poloksamer 407 ve sodyum alginat gibi polimerler kullanılarak soğuk yöntemle hazırlanmıştır. Son olarak, formülasyonların içindeki vorikonazol konsantrasyonu %0.1 (a/a)dir. Bu formülasyonlar berraklık, sol-jel geçiş sıcaklığı, jelleşme kapasitesi, pH, viskozite ve ilaç içeriği açısından değerlendirilmiştir. Bütün formülasyonların jelleşme sıcaklığı 32-34°C arasında değişmektedir. Bütün formülasyonlar oldukça uygun ilaç içeriğini göstermiştir. Ayrıca bu formülasyonların *in vitro* ilaç salımı ve antifungal aktivitesi de değerlendirilmiştir. Bütün formülasyonların ilaç salım çalışması sürekli salım özelliği göstermiştir. Sonuç olarak, vorikonazol yüklü *in situ* jeller fungal keratit tedavisinde göze ilaç hedeflendirilmesi için gelecek vadeden bir strateji olarak sunulabilir.

**Anahtar kelimeler:** Vorikonazol, *In situ* jel, Oküler ilaç dağılımı, Karakterizasyon, Mikrobiyolojik çalışma

### INTRODUCTION

The eye is unique in terms of its anatomical and physiological nature and defence mechanisms, which make the targeting of drugs to eye tissues one of the greatest challenges in drug delivery (1,2). One of the major limitations faced in ophthalmic delivery is the attainment and retention of optimum drug concentration at the site of action within the eye (3,4,5). Poor bioavailability of drugs from conventional ocular dosage forms is mainly due to tear production, nasolacrimal drainage and transient residence time, drainage of the instilled solution, tear turnover and limited corneal area

(6). Various ophthalmic vehicles such as inserts, ointments, suspensions and aqueous gels, have been developed in order to lengthen the residence time of instilled dose and enhance the ophthalmic bioavailability. These ocular drug delivery systems, however, have not been used extensively because of some drawbacks such as blurred vision from ointments or lack of patient compliance are the main reasons that they have not universally accepted (6,8,9). The effective dose administered may be altered by increasing the retention time of medication into the eye by using *in situ* gel forming systems (10). *In situ* gel forming systems are liquid aqueous solutions

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before administration but turn to gel under physiological conditions. There are several possible mechanisms that lead to *in situ* gel formation, such as pH change, ionic cross-linkage, and temperature modulation (11,12).

Several *in situ* gel forming systems have been developed to prolong the precorneal residence time of a drug and improve ocular bioavailability. Polymers are employed in such delivery systems to carry various drugs and they may demonstrate a transition from sol(liquid) to gel state once instilled in the cul-de-sac of the eye (13). *In situ* gel forming formulations current a novel idea of deliver drugs to patients as a liquid dosage form, yet achieve sustained release of drug for the desired period. Different delivery systems based on polymers have been developed, which are able to increase the residence time of the formulation at absorption site of drugs. In recent years, there has been an increasing interest in water-soluble polymers are highly advantageous compared with other polymers because, in contrast to very strong gels, they can be easily applied in liquid form to the site of drug absorption. At the site of drug absorption, they swell to form a strong gel that is capable of prolonging the residence time of drug (14).

Keratitis, a disease of the cornea, results from direct infection with viruses, bacteria, fungi, yeast, and amoebae or from immune-related complications (4). Fungal keratitis is a leading cause of serious ocular morbidity and blindness. It is worldwide in distribution, but is more common in the tropics and subtropical regions. In fungal keratitis, early diagnosis and antifungal therapy is necessary in preventing further

complications such as hypopyon formation, endophthalmitis, or loss of vision (15).

Voriconazole (VCZ),  $C_{16}H_{14}F_3N_5O$ , a second generation antifungal agent possesses phenomenal characteristics like broad-spectrum activity, activity against resistant fungal species, and acceptable tolerability. Almost 100% *in vitro* susceptibility was observed against various fungal isolates associated with keratitis and endophthalmitis. Moreover, studies suggested an excellent efficacy of voriconazole against several ocular mycoses following topical administration (16).

In this study, a new ocular drug delivery system based on the dispersion of voriconazole loaded *in situ* gels coating into sodium alginate was proposed. The characterization properties of the system were investigated, including clarity, gelling capacity, pH, viscosity and drug content. *In vitro* drug release and antifungal activity of these formulations were also evaluated.

## EXPERIMENTAL

### Materials

VCZ and sodium alginate were purchased from Sigma, Germany. Poloxamer 407 and poloxamer 188 were kind gift from BASF, Turkey. Distilled water was used throughout the study. High pressure liquid chromatography (HPLC) grade acetonitrile (Sigma, Germany) was used for HPLC studies. Roswell Park Memorial Institute (RPMI 1640) medium was purchased from Sigma-Aldrich, Germany (R65504). All the other chemicals and solvents were of analytical or HPLC grade. Dialysis membrane (Spectro/por Dialysis Membrane, Spectra/por 4, diameter 16 mm, molecular weight of 12-14 kDa) was purchased from Spectrum.

### Preparation of *in situ* gel formulations

Poloxamer analogs were used as the gelling agents, and the *in situ* gels were prepared by using a cold method (17). The polymeric solutions were prepared by dispersing required quantity of Poloxamer 407 and Poloxamer 188 in cold water (5 °C) using a magnetic stirrer until the poloxamer completely dissolve (approximately 2 hours). The dispersion was kept in a refrigerator for 48 hours to get clear solution.

### Determination of sol-gel temperature (*Tsol-gel*)

20 g of cold sample solution were put into a beaker and placed in a temperature-controlled stirrer. A thermometer was immersed into the sample solution for constant monitoring. The solution was heated at the rate of 2 °C/min with the continuous with stirring at 200rpm. The temperature at which the magnetic bar stopped moving due to gelation was reported as the gelation temperature. The maximum limit for gelation was checked up to 60 °C. Optimum poloxamer ratios were determined and selected with sol-gel temperature as 32-34 °C which is the eye surface temperature (18). The experiments were repeated four times.

**Table 1. Poloxamers concentrations, gelling temperature and pH of the formulations**

| Codes | Poloxamer 407 (%) | Poloxamer 188 (%) | Gelling temperature (°C) | pH   |
|-------|-------------------|-------------------|--------------------------|------|
| F1    | 15                | 10                | 42-43                    | 7.15 |
| F2    | 15                | 15                | 39-40                    | 7.31 |
| F3    | 15                | 20                | 35-36                    | 7.44 |
| F4    | 15                | 23                | 34-35                    | 7.53 |
| F5    | 15                | 25                | 30-32                    | 7.63 |
| F6    | 18                | 15                | 41-42                    | 7.51 |
| F7    | 18                | 18                | 38-39                    | 7.57 |
| F8    | 18                | 20                | 33-34                    | 7.53 |
| F9    | 18                | 22                | 32-33                    | 7.63 |
| F10   | 18                | 25                | 28-29                    | 7.85 |
| F11   | 20                | 5                 | 30-32                    | 7.07 |
| F12   | 20                | 10                | 33-34                    | 7.22 |
| F13   | 20                | 15                | 36-37                    | 7.30 |
| F14   | 20                | 20                | 32-33                    | 7.34 |
| F15   | 20                | 23                | 32-33                    | 7.67 |

#### *Preparation of voriconazole loaded in situ gel formulations*

*In situ* gels were selected according to pH values and gelling temperatures of formulations. F11 (ratio of P407 and P188 were 20% and 5%, respectively) was selected as optimum formulation for preparation an ocular formulation. After detection of the optimum *in situ* gel compositions sodium alginate of different concentrations (0.1%, 0.3%, and 0.5%) and for each formulations same concentration of VCZ were added in poloxamer solutions with continuous stirring until completely dissolved. Benzalkonium chloride (0.02% w/w) was added as a preservative to the solutions. Sufficient amount of sodium chloride (0.9% w/w) was added to the mixture to maintain the isotonicity. The effect of drug and the other compositions of formulations on gel temperature were also evaluated.

#### *Characterization of in situ gels*

The prepared ocular formulations were characterized such as clarity, gelling capacity, pH, viscosity and drug content. In addition gelling temperature of formulations was determined and statistical analysis was performed using t-test. Data were considered statistically significant at  $p < 0.05$ . The experiments were repeated four times.

#### *Clarity of formulations*

The clarity of formulations was determined by visual inspection under black and white background, and it was graded as follows: turbid, +; clear, ++; and very clear (glassy), +++.

#### *Gelling capacity*

The gelling capacity of the prepared formulation was determined by placing a drop of the formulation in a beaker at 32-34 °C and it was visually observed for gelling time. It was graded as follows: +; gel after few minutes dissolves rapidly, ++; immediate gelation remains for few mins, +++; immediate gelation remains for nearly an hour.

#### *Determination of pH*

The pH of the gel was determined using calibrated pH meter (Mettler Toledo, Switzerland). Determinations were carried out four times and an average of these determinations was taken as the pH of the prepared gels.

#### *Determination of viscosity*

The viscosity of the developed formulations was performed with a digital viscometer (Brookfield) equipped with spindle RV2 with 50 rpm at 32±2°C.

#### *Drug content*

1 mL of the developed *in situ* gel formulation was dissolved in 100 mL pH 7.4 simulated tear fluid buffer (NaCl: 0.670g, NaHCO<sub>3</sub>: 0.200g, CaCl<sub>2</sub>.2H<sub>2</sub>O: 0.008 g and distilled water q.s. to 100 g) (19) followed by HPLC estimation of the aliquot to determine drug concentration. The experiments were repeated four times.

#### *HPLC analysis*

The HPLC system consisted of a gradient pump and a UV detector supplied by Agilent 1100. C18 column (150x4.6mm, 5µm) (GL Sciences, Japan) was used. The samples were analyzed at 256 nm with a 1 mL/min flow rate at 25°C. The mobile phase was a mixture of acetonitrile: ultrapure water (50:50). Retention time of drug was 4.098 min. The method was validated partially linearity, limit of detection (LOD) and limit of quantitation (LOQ), precision, accuracy and specificity, selectivity and stability.

#### *In vitro drug release study*

The *in vitro* drug release study was performed using the dialysis bag method (5). *In vitro* release study of *in situ* gel formulations was carried out in simulated tear fluid (pH=7.4) at 50 rpm. The temperature was maintained at 33±1°C to mimic eye surface temperature. 5 g formulation was separated from release media by means of dialysis membrane (Spectra/por, MW of 12-14 kDa) and capped with closures. The membrane was heated 33±1°C for 30 min in bidistilled water before use. 0.5 mL of sample was withdrawn at a predetermined time interval of 1 h to 8 h and the same volume of fresh medium was replaced. The samples were analyzed with HPLC for the drug content. The experiments were repeated three times.

#### *Stability of the in situ gels*

In physical stability studies, VCZ loaded *in situ* gels were stored at 5±1°C in the refrigerator and 25±2°C and 40±2°C for 3 months in the stability cabinets (Nüve, Turkey). After storage for 3 months visual appearance, clarity, pH, gelation time of *in situ* gels and VCZ content were investigated. The experiments were repeated three times.

#### *Microbiological studies*

##### *Sterility studies*

*In situ* gel formulations in the presence or absence of VCZ were prepared at Laminar air flow Cabinet (Haier HR40-IIA2).

To check the sterility of the prepared ocular formulations sterility control testing were performed. Sterility testing of the *in situ* gel formulations with or without VCZ was carried out under aseptic conditions according to the international pharmacopoeia. For anaerobic bacteria Fluid thioglycollate medium was used. For fungi and aerobic bacteria soya-bean casein digest medium was used. 1 mL of formulation solution was added to each medium and incubated at 35°C for bacteria and 25°C for fungi for 14 days.

To check the suitability of the used mediums for the sterility testing promotion test were performed. For growth promotion test of aerobes, anaerobes and fungi, fluid thioglycollate media (using separate portion of media for each microorganism) were inoculated with 100 CFU of *Staphylococcus aureus* ATCC 6538, *Clostridium sporogenes* ATCC 19404 and *Candida albicans* ATCC 10231. Media were incubated at 35°C for 48 h.



### Determination of MIC of VCZ

The broth microdilution test was done in accordance with CLSI guidelines for filamentous fungi (20) and yeasts (21). VCZ was dissolved in dimethyl sulfoxide, final dilutions were made in RPMI 1640 medium buffered to pH 7.0 with 0.165 M MOPS buffer [3-(N-morpholino) propanesulfonic acid] and final concentrations were 500-0.125 µg/mL. Using the spectrophotometric method of inoculum preparation, an inoculum concentration of  $1.5(\pm 1.0) \times 10^3$  cells/mL for yeasts and  $0.4-5 \times 10^4$  spores/mL for moulds, and RPMI 1640 medium buffered with MOPS were used. *C. albicans* ATCC 10231, *C. tropicalis* RSKK 2412, *A. fumigatus* ATCC 204305 and *A. flavus* ATCC 204304 which may be possible causes of fungal keratitis were used to evaluate antifungal activity of developed formulations (22,23). A 0.1 mL inoculum was added to each well of the microdilution trays. The MICs were determined after 48 h of incubation. The plates were shaken before the comparison of growth in wells. With an aid of a reading mirror, growth in each well was compared with the VCZ free growth control well. The MIC endpoints were evaluated for the lowest drug concentration that showed a prominent reduction (50%) of the growth in the control well.

### Disk diffusion testing

Disk diffusion testing was performed according to CLSI standard M44-A2 for yeasts and M51-A for filamentous fungi. Mueller-Hinton agar (Difco) supplemented with 2% glucose and 0.5 µg/mL methylene blue dye (SSI Diagnostica, Hillerød, Denmark) was used. Blank disks that were 12.1 mm in diameter were impregnated with 20 µL of formulations at final concentration of 1 µg/disk and allowed to dry at room temperature. *C. albicans* ATCC 10231 and *C. tropicalis* RSKK 2412, *A. fumigatus* ATCC 204305 and *A. flavus* ATCC 204304 which may be possible causes of fungal keratitis (22,23) were used to evaluate antifungal activity of developed formulations. The mold inocula were prepared at optical densities ranging from 80 to 82% and a suspension with a 0.5 McFarland standard was utilized for yeasts. The plates were incubated at 35°C and inhibition zone (IZs, in millimeters) diameters were read by using a digital ruler at 24 and 48 h. Minor trailing growth in the inhibition zones was ignored.

## RESULTS AND DISCUSSION

### Preparation of *in situ* gel formulations

Preliminary studies were carried out using different concentrations of polymers evaluated for their gelling

**Table 2. Compositions of the *in situ* gels (%)**

| Components (%)        | S    | S1   | S2   | S3   | S4   |
|-----------------------|------|------|------|------|------|
| Poloxamer 407         | 20   | 20   | 20   | 20   | 20   |
| Poloxamer 188         | 5    | 5    | 5    | 5    | 5    |
| Sodium alginate       | -    | -    | 0.1  | 0.3  | 0.5  |
| Sodium chloride       | 0.9  | 0.9  | 0.9  | 0.9  | 0.9  |
| Voriconazole          | -    | 0.1  | 0.1  | 0.1  | 0.1  |
| Benzalkonium chloride | 0.02 | 0.02 | 0.02 | 0.02 | 0.02 |
| Distilled water qs to | 100  | 100  | 100  | 100  | 100  |

temperature in order to identify the compositions suitable for use in the *in situ* gelling system for ocular drug delivery. Temperature sensitive *in situ* gels were successfully prepared by cold method using poloxamer 407, poloxamer 188. Cold method is one of the preferred methods due to providing clear solution for *in situ* gel while hot process causes formation of lumps of polymer as reported and observed in literature (24). Poloxamer 407 (ethylene oxide and propylene oxide blocks) has excellent thermo-sensitive gelling properties, which is of the most interest in optimising drug formulation. Poloxamer 407 formulations led to enhanced solubilization of poorly water-soluble drugs and prolonged release profile for many applications (e.g., ophthalmic, oral, rectal, topical, nasal and injectable preparations) but did not clearly show any relevant advantages when used alone. Combination with other excipients like Poloxamer 188 or mucoadhesive polymers promotes the action of Poloxamer 407 by optimising sol-gel transition temperature or increasing bioadhesive properties (25). For this purpose poloxamer 188 and poloxamer 407 mixture were used to develop *in situ* gels.

To find out optimum gelling temperature ( $32 \pm 2^\circ\text{C}$ ), the poloxamer 407 and poloxamer 188 were mixed various concentrations. Table 1 shows poloxamer concentrations, gelling temperature and pH of the prepared formulations. F11 was selected optimum composition which contains poloxamer 407 (20%) and poloxamer 188 (5%).

After selecting F11 formulation, benzalkonium chloride (0.02%, w/w) was added as a preservative to the solution. Sufficient amount of sodium chloride (0.9% w/w) was added to the mixture to maintain the isotonicity and the formulation was coded as S. 0.1% (w/w) VCZ was loaded and different concentrations of sodium alginate of different concentrations (0.1%, 0.3%, and 0.5%) was added in poloxamer solutions with continuous stirring until completely dissolved. Finally, concentration of VCZ in formulations was 0.1% (w/w). The dispersion was kept in a refrigerator for 48 hours to get clear solution. The components of ocular *in situ* gels in absence or presence of VCZ are shown in Table 2.

### Characterization of *in situ* gels

Characterization of the new drug delivery systems are major issues to be considered in the formulation stage, especially those intended for ocular administration. The physicochemical characterization parameters of *in situ* gels are reported in Table 3.

The clarity of all the formulations was found to be satisfactory, as shown in Table 3. Gelation temperature was changed from  $32.5^\circ\text{C}$  to  $34.3^\circ\text{C}$  with incorporation of 0.1% w/w VCZ in the poloxamer solution, while the addition of the mucoadhesive polymers played a reverse role on gelation temperature. The results showed that incorporation of sodium alginate into *in situ* gel formulation significantly decreased the gelling temperature ( $p < 0.05$ ). When the concentration of

sodium alginate was increased in the *in situ* gels, the gelling temperature was decreased (Table 3). Gelling temperature of the all formulations (S, S1-S4) were found between 32-35°C. This indicates that the formulations can be converting the gel when they installed the eye surface. At 32-34°C, the solutions are converted into gels with high viscosity. The gelling capacity data of prepared formulations presented in Table 3 represent that the formulations all formulations had immediate gelation and exist for an hour.

The pH of an ophthalmic formulation is important for patient compliance. The pH of the prepared formulations ranged between 7.07 and 7.13. The pH of the formulations was appropriate for ocular delivery since they were iso-hydric. This indicated the nonirritancy of the formulation in ocular mucosa. When a formulation is administered to the eye, it stimulates the flow of tears. Tear fluid is capable of quickly diluting and buffering small volumes of added substances, thus eye can tolerate a fairly wide pH range. Ophthalmic solutions may range from 6.5 to 8.5 (26,27).

The formulation should have an optimum viscosity, which will allow its instillation into the eye as a liquid, which will then undergo rapid sol-gel transition due to temperature change. When the *in situ* solutions were installed the 32-34°C surface, the solutions were converted to gel form after few seconds.

All the formulations reflected fairly uniform drug content ensuring adequacy in the method of preparation of the *in situ* gel. Drug content was found to be within the range of 82.68-92.58%. The drug content of the prepared formulations was within acceptable limits and ensured dose uniformity.

#### *In vitro* drug release study

The *in situ* gelling formulations of VCZ, S1-S4, were subjected to *in vitro* release studies, which were carried out using simulated tear fluid of pH 7.4 as release medium. Formulations showed sustained drug release for a period of 8 hours (Figure 1). At the end of the 8 h, *in vitro* VCZ release from S1, S2 and S3 formulations was found as 63%, 60% and 60%, respectively ( $p > 0.05$ ). S4 formulation was showed slower release than the other formulations. This could be the reason of higher concentration of sodium alginate among the developed formulations. Mandal et al. prepared moxifloxacin hydrochloride loaded *in situ* gel using sodium alginate and hydroxy propyl methyl cellulose as polymers. They were found that when sodium alginate and hydroxy propyl methyl cellulose concentration was increased, release rate was decreased (28).

#### Stability

The stability studies were carried out at  $5 \pm 1^\circ\text{C}$ ,  $25 \pm 2^\circ\text{C}$  and  $40 \pm 2^\circ\text{C}$  for 3 month using stability cabinets. The samples were analyzed periodically on every month, and found that there are no changes in visual appearance, clarity, and gelation time. After 3 month pH values of S1, S2 S3 and S4 formulations were found as  $7.12 \pm 0.002$ ,  $7.18 \pm 0.001$ ,  $7.20 \pm 0.001$  and  $7.21 \pm 0.002$ , respectively. In addition 92-98% of initial drug content of formulations was kept its stability for in 3 month.

#### Microbiological studies

The optimized *in situ* gels passed the test for sterility as there was no appearance of turbidity and hence no evidence of microbial growth when incubated for 14 days at  $35^\circ\text{C}$  in case of fluid thioglycolate medium and at  $20-25^\circ\text{C}$  in the case of soya-bean casein digest medium. Furthermore to control the used

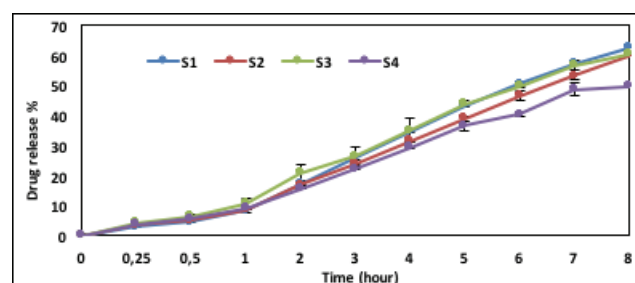


Figure 1. *In vitro* release of voriconazole from *in situ* gels at pH 7.4 simulated tear fluid buffer

| Groups/<br>Zone inhibition<br>ring | <i>A. flavus</i> | <i>A. fumigatus</i> | <i>C. albicans</i> | <i>C. tropicalis</i> |
|------------------------------------|------------------|---------------------|--------------------|----------------------|
| S                                  |                  |                     |                    |                      |
| Voriconazole<br>solution (0.1%)    |                  |                     |                    |                      |
| S1-S2                              |                  |                     |                    |                      |
| S3-S4                              |                  |                     |                    |                      |

Figure 2. Zone inhibition diameters of formulations against *A. flavus*, *A. fumigatus*, *C. albicans* and *C. tropicalis*

**Table 3. Clarity, gelling temperature, pH, drug content, *in vitro* gelation capacity, viscosity, and drug content of *in situ* gels**

| Formulations             | S          | S1         | S2         | S3         | S4         |
|--------------------------|------------|------------|------------|------------|------------|
| Clarity                  | +++        | +++        | +++        | ++         | ++         |
| Gelling temperature (°C) | 32.5±0.03  | 34.3±0.01  | 33.1±0.023 | 32.6±0.012 | 32.4±0.001 |
| Gelling capacity         | +++        | +++        | +++        | +++        | +++        |
| pH                       | 7.07±0.002 | 7.08±0.001 | 7.1±0.002  | 7.11±0.001 | 7.13±0.001 |
| Viscosity (cP)           | 358±3.1    | 362±6.21   | 400±11.32  | 424±23.15  | 440±9.17   |
| Drug content (%)         | -          | 83.14±0.36 | 82.68±1.06 | 84.11±0.7  | 92.58±0.35 |

media for suitability of sterility test growth promotion test were performed and it was found that both microorganisms showed visible growth in all media.

The MIC endpoints were evaluated for the lowest drug concentration that showed a prominent reduction (90% and 50%) of the growth in the control well. MIC<sub>90</sub> values of VCZ against *C. albicans*, *C. tropicalis*, *A. fumigatus* and *A. flavus* were 2, 1, 0.5 and 0.5 µg/mL, respectively and MIC<sub>50</sub> values were 0.25 µg/mL for all of the microorganisms. In accordance with our study (29) also evaluated the MIC<sub>90</sub> results in *Aspergillus sp.* as 1 mg/L. (30) determined the MIC values for both clinical and environmental strains of *A. fumigatus* and *A. flavus* and the MIC ranges were 0.25-2 µg/mL for both of the microorganisms and (31) MIC ranges of *A. fumigatus* and *A. flavus* isolates were 0.25-1 and 0.125-1 µg/mL, respectively. For *Candida* species according to CLSI M27-A3 MIC values ≤ 1 µg/mL are accepted as susceptible; 2 µg/mL are dose dependent susceptible and ≥4 µg/mL are accepted as resistant.

The efficacy of groups was investigated using the inhibition zone diameters in disc diffusion test. Disk diffusion testing was performed according to CLSI standard M44-A2 for yeasts and M51-A for filamentous fungi. *C. albicans*, *C. tropicalis*, *A. fumigatus* and *A. flavus* were used as they are the common organisms causing ocular fungal infections (22,23). Figure 2 shows inhibition zone (IZ) diameters of formulations. The developed VCZ loaded *in situ* gels were found to be more effective on more effective on moulds in comparison with yeasts (*A. flavus* than *C. albicans*) (Table 4). In addition to this, sodium alginate is not effective on the *C. albicans*, *C. tropicalis*, *A. fumigatus* and *A. flavus*, because addition of sodium alginate into the *in situ* gel (S1) did not changed the IZ diameter. While absence of voriconazole (S) in the *in situ* gel (blank *in situ* gel) is not effective on *C. albicans*, *A. fumigatus* and *A. flavus*; a minor inhibition zone was seen in *C. tropicalis* which may be caused by benzalkonium chloride (0.02%, w/w) or other components of the gel. Since the interpretive breakpoints of VCZ for susceptible species is ≥17 mm and for resistant is ≤13 mm, we can say that all of organisms are susceptible for all of the formulations (32).

## CONCLUSION

In the present study, the potential of VCZ loaded *in situ* gels as drug carriers for ocular delivery was investigated. The

present study showed that *in situ* gels of VCZ can successfully be prepared with cold method. The clarity, pH, gellation time and drug content of all formulations was found to be satisfactory. In addition the formulations were found stable for 3 month. Further, all the formulations showed sustained drug release for a period of 8 h, which satisfied to treat ocular disease. The developed *in situ* gels showed anti fungal activity on *C. albicans*, *C. tropicalis*, *A. fumigatus* and *A. flavus*. In conclusion, this study showed that developed *in situ* gel formulations could be alternatively used as ocular delivery of voriconazole. The present study can open up a window for ocular application of *in situ* gels loaded with VCZ, they would be a better alternative to conventional eye drops in the treatment of fungal keratitis of the eye.

## Declaration of Interest

The authors declare no conflict of interest.

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**Table 4. Zone inhibition diameters of formulations**

| Groups/<br>Inhibition Zone<br>(IZ) (cm±SD) | <i>A. flavus</i> | <i>A. fumigatus</i> | <i>C. albicans</i> | <i>C. tropicalis</i> |
|--|------------------|---------------------|--------------------|----------------------|
| S  | 0±0              | 0±0                 | 0±0                | 2.0±0.1              |
| S1   | 5.7±0.5          | 4.8±0.3             | 2.1±0.2            | 3.1±0.1              |
| S2   | 5.2±0.4          | 5.0±0.3             | 2.0±0.1            | 2.8±0.3              |
| S3   | 5.8±0.1          | 4.9±0.2             | 1.9±0.1            | 3.3±0.1              |
| S4   | 5.1±0.3          | 5.0±0.5             | 2.1±0.1            | 3.4±0.2              |

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# Determination of Cytotoxic and Anticandidal Activities of Three *Verbascum L.* Species from Turkey: *V. cheiranthifolium* Boiss. var. *asperulum* (Boiss.) Murb. Monorg., *V. pycnostachyum* Boiss. & Heldr and *V. orgyale* Boiss. & Heldr

Türkiye'den Üç *Verbascum L.* Türünün Sitotoksik ve Antikandidal Aktivitelerinin Belirlenmesi; *Verbascum cheiranthifolium* Boiss. var. *Asperulum* (Boiss.) Murb. Monorg., *Verbascum pycnostachyum* Boiss. & Heldr ve *Verbascum orgyale* Boiss. & Heldr

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## ABSTRACT

Purpose of this study is to determine of cytotoxic and *anticandidal* activities of *Verbascum cheiranthifolium* Boiss. var. *asperulum* (Boiss.) Murb. Monorg., *Verbascum pycnostachyum* Boiss. & Heldr and *Verbascum orgyale* Boiss. & Heldr belonging to *Verbascum* genus growing in Turkey. The cytotoxic effects of methanolic extract of *Verbascum cheiranthifolium* var. *asperulum*, *V. pycnostachyum* and *V. orgyale* species on the cervical (HeLa) and ovarian cancer (Skov-3) cells were investigated using colorimetric assay. The results indicated that methanolic-extract of *V. pycnostachyum* had a promising toxic effect on both cell lines as compared to the other species. Furthermore, this effect was more significant on Skov-3 cells rather than HeLa cells. Anticandidal effects of the methanolic extracts were evaluated in comparison with standard antifungal agents according to Clinical Laboratory Standards Institute (CLSI) reference methods, for the first time here. *V. pycnostachyum* and *V. orgyale* extracts were demonstrated stronger inhibitory effects than the *V. cheiranthifolium* var. *asperulum*. Remarkably, *Candida krusei* was inhibited by *V. pycnostachyum* extract at the concentration of the 62.5 µg/mL.

**Key words:** Scrophulariaceae, *Verbascum*, Cytotoxicity, Anticandidal activities

## ÖZ

Bu çalışmada Türkiye'de yetişen *Verbascum L.* cinsine ait üç türün; *Verbascum cheiranthifolium* Boiss. var. *asperulum* (Boiss.) Murb. Monorg., *V. pycnostachyum* Boiss. & Heldr, ve *V. orgyale* Boiss. & Heldr. türlerinin sitotoksik ve antikandidal aktivitelerinin belirlenmesi amaçlanmıştır. *Verbascum cheiranthifolium* var. *asperulum*, *V. pycnostachyum* and *V. orgyale* türlerinin metanol ekstrelerinin sitotoksik etkileri servikal (HeLa) ve ovaryum kanser (Skov-3) hücrelerinde kolorimetrik metod kullanılarak araştırılmıştır. Elde edilen sonuçlar; *V. pycnostachyum* türünün metanol ekstresinin diğer türlere oranla her iki hücre hattında da umut verici toksik etkiye sahip olduğu gösterilmiştir. Buna ek olarak; bu etki Skov-3 hücrelerinde HeLa hücrelerine kıyasla daha anlamlıdır. Üç türe ait metanol ekstresinin antikandidal etkileri "Klinik Laboratuvar Standartları Enstitüsü" (CLSI)'nin mikrodilüsyon standart protokolleri kullanılarak standart antifungal ajanlarla karşılaştırılmalı şekilde ilk kez bu çalışma ile ortaya konmuştur. *V. pycnostachyum* ve *V. orgyale* ekstreleri *V. cheiranthifolium*'a göre daha kuvvetli inhibitör etkiler göstermiştir. *Verbascum pycnostachyum* ekstresi dikkat çekici olarak *Candida krusei*'yi, 62.5 g/mL konsantrasyonda inhibe etmiştir.

**Anahtar kelimeler:** Scrophulariaceae, *Verbascum*, Sitotoksosite, Antikandidal aktivite

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## INTRODUCTION

*Verbascum* L. (1753: 177) (Scrophulariaceae) includes about 360 species throughout world (1). In Turkey, with the additional 130 hybrids, the genus is represented by 246 species, 6 imperfectly known or doubtful records (2-5). The endemism ratio (80%) of the genus is very high with 196 endemic species (4,5).

In Turkey, the species *V. cheiranthifolium* var. *asperulum*, *V. pycnostachyum* and *V. orgyale* known as "Bozkulak", "Eğirdir siğir kuyruğu" and "Söke siğir kuyruğu" respectively (2,4).

Many plant species among the flora of Turkey play an important role in traditional medicine. There are approximately 9000 plant species, some of them are widely used in folkloric medicine due to their antimicrobial and anticarcinogenic properties, in Turkish flora (6,7). One of the well-known *Verbascum* species is *V. thapsus* L., which has been used for the treatment of several diseases including asthma, spasmodic cough, migraine and earache. Moreover, *V. thapsus*, *V. fruticosum*, *V. undulatum* and *V. georgicum* had anti-malarial and anti-viral effects that were investigated by both *in vitro* and *in vivo* studies (6).

It is reported that leaves and flowers of *Verbascum* species have expectorant, mucolytic and demulcent properties, and they are used to treat respiratory disorders such as bronchitis, dry coughs, tuberculosis, asthma in Anatolia (8,9). *Verbascum* species are also used to treat hemorrhoids, rheumatic pain, superficial fungal infections, wounds and diarrhea. Furthermore these species demonstrate several inhibitory activities against the murine lymphocytic leukemia and influenza viruses A2 and B. Macerated oil prepared from the flowers is used for reducing earache, applied externally for eczema and other types of inflammatory skin disorders (10).

*Verbascum* species have some folkloric usages such as sedative and treatment of dysmenorrhoea and rheumatism. It was also notified the usage for healing wounds in animal care.

Iridoid and neolignan type glycosides, oleanan type terpenes, flavonoids, polysaccharides, saponins, steroids and alkaloids were major compounds isolated from *Verbascum* species (11). In several bioactivity studies on *Verbascum* sp. reported that crude extracts of roots, leaves, flowers and aerial parts have been shown anti-proliferative (12), anti-inflammatory (13), antioxidant (14,15), anti-histaminic, anti-fungal, anti-bacterial, (16), wound healing (17), anti-microbial (18) and anti-cancer effects (19).

In the present study, three species belonging to *Verbascum* genus, were evaluated for their cytotoxic (on cervical and ovarian cancer cell lines) and anticandidal effects for the first time.

## EXPERIMENTAL

### Plant materials

The plant materials were collected from following localities; *Verbascum cheiranthifolium* var. *asperulum* B3 Eskişehir, Bozdağ region, 18.6.2014, 39° 53' 24" K - 030° 33' 16" D, 1267 m, (ESSE:14686); *Verbascum pycnostachyum* C3: Antalya, Korkuteli-Fethiye region, 37° 02' 53" N, 30° 06' 26", 1370 m, 20.06.2007, ESSE 14730 (AKDU 6093) and *Verbascum orgyale* C3:Antalya: Antalya-Geyikbayırı region, 36° 52' 41" N - 30° 26' 37" E, 1008 m, 15.07.2007, (ESSE 14622, AKDU 6064) in Turkey. Voucher specimens are deposited in the Herbarium of the Faculty of Pharmacy (ESSE), Anadolu University in Eskişehir and Herbarium of the Biology Department, Akdeniz University in Antalya, Turkey (AKDU).

### Extraction

Air dried plant materials were macerated with 70% MeOH (MERCK) at 25°C for 24h on orbital shaker. After evaporation and lyophilization steps the dry extract was kept at +4°C until bioactivity studies.

### Cell culture

The human cervical adenocarcinoma cells (HeLa) were maintained in Eagle's Minimum Essential Medium (EMEM) (Sigma-Aldrich, UK) supplemented with 20% Fetal Bovine Serum (FBS) (Gibco, UK), 1% penicillin-streptomycin and 4% sodium bicarbonate as adherent monolayers. The human ovarian adenocarcinoma cells (Skov-3) were cultured in Dulbecco's Modified Eagle Medium (DMEM) (Sigma-Aldrich, UK) supplemented with 10% FBS and 1% penicillin-streptomycin. The cell lines were routinely subcultured using 0.25% trypsin-EDTA solution (Sigma-Aldrich, UK).

Stock solution of extract of *Verbascum* sp. were prepared in sterile ddH<sub>2</sub>O and that was diluted in culture medium to prepare final concentrations of extracts. The cells were incubated with each *Verbascum* sp. (0,1-3 mg/mL) for 24 hours at 37°C (20).

### Cell viability assay

MTT[3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyl-2H-tetrazolium bromide] is a non-radioactive assay and measures the reduction of yellow MTT by mitochondrial succinate dehydrogenase. The reduction of MTT can only occur in metabolically active cells. The assay was performed as mentioned in Mossman. HeLa and Skov-3 cells ( $2 \times 10^4$ ) were seeded in 96-well plates in the presence and absence of different concentrations of *Verbascum* sp. for 24 hours at 37°C in a 5% CO<sub>2</sub>/95% air atmosphere. After incubation time, 20 ml of MTT (5 mg/mL) was added to each well and the cells were incubated for a further 2 hours. The reduction of MTT was measured by ELISA (ELX 808 IU) reader at a wavelength of 540 nm.

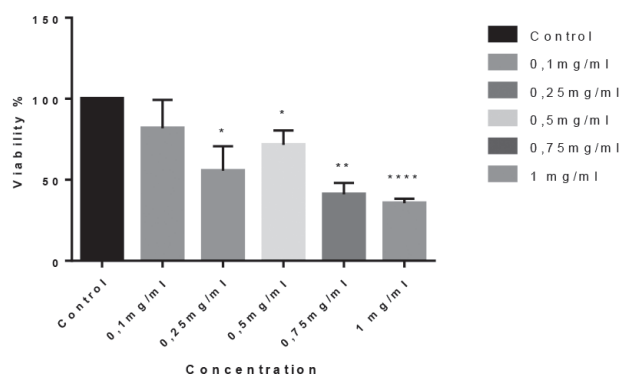
Viability (%)=(Absorbance of the treated cells) / (Absorbance of the control wells) ×100. Each concentrations was tested in two different experiments run in triplicate.

**Anticandidal activity**

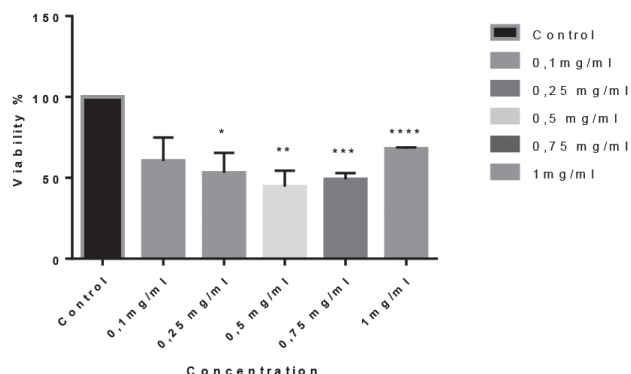
Anticandidal activities of the methanolic extracts were evaluated by partly modified reference method of Clinical and Laboratory Standards Institute (CLSI) M27-A2 (21).

*Candida albicans* ATCC 90028, *C. utilis* NRRL Y-900, *C. glabrata* ATCC 66032, *C. tropicalis* ATCC 750, *C. parapsilosis* ATCC 22019 and *C. krusei* ATCC 6258 were used as pathogenic test microorganisms. Stock cultures stored in

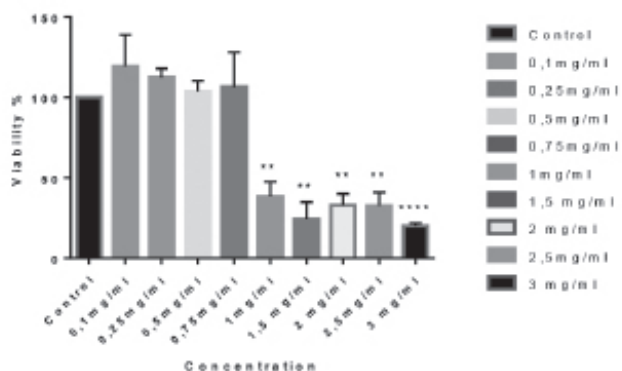
50% glycerol at -85°C, were inoculated in Mueller Hinton Agar (Acumedia) plates and incubated at 37°C for 24 h for checking purity and viability. After incubation, selected colonies were suspended in 0.85% NaCl solution and adjusted to McFarland No: 0.5. Serial dilutions of the extracts were prepared in range of 4000 to 7 µg/mL. After incubation at 37°C for 24h, MIC values was determined by visual reading of wells without growing. Amphotericin B (Sigma) and Ketoconazole (Sigma) were used as standard antifungal agents.



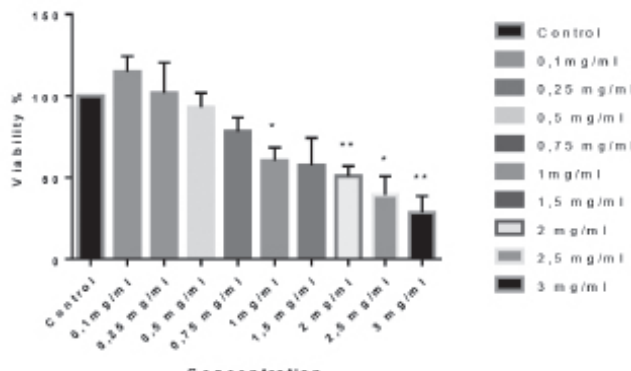
*Verbasicum pycnostochyllum* (A)



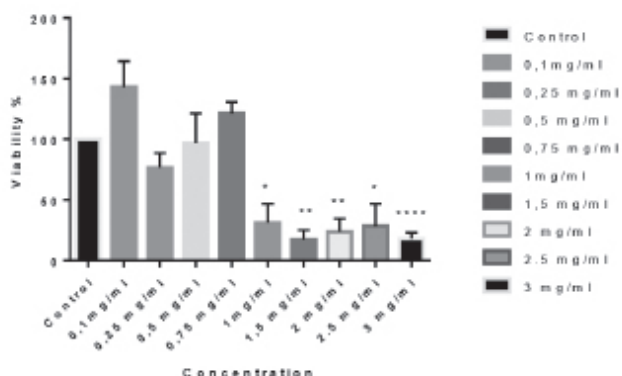
*Verbasicum pycnostochyllum* (A)



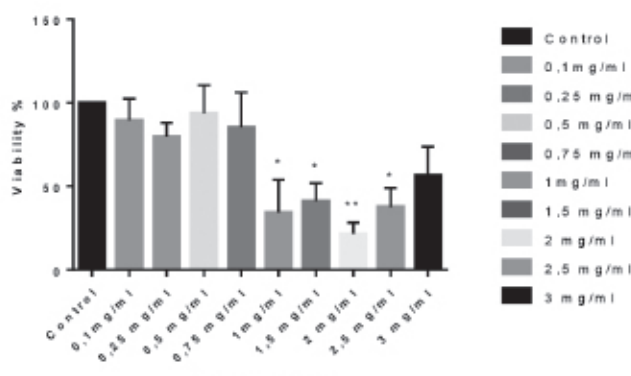
*Verbasicum cheiranthifolium* var. *asperulum* (B)



*Verbasicum cheiranthifolium* var. *asperulum* (B)



*Verbasicum orgyale* (C)



*Verbasicum orgyale* (C)

**Figure 1.** Treatment of either *V. pycnostachyllum*, *V. cheiranthifolium* var. *asperulum* or *V. orgyale* extracts with HeLa cells decreased the cell viability in a dose-dependent manner. Bars indicate mean ± standard deviation. All comparisons were made relative to untreated control cells. The significant differences were indicated as p<0.05 using one-way ANOVA. The graphic was created by using GraphPad Prism 6 software. [\*p<0.1; \*\*p<0.01; \*\*\*\*p<0.001].

**Figure 2.** The percentage of cell viability after treating Skov-3 cells with either *V. pycnostachyllum*, *V. cheiranthifolium* var. *asperulum* or *V. orgyale* methanolic-extract. Bars indicate mean ± standard deviation. All comparisons were made relative to untreated control cells. The significant differences were indicated as p<0.05 using one-way ANOVA. The graphic was created by using GraphPad Prism 6 software. [\*p<0.1; \*\*p<0.01; \*\*\*\*p<0.001].

**Table 1. Anticandidal Activity ( $\mu\text{g/mL}$ , MIC)**

|  | <i>C. albicans</i> | <i>C. tropicalis</i> | <i>C. parapsilosis</i> | <i>C. utilis</i> | <i>C. glabrata</i> | <i>C. krusei</i> |
|--|--------------------|----------------------|------------------------|------------------|--------------------|------------------|
| <i>V. cherianthifolium</i> var. <i>asperulum</i> | 1000               | 250                  | >4000                  | 2000             | >4000              | 250              |
| <i>V. oryale</i>                                 | 125                | 250                  | 125                    | 250              | >4000              | 125              |
| <i>V. pycnostachyum</i>                          | 1000               | 250                  | 125                    | 250              | >4000              | <b>62.5</b>      |
| Amphotericin B                                   | 0.031              | 1.0                  | 0.25                   | 0.5              | 2.0                | 1.0              |
| Ketoconazole                                     | 0.008              | 0.031                | 0.031                  | 0.25             | 0.25               | 0.25             |

## RESULTS AND DISCUSSION

### Cytotoxicity results

The effects of *V. pycnostachyum*, *V. cherianthifolium* var. *asperulum* and *V. oryale* methanol-extracts were assessed on HeLa (Figure 1) and Skov-3 (Figure 2) cells after 24 hours incubation with each extract using the MTT assay. The results obtained here indicated that all *Verbascum* sp. reduced the cell viability of both HeLa and Skov-3 cells in a dose-dependent manner. Particularly, the cell viability of both cell lines was significantly declined after treatment of *V. pycnostachyum* extract as compared to other *Verbascum* sp. that cytotoxic effect was observed at lower concentration (0.5 mg/mL - 44.62% cell viability) on Skov-3 cells rather than HeLa cells (0.5 mg/mL - 71.54% cell viability).

*V. oryale* methanolic-extract was shown a similar effect on both cell lines; HeLa (1 mg/mL - 30.96% cell viability) (Figure 1C) and Skov-3 (1 mg/mL - 34.22% cell viability) (Figure 2C). On the other hand, a dramatic decrease in cell viability for HeLa was observed after incubation of 0.93 mg/mL *V. cherianthifolium* var. *asperulum* methanol-extract (Figure 1B) as compared to the cell viability rate of Skov-3 cells treated with 2.01 mg/mL extract (Figure 2B).

The studies about the isolation of bioactive compounds have been reported that flavonoids, saponins, phenylpropanoid (12) and the phenylethanoid glycosides (22) were isolated although the type of bioactive compounds varies depending on the various *Verbascum* sp. Specifically, the isolation works on methanolic-extract and structure elucidation studies of *V. pycnostachyum* were shown that it contained iridoids-glycosides, aukubin, ajugol, ajugosid, harpagoside, phenylethanoid glycoside and verbascoside (10). It has been reported that verbascoside has a hydrophilic character (19) and saponins (23) to possess anti-cancer and antimicrobial activity.

In this study, particularly *V. pycnostachyum* species having a significant cytotoxic effect on Skov-3 cells that might be caused by the compounds such as verbascoside. However, in order to explain the relationship between activity-structure, it is necessary to determine the content of bioactive compounds of *V. pycnostachyum* methanolic-extract.

### Anticandidal activity results

Anticandidal activities of the methanolic extracts of *V. cherianthifolium* var. *asperulum*, *V. oryale* and *V.*

*pycnostachyum* were evaluated by using CLSI M27-A2 reference method. Tested *Candida* species were moderately inhibited by the extracts between the concentrations of the 62,5-4000  $\mu\text{g/mL}$  (minimal inhibitory concentration). Remarkably, *V. pycnostachyum* showed strong effects on *Candida krusei* having a MIC value of 62,5  $\mu\text{g/mL}$ . *V. oryale* and *V. pycnostachyum* demonstrated better effects than *Verbascum cherianthifolium* var. *asperulum* against all tested *Candida* strains. All extracts were assumed to have the MIC values outside of the tested range against *Candida glabrata* ATCC 66032 (Table 1). In the previous study on *Verbascum* species, extract of the *V. sinuatum* L. showed anticandidal effect at the concentration of 32  $\mu\text{g/mL}$  against *C. albicans* (25). In another study methanolic extract of the *V. georgicum* which have antimicrobial constituents reported as a novel antimicrobial raw material (6). According to a scientific review on bioactivities of *Verbascum* species, methanol and ethanol extracts showed strong inhibitory effects on *Candida albicans* and Gram (+) bacteria strains due to the their saponin content (26).

Today, especially in immunocompromised people, *Candida* infections causes major health problems. There are few available systemic antifungal drugs, additionally the rate of drug resistance is increasing dramatically to available drugs. The search for new natural antifungal agents against pathogenic *Candida* species is extremely important (24).

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# The Importance of *Asphodeline* Species on Enzyme Inhibition: Anti-Elastase, Anti-Hyaluronidase and Anti-Collagenase Potential

## *Asphodeline* Türlerinin Enzim İnhibisyonundaki Önemi: Anti-Elastaz, Anti-Hyalüronidaz ve Anti-Kollajenaz Potansiyeli

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### ABSTRACT

*Asphodeline* species are widespread in the inner Anatolia region in Turkey and used for the treatment of skin disorders, earaches and haemorrhoids. The aim of the present study is to investigate *in vitro* inhibitory effects of the extracts prepared from the stems, seeds, leaves and roots of *Asphodeline brevicaulis* subsp. *brevicaulis* var. *brevicaulis*, *A. baytopae* and *A. cilicica* on hyaluronidase, collagenase and elastase enzymes. Hyaluronidase, collagenase and elastase inhibitory effects of the extracts were performed by using *in vitro* enzyme inhibitory assays based on spectrophotometric evaluation. The methanol extract of the roots of *A. cilicica* displayed the highest hyaluronidase, collagenase and elastase inhibitory activities. On the other hand, the acetone extract of the roots of *A. cilicica*, the acetone extract of the leaves of *A. brevicaulis* subsp. *brevicaulis* var. *brevicaulis*, the acetone and the methanol extracts of seeds of *A. baytopae*, acetone and the methanol extract of the roots of *A. baytopae*, the aqueous extract of the stems of *A. cilicica*, the acetone and the methanol extracts of the leaves of *A. cilicica*, the aqueous extract of the roots of *A. cilicica* possessed significant collagenase, hyaluronidase and elastase inhibitory activities. In the present study, extracts of *Asphodeline* species significantly inhibited collagenase, elastase and hyaluronidase enzymes, suggesting their utilization for the treatment of wounds, cancer, cardiovascular diseases, inflammation, bone destruction and fibrosis, as well as skin aging.

**Key words:** *Asphodeline*, Xanthorrhoeaceae, Collagenase, Elastase, Hyaluronidase

### ÖZ

*Asphodeline* türleri Türkiye’de İç Anadolu Bölgesi’nde geniş bir yayılış göstermektedir ve halk arasında deri hastalıkları, kulak ağrıları ve hemoroit gibi rahatsızlıklara karşı kullanılmaktadır. Bu çalışmanın amacı, *Asphodeline brevicaulis* subsp. *brevicaulis* var. *brevicaulis*, *A. baytopae* ve *A. cilicica* bitkilerinin gövde, tohum, yaprak ve köklerinden hazırlanan ekstrelerin hyalüronidaz, kollajenaz ve elastaz enzimleri üzerindeki *in vitro* inhibitör etkinliklerini araştırmaktır. Bu ekstrelerin inhibitör etkileri spektrofotometrik yöntemlerle değerlendirilmiştir. *A. cilicica* köklerinden hazırlanan metanol ekstresinin en yüksek hyalüronidaz, kollajenaz ve elastaz inhibitör aktivitelere sahip olduğu belirlenmiştir. Diğer yandan, *A. cilicica* köklerinden hazırlanan aseton ekstresinin, *A. brevicaulis* subsp. *brevicaulis* var. *brevicaulis* yapraklarından hazırlanan aseton ekstresinin, *A. baytopae* köklerinden ve tohumlarından hazırlanan aseton ve metanol ekstrelerinin, *A. cilicica* gövdesinden hazırlanan sulu ekstrelerinin, *A. cilicica* yapraklarından hazırlanan aseton ve metanol ekstrelerinin, *A. cilicica* köklerinden hazırlanan sulu ekstrelerinin önemli ölçüde kollajenaz, elastaz ve hyalüronidaz etki gösterdiği belirlenmiştir. Sonuç olarak bu çalışmada *Asphodeline* türlerinden hazırlanan ekstrelerin kollajenaz, elastaz ve hyalüronidaz enzimlerini önemli ölçüde inhibe ettiği ortaya çıkmıştır. Bu nedenle bu sonuçlar bitkinin yara, kanser, kardiyovasküler rahatsızlıklar, enflamasyon, kemik yıkımı ve fibrozis gibi hastalıkların tedavisinde kullanılabileceğini destekler niteliktedir.

**Anahtar kelimeler:** *Asphodeline*, Xanthorrhoeaceae, Kollajenaz, Elastaz, Hyalüronidaz

### INTRODUCTION

*Asphodeline* Rchb. genus (Xanthorrhoeaceae), represented by 14 species in the world, is growing wild in south-west Asia and Middle-Eastern countries as well as in Mediterranean region. In Turkey, *Asphodeline* genus comprises of 20 taxa, 12 of which are endemic to Turkey showing that Turkey is

one of the gene centres of this genus (1-3). In rural areas in Turkey, several *Asphodeline* species are known in the name of “çiriş” and are widespread in inner Anatolia region. Especially *Asphodeline cilicica* E. Tuzlacı, *A. damascena* (Boiss.) Baker, *A. globifera* J. Gay Ex Baker, *A. lutea* (L.) Rchb. and *A. taurica* (Pallas) Kunth are consumed as food in salads (4). This genus

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was recorded to have economical and nutritional importance with high levels of essential amino acids and polyphenols (3-7). Besides their nutritional properties, *Asphodeline* species also possess medicinal features and are employed in medical practices in traditional medicine. For instance, *A. globifera* has been used for alleviating the symptoms of haemorrhoids; *A. damascena* and *A. cilicica* were recorded to be utilized for the treatment of earaches (8); and *A. lutea* has been used for the treatment of skin diseases (9). Due to their mentioned use above, many phytochemical researches have been conducted on *Asphodeline* species revealing the presence of secondary metabolites such as anthraquinones, sesquiterpenes, flavonoids and naphthalene type compounds (6,10-13). Enzymes have very important roles in the pathogenesis of several diseases including cancer, inflammation, Alzheimer's and Parkinson's diseases, familial hypercholesterolemia, myasthenia gravis etc. (14-17). Due to the important roles of these enzymes on several diseases, novel drugs that display inductive or inhibitory effects should be developed.

In the present research, *in vitro* inhibitory effects of the extracts obtained from the stems, seeds, leaves and roots of *A. brevicaulis* (Bertol.) J. Gay ex Baker subsp. *brevicaulis* (Bertol.) J. Gay ex Baker var. *brevicaulis* (Bertol.) J. Gay ex Baker, *A. baytopae* E. Tuzlacı and *A. cilicica* on hyaluronidase, collagenase and elastase enzymes, which are the major enzymes responsible for dehydration of the skin, were investigated.

## EXPERIMENTAL

### Plant materials

*Asphodeline* species were collected at the end of flowering stage (May-July) and information regarding the collection

sites of the plants and herbarium numbers were presented in Table 1. Taxonomic identification of the plant materials were confirmed by the senior taxonomist Dr. Murad Aydın Sanda, from the Department of Biology, Selçuk University. The voucher specimens were deposited at the KNYA Herbarium of Department of Biology, Selçuk University, Konya, Turkey.

### Preparation of the plant extracts

The plant materials (stem, root, seed and leaf) were dried at the room temperature. The dried parts were ground to a fine powder using a laboratory mill. For each of the powdered parts (10 g) were separately extracted with 250 mL acetone and methanol in a Soxhlet apparatus for 6-8 h. The residue was extracted with 250 mL hot distilled water for 30 min and the extracts were filtered concentrated under vacuum at 40°C by using a rotary evaporator. The aqueous extracts were lyophilized (-80°C, 48 h). Yields of the extracts were given in Table 1.

### In vitro enzyme inhibitory assays

#### Hyaluronidase inhibition assay

Hyaluronidase inhibition assay was performed according to the methods described by Lee & Choi (1999) and Sahasrabudhe & Deodhar (2010) with some modifications (18-20).

An amount of 50 µL an aliquot of bovine hyaluronidase (7900 units/mL) was dissolved in 0.1 M acetate buffer (pH 3.6). This mixture then was mixed with 50 µL of different concentrations of the extracts prepared in 5% DMSO. An aliquot of 50 µL of 5% DMSO was added instead of the extracts in the control group. After incubation at 37°C for 20 min, 50 µL of calcium chloride (12.5 mM) was added to the mixture and reincubated for another 20 minutes at 37°C. 250 µL sodium hyaluronate (1.2 mg/mL) was added and incubated for 40 minutes at 37°C.

**Table 1. The collection cites and herbarium numbers of *Asphodeline* species; parts used for the extraction, extract types and percentage yields of the extracts**

| Plant name  | Collection site   | Herbarium no | Extract type | Parts used     | Yield (w/w, %)             |
|---|---|--------------|--------------|----------------|----------------------------|
| <i>A. brevicaulis</i> subsp. <i>brevicaulis</i> var. <i>brevicaulis</i> | Mersin, Arslanköy, between Arslanköy and Yeniköy, 37° 00' 20.9"N, 34° 29' 24.6 E, alt. 1077 m | KNYA-GZ1004  | Acetone      |                | 1.73/ 1.28/ 2.71/ 3.91     |
|   |   |              | Methanol     | Sm/ R/ Lf / Sd | 12.64/ 16.04/ 11.21/ 10.27 |
|   |   |              | Aqueous      |                | 18.05/ 14.1/ 20.35/ 20.41  |
| <i>A. baytopae</i>  | Mersin, Gulnar, between Gulnar and Aydıncık, 36° 16' 07" N, 33° 22' 11" E, alt. 751 m         | KNYA-GZ1003  | Acetone      |                | 0.88/ 4.68/ 2.66/ 2.71     |
|   |   |              | Methanol     | Sm/ R/ Lf / Sd | 4.16/ 39.06/ 22.38/ 6.42   |
|   |   |              | Aqueous      |                | 4.32/ 27.56/ 25.91/ 10.00  |
| <i>A. cilicica</i>  | Adana, between Catalan and Aladag, 37° 27'37" N, 35° 20' 12" E, alt. 1080 m                   | KNYA-GZ1005  | Acetone      |                | 2.88/ 2.42/ 4.07/ 3.52     |
|   |   |              | Methanol     | Sm/ R/ Lf / Sd | 8.85/ 7.39/ 13.32/ 14.47   |
|   |   |              | Aqueous      |                | 10.78/ 9.35/ 13.04/ 24.49  |

Abbreviations: alt.: altitude; Sm: Stem, R: Root, L: Leaf, Sd: Seed

**Table 2. Collagenase and elastase inhibitory activity of the extracts of *Asphodeline* species**

| Material  | Parts used | Extract type | Concentration (µg/mL) | Collagenase inhibition (%) ± S.E.M. | Elastase inhibition (%) ± S.E.M. |
|---|------------|--------------|-----------------------|-------------------------------------|----------------------------------|
| <i>A. brevicaulis</i> subsp. <i>brevicaulis</i> var. <i>brevicaulis</i> | Stem       | Acetone      | 100                   | 15.79±1.83                          | 19.43±1.95                       |
|   |            | Methanol     | 100                   | 14.66±1.48                          | 12.63±1.59                       |
|   |            | Aqueous      | 100                   | 15.99±1.64                          | 19.47±1.93                       |
|   | Seed       | Acetone      | 100                   | 13.44±2.14                          | 8.12±1.66                        |
|   |            | Methanol     | 100                   | 17.06±1.64                          | 16.42±1.82                       |
|   |            | Aqueous      | 100                   | 14.73±2.14                          | 10.17±2.34                       |
|   | Leaf       | Acetone      | 100                   | <b>31.38±1.14**</b>                 | <b>39.39±1.61**</b>              |
|   |            | Methanol     | 100                   | 10.93±2.26                          | 8.37±2.11                        |
|   |            | Aqueous      | 100                   | 8.66±1.22                           | 6.30±1.82                        |
|   | Root       | Acetone      | 100                   | 20.34±2.42                          | 22.49±2.22                       |
|   |            | Methanol     | 100                   | 23.64±2.31                          | 28.78±1.98                       |
|   |            | Aqueous      | 100                   | 14.38±1.91                          | 15.85±1.80                       |
| <i>A. baytopae</i>  | Stem       | Acetone      | 100                   | 19.31±2.64                          | 25.46±1.64                       |
|   |            | Methanol     | 100                   | 13.43±2.54                          | 20.10±1.76                       |
|   |            | Aqueous      | 100                   | 7.56±1.48                           | 9.29±1.84                        |
|   | Seed       | Acetone      | 100                   | <b>30.55±1.28**</b>                 | <b>40.22±1.46**</b>              |
|   |            | Methanol     | 100                   | <b>37.22±1.40***</b>                | <b>42.32±1.76**</b>              |
|   |            | Aqueous      | 100                   | 9.53±1.82                           | 11.83±1.80                       |
|   | Leaf       | Acetone      | 100                   | 16.33±1.77                          | 18.29±1.85                       |
|   |            | Methanol     | 100                   | 18.89±1.82                          | 21.16±2.34                       |
|   |            | Aqueous      | 100                   | 8.46±2.54                           | 9.68±1.49                        |
|   | Root       | Acetone      | 100                   | <b>31.70±1.56**</b>                 | <b>41.51±1.40**</b>              |
|   |            | Methanol     | 100                   | <b>29.85±2.49*</b>                  | <b>31.88±1.92*</b>               |
|   |            | Aqueous      | 100                   | 10.36±1.99                          | 14.87±1.55                       |
| <i>A. cilicica</i>  | Stem       | Acetone      | 100                   | 12.26±2.83                          | 14.64±1.72                       |
|   |            | Methanol     | 100                   | 9.41±1.34                           | 7.84±1.90                        |
|   |            | Aqueous      | 100                   | <b>35.38±1.43***</b>                | <b>45.97±1.44**</b>              |
|   | Seed       | Acetone      | 100                   | 16.66±2.04                          | 20.66±1.89                       |
|   |            | Methanol     | 100                   | 15.36±1.92                          | 18.54±2.34                       |
|   |            | Aqueous      | 100                   | 8.52±1.78                           | 9.44±1.62                        |
|   | Leaf       | Acetone      | 100                   | <b>36.71±1.21**</b>                 | <b>43.32±1.58**</b>              |
|   |            | Methanol     | 100                   | <b>34.22±1.49**</b>                 | <b>47.42±1.41**</b>              |
|   |            | Aqueous      | 100                   | 8.33±1.63                           | 13.83±1.63                       |
|   | Root       | Acetone      | 100                   | <b>37.61±1.73***</b>                | <b>48.44±1.53**</b>              |
|   |            | Methanol     | 100                   | <b>39.90±1.15***</b>                | <b>53.78±1.33**</b>              |
|   |            | Aqueous      | 100                   | <b>28.77±1.70*</b>                  | <b>30.15±1.70*</b>               |
| Epigallocatechin gallate  |            |              | 100                   | <b>48.62±1.14***</b>                | <b>84.31±1.24***</b>             |

\*: p&lt;0.05; \*\*: p&lt;0.01; \*\*\*: p&lt;0.001; S.E.M.: Standard error of the mean

The mixture was treated with 50 µL of 0.4 M NaOH and 100 µL of 0.2 M sodium borate and then incubated for 3 min in the boiling water. *p*-Dimethylaminobenzaldehyde solution (1.5 mL) was added to the reaction mixture after cooling to room temperature and was further incubated at 37°C for 20 minutes to develop a color. The absorbance of this colored solution was measured at 585 nm by using Beckmann Dual Spectrometer (Beckman, Fullerton, CA, USA).

#### Collagenase inhibition assay

The method for the collagenase inhibition was performed according to the method of Barrantes & Guinea (2003) with some modifications (20,21).

The samples were dissolved in DMSO. *Clostridium histolyticum* (ChC) was dissolved in 50 mM Tricine buffer (with 0.4M NaCl and 0.01M CaCl<sub>2</sub>, pH 7.5). Then, 2 mM N-[3-(2-Furyl)acryloyl]-Leu-Gly-Pro-Ala (FALGPA) solution was prepared in the same buffer. 25 µL buffer, 25 µL test sample and 25 µL enzyme were added to each well and incubated for 15 minutes. 50 µL substrate was added into the mixture. The decrease of the optical density (OD) was immediately measured at 340 nm using a spectrophotometer.

The ChC inhibitory activity of each sample was calculated according to the following formula:

ChC inhibition activity (%) =  $\frac{OD_{Control} - OD_{Sample}}{OD_{Control}} \times 100$  /  $OD_{Control}$  where  $OD_{Control}$  and  $OD_{Sample}$  represent the optical densities in the absence and presence of sample, respectively.

#### Elastase inhibition assay

According to the method described by Melzig et al. (2001), the sample solution and human neutrophil elastase enzyme (HNE) (17 mU/mL) were mixed in 0.1 M Tris-HCl buffer (pH 7.5). The mixture was incubated at 25°C for 5 min. N-Methoxysuccinyl-Ala-Ala-Pro-Val *p*-nitroanilide (MAAPVN) was added into the mixture and incubated at 37°C for 1 h. By the addition of 1 mg/mL soybean trypsin inhibitor, the reaction was stopped and the optical density was immediately measured at 405 nm. The HNE inhibitory activities were calculated according to the equation given in the ChC inhibitory assay (20,22).

#### Statistical analysis of the data

The data was statistically analyzed using one-way analysis of variance (ANOVA). The values of  $p \leq 0.05$  were considered statistically significant.

## RESULTS

Turkey is considered as one of the gene centers of *Asphodeline* genus with 20 taxa, as it is represented in Europe by only three species. Plants of the genus *Asphodeline* have traditionally been used as either food or therapeutic agent in various parts of the world as well as in Turkey (1,2). In the previous study, *Asphodeline* species were found to possess acetylcholinesterase, butyrylcholinesterase, amylase, glucosidase and tyrosinase inhibitory activities (3). Indeed, enzymes are known to be involved in the pathogenesis of several diseases, for instance hyaluronidase, collagenase and elastase enzymes involves in the pathogenesis of wound, cancer, cardiovascular diseases, inflammation, bone destruction and fibrosis (20,23-26). Collagen and elastin are the major components of the connective tissue and hyaluronic acid keeps the moist.

**Table 3. Hyaluronidase inhibitory activity of the extracts of *Asphodeline* species**

| Material  | Parts used | Extract type | Concentration (µg/mL) | Hyaluronidase inhibition (%) ± S.E.M. |
|---|------------|--------------|-----------------------|---------------------------------------|
| <i>A. brevicaulis</i> subsp. <i>brevicaulis</i> var. <i>brevicaulis</i> | Stem       | Acetone      | 100                   | 20.16±1.42                            |
|   |            | Methanol     | 100                   | 15.72±1.93                            |
|   |            | Aqueous      | 100                   | 14.78±2.16                            |
|   | Seed       | Acetone      | 100                   | 20.16±1.42                            |
|   |            | Methanol     | 100                   | 15.72±1.93                            |
|   |            | Aqueous      | 100                   | 14.78±1.46                            |
|   | Leaf       | Acetone      | 100                   | 21.49±1.32                            |
|   |            | Methanol     | 100                   | 15.72±1.93                            |
|   |            | Aqueous      | 100                   | 14.78±2.16                            |
|   |            | Root         | Acetone               | 100                                   |
|   |            | Methanol     | 100                   | 28.12±1.74                            |
|   |            | Aqueous      | 100                   | 14.39±1.44                            |
| <i>A. baytopae</i>  | Stem       | Acetone      | 100                   | 27.48±2.69                            |
|   |            | Methanol     | 100                   | 12.64±1.81                            |
|   |            | Aqueous      | 100                   | 7.74±1.21                             |
|   | Seed       | Acetone      | 100                   | 29.95±1.86                            |
|   |            | Methanol     | 100                   | 28.67±1.72                            |
|   |            | Aqueous      | 100                   | 10.46±2.18                            |
|   | Leaf       | Acetone      | 100                   | 22.83±1.39                            |
|   |            | Methanol     | 100                   | 28.37±2.66                            |
|   |            | Aqueous      | 100                   | 18.53±2.28                            |
|   |            | Acetone      | 100                   | <b>35.14±1.44*</b>                    |
|   | Root       | Methanol     | 100                   | 18.22±1.93                            |
|   |            | Aqueous      | 100                   | 13.82±2.94                            |
| <i>A. cilicica</i>  | Stem       | Acetone      | 100                   | 17.26±1.95                            |
|   |            | Methanol     | 100                   | 9.50±1.38                             |
|   |            | Aqueous      | 100                   | <b>45.51±1.25**</b>                   |
|   | Seed       | Acetone      | 100                   | 32.29±2.12                            |
|   |            | Methanol     | 100                   | 16.23±2.19                            |
|   |            | Aqueous      | 100                   | 12.21±1.63                            |
|   | Leaf       | Acetone      | 100                   | <b>39.26±1.11**</b>                   |
|   |            | Methanol     | 100                   | 42.30±1.14**                          |
|   |            | Aqueous      | 100                   | 16.48±1.86                            |
|   |            | Acetone      | 100                   | <b>47.20±1.01**</b>                   |
|   | Root       | Methanol     | 100                   | <b>49.49±1.17**</b>                   |
|   |            | Aqueous      | 100                   | 25.14±1.81                            |
| Tannic acid   |            |              | 100                   | <b>87.33±0.94***</b>                  |

\*: p<0.05; \*\*: p<0.01; \*\*\*: p<0.001; S.E.M.: Standard error of the mean

Inhibition of hyaluronidase, collagenase and elastase enzymes could therefore improve skin aging (20,27-29). In the present study we aimed to investigate *in vitro* inhibitory effects of the extracts prepared from the different parts of *A. baytopae*, *A. brevicaulis* subsp. *brevicaulis* var. *brevicaulis*, *A. cilicica* on hyaluronidase, collagenase and elastase enzymes.

The results revealed that the methanol extract prepared from the roots of *A. cilicica* (ACRM) displayed the highest hyaluronidase, collagenase and elastase inhibitory activity with the inhibition value of 49.49%, 39.90% and 53.78%, respectively. It was also found that acetone extract of the roots of *A. cilicica* (ACRAc) demonstrated 37.61% and 48.44% inhibitory effect on collagenase and elastase enzymes, respectively. In addition, acetone extract of the leaves of *A. brevicaulis* subsp. *brevicaulis* var. *brevicaulis* (ABrLAc) demonstrated inhibitory effect with the values of 31.38% and 39.39% on collagenase and elastase enzymes, respectively. Acetone and methanol extracts of the seeds of *A. baytopae* (ABaSdAc and ABaSdM), acetone and methanol extracts of the roots of *A. baytopae* (ABaRac and ABaRM), aqueous extract of the stems of *A. cilicica* (ACSmAq), acetone and methanol extracts of the leaves of *A. cilicica* (ACLAc and ACLM), aqueous extract of the roots of *A. cilicica* (ACRAq) possessed significant collagenase and elastase inhibitory activities. On the other hand, ABaRac, ACSmAq, ACLAc, ACLM and ACRAc were detected to possess significant hyaluronidase inhibitory effect (Table 2 and 3).

## DISCUSSION

According to the ethnobotanical studies, *Asphodeline* species such as *A. cilicica*, *A. damascena*, *A. globifera*, *A. lutea*, and *A. taurica* are consumed as food in salads (4). Due to the information regarding its consumption as food, the nutritional features of these species were investigated in our previous research revealing their high amount amino acid composition (4). *Asphodeline* species are used not only as food, but also as therapeutic agents for earaches, skin disorders and haemorrhoids in folk medicine (8,9,13). Due to their several medicinal utilization by people living in rural areas, *Asphodeline* species recently have attracted the researchers' attention to either verify the therapeutical usage in scientific platform or to investigate the phytochemical ingredients.

There have been several studies indicating the enzyme inhibitory activities of phenolic compounds and anthraquinones. Sawabe et al. (1998) investigated the inhibitory effects of water extracts obtained from sixty-six natural medicines on hyaluronidase, elastase and tyrosinase enzymes. The study pointed out that the enzyme inhibitory effect is positively correlated with high amount of phenolic content (30). Moreover, Lee et al. (2001) isolated a new phenolic compound, encoded CC-517, from *Areca catechu* L. and revealed its significant anti-hyaluronidase and anti-elastase activities. The compound inhibited human neutrophil elastase with the IC<sub>50</sub> value of 60.8 µg/mL; hyaluronidase with the IC<sub>50</sub> value of 210 µg/mL. It also exhibited more potent elastase inhibitory effect than oleanolic acid and ursolic acid (31).

Tanaka et al. (1990) conducted a study on collagenase inhibitory effect of 44 anthraquinone type compounds. Results of the

study demonstrated the inhibitory activity of anthraquinones, amongst emodin being the most potent active inhibitor with the  $IC_{50}$  value of  $4 \times 10^{-5}$  M (32). Furthermore, Zembower et al. (1992) synthesized several anthraquinone analogues and evaluated their elastase inhibitory activity on human leukocyte. Consequently, it was reported that 1,8 dihydroxyanthraquinone analogues possess elastase inhibitory effect (33).

Previous researches reported that *Asphodeline* species have high antioxidant capacity and phenolic content (7,13,34). Similarly, Zengin et al. (2015) recently investigated the antioxidant and enzyme inhibitory effects as well as anthraquinone profile of the methanol extracts obtained from the roots of eight *Asphodeline* species. According to the results, *A. cilicica* was found to possess the highest total phenolic content, while *A. brevicaulis* subsp. *brevicaulis* var. *brevicaulis* and *A. baytopae* had the highest total anthraquinone content. Therefore, the inhibitory effects of *A. cilicica*, *A. brevicaulis* subsp. *brevicaulis* var. *brevicaulis* and *A. baytopae* could be attributed to its phenolic and anthraquinone contents (3).

## CONCLUSION

In the present study, ACRM, ACRAc, ABrLAc, ABaSdAc, ABaSdM, ABArAc, ABArM, ACSmAq, ACRAq significantly inhibited collagenase, elastase and hyaluronidase enzymes, suggesting that these extracts could be used for the treatment of several diseases including wound, cancer, cardiovascular diseases, inflammation, bone destruction and fibrosis as well as potential ingredients for the cosmetic formulations to avoid skin aging.

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# Gastroprotective Effects of Various *Scrophularia striata* Extracts on Ethanol-Induced Gastric Ulcer in Rats

## *Scrophularia striata* Ekstrelerinin Ratlarda oluşturulan Etanol-Nedenli Gastrik Ülser Modeli Üzerindeki Gastroprotektif Etkileri

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### ABSTRACT

This study was conducted to investigate the influence of various extracts of *S. striata* in ethanol-induced gastric ulcer model in rats. 100 male rats were divided into 10 groups and received the following medications: Normal control group without treatment; comparative standard control group received 20 mg/kg omeprazole; Groups 4-9 were given aqueous, hydroalcoholic and etheric extracts at 100 and 400 mg/kg, respectively; vehicle control group given DMSO as solvent solution. After one hour all the rats (except normal control group) and also ulcer control group were given 4 mL/kg 75% EtOH solution to induce ulceration. The rats were sacrificed after one hour; the gastric mucosal injuries were estimated through assessment of the gross appearance of ulcer areas, histopathology and parameters including MDA, TAC, PGE<sub>2</sub> and HSP70 in the gastric tissue homogenate. The ulcer control group showed severe mucosal injury compared with aqueous and etheric extracts which grossly showed significant reduction of ulcer areas and histopathologically showed marked reduction of mucosal necrosis, edema and leukocytes infiltration. A significant increase in the levels of HSP70, PGE<sub>2</sub> and TAC with a reduction in the level of MDA was observed in the rats treated with etheric and especially aqueous extracts. The results of the present study revealed significant protection of *S. striata* towards ethanol-induced gastric mucosal injury.

**Key words:** Ethanol, Gastric ulcer, Gastroprotective, *Scrophularia striata*, Antioxidant, Histopathology

### ÖZ

Bu çalışma, *Scrophularia striata*'nın farklı polaritedeki ekstrelerinin ratlarda etanol-nedenli gastrik ülser modelindeki etkisini araştırmak amacıyla yürütüldü. Yüz erkek sıçan 10 gruba ayrıldı: Normal kontrol grubu hayvanlarına hiçbir tedavi uygulanmadı; karşılaştırmalı standart kontrol grubuna 20 mg/kg omeprazol; grup 4-9'a sırasıyla 100 ve 400 mg/kg dozlarda sulu, sulu alkollü ve eterli ekstreler; taşıyıcı kontrol grubuna ise çözücü olarak DMSO uygulandı. Bir saat sonra ülser oluşturmak için tüm ratlara (normal kontrol grubu dışında) ve ülser kontrol grubuna 4 mL/kg %75 ETOH verildi. Ratlar bir saat sonra sakrifiye edildi; gastrik mukozal hasarlar, gözle görünür ülser bölgelerinin tespit edilmesiyle, histopatolojik analizlerle ve mide dokusu homojenatında MDA, TAC, PGE<sub>2</sub> ve HSP70 belirlenmesiyle değerlendirildi. Ülser kontrol grubunda şiddetli mukozal hasar tespit edilirken, sulu ve eterli ekstre grubunda ülser alanlarında anlamlı küçülme, histopatolojik analizlerde mukozal nekrozda, ödem ve lökosit infiltrasyonunda azalma tespit edildi. Sulu ve eterli ekstre ile tedavi edilen ratlarda HSP70, PGE<sub>2</sub> ve TAC seviyelerinin anlamlı bir şekilde arttığı, MDA seviyesinin ise azaldığı görüldü. Çalışma sonuçları, etanol-nedenli gastrik mukozal hasara karşı *Scrophularia striata*'nın anlamlı derecede koruyucu etki sağladığını gösterdi.

**Anahtar kelimeler:** Etanol, Gastrik ülser, Gastroprotektif, *Scrophularia striata*, Antioksidan, Histopatoloji

### INTRODUCTION

Peptic ulcer disease is one of the most common disruptions of the mucosal integrity of the stomach (a gastric ulcer) and small intestine (a duodenal ulcer) (1). It has generally been accepted that gastric ulcers are multifactorial and appear to be due to an imbalance among the aggressive factors (such as acid/pepsin, bile, alcohol, tobacco and caffeine, *H. pylori* infection, NSAIDs, stresses) and mucosal defensive mechanisms (including mucus secretion, bicarbonate

production, mucosal blood flow, cellular repair mechanisms, prostaglandin E, growth factors) (2).

Peptic ulcer treatment using synthetic drugs (such as H2 blockers, proton pump inhibitors and NSAIDs) results in adverse effects, relapses and drug interactions (3). Therefore, herbal medicines containing active chemical components are considered as the main source of new drugs and appropriate alternatives for treatment of the various diseases including peptic ulcer.

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*Scrophularia striata*, commonly known as figwort, belongs to a family of flowering plants called Scrophulariaceae. It is native to Iran and grows wild in meadows, hillsides and impassable areas of Ilam Province (4). The Scrophulariaceae family consists of about 3000 species and 220 genera. Species of *Scrophularia* share square stems, opposite leaves and open two-lipped flowers forming clusters at the end of their stems (5).

Several chemical components including cinnamic acid, three flavonoids (quercetine, isorhamnetin-3-O-rutinoside and nepitrin) and one phenylpropanoid glycoside (acteoside 1) have been identified in the aerial parts of *S. striata* (6). It appears that some compounds isolated from this species have the inhibitory effects on a variety of malignant and inflammatory disorders (7). Hence, the present study was carried out to determine the gastroprotective effects of *Scrophularia striata* on ethanol-induced gastric ulcer.

## EXPERIMENTAL

### *Omeprazole*

In the present study, omeprazole was used as a comparative standard control drug for antiulcer study. The drug was dissolved in distilled water and administered orally to the rats at 20 mg/kg (5 mL/kg) according to previous study (8).

### *Plant specimen and preparation of extraction*

The aerial parts of *S. striata* were collected from the Zagros mountain range, Kermanshah Province, in May 2014. The plant sample was authenticated and voucher specimen was deposited at the Herbarium of Faculty of Sciences, Kharazmi University, Tehran, Iran (No: 5379). The plant aerial parts were air dried at room temperature and made into powder using a blender. The aqueous, hydroalcoholic and etheric extracts were obtained by maceration of the 200 g plant powder with 1L distilled water, ethanol/water (70/30) and petroleum ether for 2 days at room temperature, respectively. Extracts were filtered through a Whatman#1 paper and the solvents were evaporated under reduced pressure at temperature below 45 °C with a rotary evaporator. Then the filtrates were lyophilized in a lyophilizer and stored under light protection and low temperature (-4 °C) prior to use.

The aqueous and hydroalcoholic extracts were then dissolved in distilled water and etheric extract was dissolved in dimethyl sulfoxide (DMSO). The extracts were administered orally (4 mL/kg) to rats at dosages of 100 and 400 mg/kg.

### *Experimental animals for gastric ulcer*

100 adult healthy male rats of *Sprague Dawley* strain weighing 200-250 g were housed in stainless steel cages and allowed to adapt to the conditions of the animal house for 14 days before the experiments. Animals were divided randomly into 10 equal groups of 10. 24 hours before the experiment, the rats were fasted and allowed access to water. Their access to water was inhibited for 2 hours before the start of experiment.

### *Gastric ulcer induction by ethanol*

The experimental protocol is detailed below: Normal control group (NC) was left without treatment. Standard control group (SC) received omeprazole (20 mg/kg) orally. Experimental groups were orally administered with aqueous extract at 100 mg/kg (AE100) and 400 mg/kg (AE400), hydroalcoholic extract at 100 mg/kg (HA100) and 400 mg/kg (HA400) and etheric extract at 100 mg/kg (EE100) and 400 mg/kg (EE400). Vehicle control group (VC) received DMSO (4 mL/kg) orally. After one hour all the rats (except normal control group) and also ulcer control group (UC) were orally administered with 75% ethanol (4 mL/kg).

After one hour, rats were sacrificed by an overdose of ether and stomachs were isolated and cut open along the greater curvature. Stomachs were gently rinsed with 0.9% normal saline solution to remove gastric contents and blood.

### *Gross evaluation of gastric lesions*

The ulcer area (mm<sup>2</sup>) of each individual hemorrhagic lesion was measured and analyzed using computer software (Axiovision, Carl Zeiss Microimaging GmbH, Germany). According to the method of Andrade et al. (9), a score for the ulcer was noted and the ulcers were classified as: Level I ulcer area <1 mm<sup>2</sup>; Level II ulcer area = 1-3 mm<sup>2</sup>; Level III ulcer area >3 mm<sup>2</sup>.

The following parameters were determined:

Ulcerative lesion index (ULI) = 1 × (number of ulcers level I) + 2 × (number of ulcers level II) + 3 × (number of ulcers level III).

Percentage protective ratio = 100 - [ULI pretreated] / [ULI control] × 100.

### *Histopathological evaluation*

A portion of gastric tissue was collected from animals and fixed in 10% neutral buffered formalin, embedded in paraffin, sectioned at 5 μm and stained with haematoxylin and eosin (H&E) for light microscopic examination.

### *Preparation of homogenate*

Tissue homogenates were prepared for the PGE<sub>2</sub>, MDA, TAC and HSP70 assays in gastric tissue of experimental groups. All the processes were handled at 4 °C throughout according to previous study (10). The gastric mucosa was weighed, minced with scissors, and homogenized using 0.1 M phosphate buffer (pH 7.4) (5 cc for each g of tissue) in homogenizer. After centrifugation at 2000-3000 rpm for 20 min, the supernatant was extracted and frizzed in -80 °C for later use.

### *Measurement of PGE<sub>2</sub>*

The supernatants were subjected to a PGE<sub>2</sub> assay using a rat PGE<sub>2</sub> Eliza kit (Shanghai Crystal Day Biotech Co., LTD).

### *Measurement of membrane lipids peroxidation (MDA)*

Tissue malondialdehyde (MDA) (mmol/L) was determined according to the method of Lykkesfeldt. A reaction mixture containing 8.1% sodium dodecyl sulfate, 20% acetate buffer (pH 3.5) and 0.8% thiobarbituric acid (TBA) was mixed well



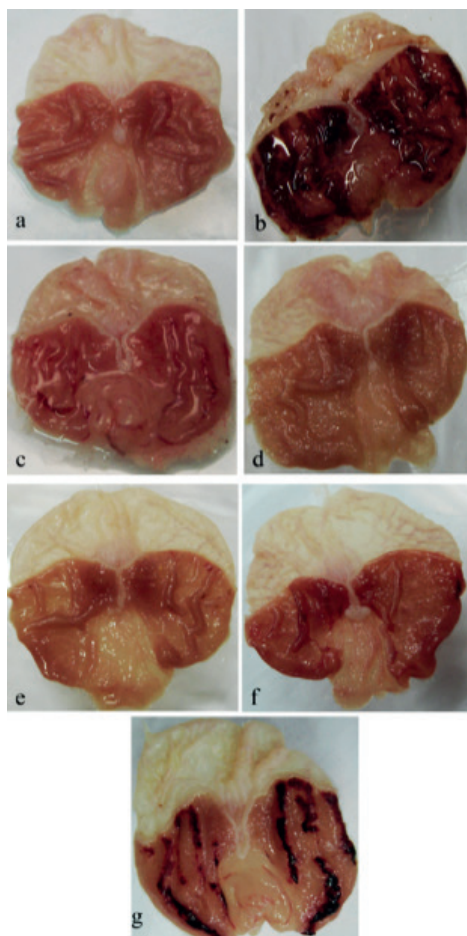
with 0.2 mL of stomach tissue homogenate for 3 min and then incubated at 95°C for 60 min. After cooling with running water, the TBA-reactive substance (MDA) was extracted with 1 mL of H<sub>2</sub>O and 2.5 mL of n-butanol: Pyridine mixture (15:1, v/v). The upper organic layer containing the MDA, which was produced by lipid peroxidation, was measured at 532 nm (11,12).

#### Measurement of total antioxidant capacity (TAC)

Determination of total antioxidant capacity (TAC) in tissue homogenate by commercial kit (Labor Diagnostika Nord (LDN) Com, Nordhorn, Germany) was based on the reaction of peroxides with peroxidase followed by a color reaction of the chromogenic substrate tetramethylbenzidine. The change in color was measured colorimetrically at 450 nm and expressed as millimoles per liter.

#### Measurement of heat shock protein 70 (HSP70)

The supernatants were subjected to a HSP70 assay using a rat HSP Eliza kit (Shanghai Crystal Day Biotech Co., LTD).



**Figure 1.** The effects of *S. striata* extracts on ethanol-induced gastric ulcer. a) Normal control group; b) Ulcer control group; c) Rats treated with omeprazole; d) Rats treated with aqueous extract at 400 mg/kg; e) Rats treated with 100 mg/kg aqueous extract; f) Rats treated with 400 mg/kg etheric extract; g) Rats treated with 400 mg/kg hydroalcoholic extract

#### Statistical analysis

The results were expressed as mean  $\pm$  standard deviation (SD). The data were analyzed statistically by one-way ANOVA with Tukey's post-hoc test, using SPSS software, version 20.  $P < 0.05$  was considered as significant.

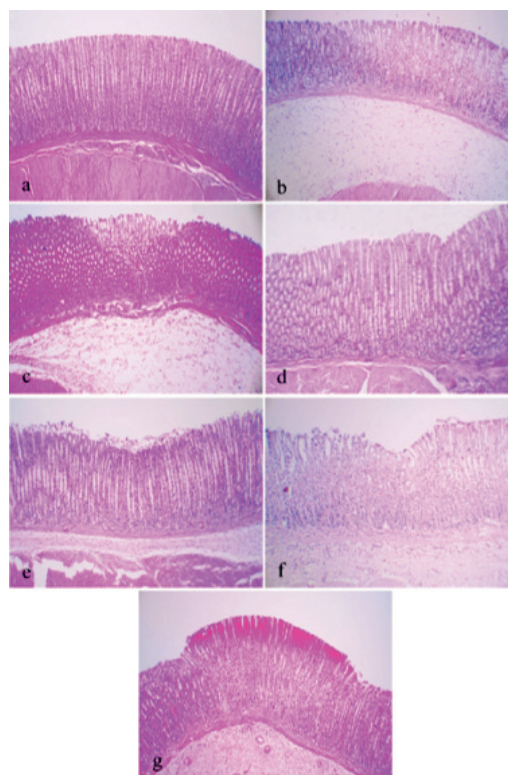
#### Animal ethics

This experiment was accomplished under the approval of the state committee on animal ethics, Shiraz University, Shiraz, Iran. Also, the recommendations of European Council Directive (86/609/EC) of November 24, 1986, regarding the standards in the protection of animals were used for experimental purposes.

## RESULTS

#### Gross evaluation of gastric lesions

Gastroprotective effects of *S. striata* extracts on ethanol-induced gastric ulcer are shown in Figure 1 and Table 1. The ulcer control rats showed severe mucosal injury with ulcer area  $20.75 \pm 11.09$  mm<sup>2</sup>. In the rats treated with aqueous and



**Figure 2.** Histopathological evaluation of gastric tissue sections of rats. a) Normal control group: No lesion was seen; b) Ulcer control group: Mucosal necrosis, severe edema with leukocytes infiltration in submucosal layer; c) Rats treated with omeprazole: Mild mucosal damage with moderate edema and infiltration of leukocytes in the submucosal layer; d) Rats treated with aqueous extract at 400 mg/kg: No edema and leukocyte infiltration was seen; e) Rats treated with 100 mg/kg aqueous extract: Mild edema and little leukocyte infiltration in the submucosal layer; f) Rats treated with 400 mg/kg etheric extract: Mild mucosal damage with mild edema and infiltration of leukocytes; g) Rats treated with 400 mg/kg hydroalcoholic extract: Moderate edema and infiltration of leukocytes in the submucosal layer with mucosal necrosis. H&E,  $\times 80$

etheric extracts, the ulcer area was significantly reduced in a dose-dependent manner ( $p < 0.05$ ). The ulcer area was significantly decreased from  $20.75 \pm 11.09 \text{ mm}^2$  in ulcer control group to  $1.00 \pm 0.86 \text{ mm}^2$  in the rats treated with 400 mg/kg of aqueous extract. Also, in the rats treated with omeprazole, a decrease was observed in ulcer area ( $5.70 \pm 4.47 \text{ mm}^2$ ). Although the ulcer area in all groups had statistically significant difference with the rats treated with hydroalcoholic extract and ulcer control group, the significant inhibition of gastric ulcer in rats treated with aqueous and etheric extracts of *S. striata* was comparable to omeprazole (a standard drug used for gastric ulcer).

#### Histopathological evaluation of gastric lesions

No lesions were observed in the tissue sections from the rats of normal control group (Figure 2a), while the rats of ulcer control group showed extensive damage to the gastric tissue, including mucosal hemorrhage and necrosis, severe edema and leukocyte infiltration in the submucosal layer (Figure 2b). In the rats treated with omeprazole, mild mucosal damage with moderate edema and infiltration of leukocytes in the submucosal layer was observed (Figure 2c).

The tissue sections of the group treated with aqueous extract at 400 mg/kg was near-normal architecture and no edema and leukocyte infiltration were seen. The rats treated with 100 mg/kg aqueous extract showed the normal glandular pattern with mild edema and little leukocyte infiltration in the submucosal layer (Figure 2d and 2e).

The tissue sections of groups treated with 100 and 400 mg/kg etheric extract revealed mild mucosal damage with mild edema and infiltration of leukocytes Figure 2f.

In the rats treated with 100 and 400 mg/kg hydroalcoholic extract, moderate edema and infiltration of leukocytes in the gastric submucosal layer with mucosal necrosis was observed in which mucosal damage was more severe at 400 mg/kg (Figure 2g).

**Table 1. The effect of *S. striata* extracts in ethanol-induced gastric ulcer**

| Group | Ulcerative lesion index | Percentage of protective ratio (%) |
|-------|-------------------------|------------------------------------|
| PC    | $20.75 \pm 11.09^a$     | -                                  |
| SC    | $5.70 \pm 4.47^b$       | 72.53                              |
| AE100 | $1.00 \pm 0.92^b$       | 95.18                              |
| AE400 | $1.00 \pm 0.86^b$       | 95.18                              |
| HA100 | $18.14 \pm 7.96^a$      | 12.58                              |
| HA400 | $21.50 \pm 5.48^a$      | 3.61                               |
| EE100 | $4.00 \pm 3.78^b$       | 80.72                              |
| EE400 | $1.14 \pm 1.06^b$       | 94.5                               |
| VC    | $5.29 \pm 3.30^b$       | 74.5                               |

Different letters indicate statistically significant differences ( $p < 0.05$ )

In the tissue sections of vehicle group, lesions were similar to the rats treated with omeprazole.

#### Evaluation of parameters in the gastric tissue homogenate

The mean  $\pm$  SD of parameters values, including MDA, TAC,  $\text{PGE}_2$  and HSP70 in the gastric tissue homogenate of rats are presented in Table 2.

MDA increased in all groups in comparison with ulcer control group and the highest amount of MDA was observed in the rats of ulcer control group which showed a significant difference with all groups ( $p < 0.05$ ). There was no significant difference between normal control group and rats treated with aqueous extract of *S. striata*. The MDA level in the groups treated with other extracts and omeprazole had statistically significant difference with the normal control group and the rats treated with aqueous extract (Figure 3).

The most significant reduction in TAC was observed in rats of ulcer control group. Although the TAC level of normal control group showed a significant difference with all groups ( $p < 0.05$ ), this parameter increased in the rats treated with various extracts and was statistically significant in groups treated with aqueous and etheric extracts in comparison with the ulcer control group. In addition, the TAC level in rats treated with aqueous extract was higher as compared with the rats treated with omeprazole and showed significant difference (Figure 4).

Although the  $\text{PGE}_2$  level presented significant difference between normal control group and other groups, this parameter increased in the rats treated with aqueous and etheric extracts in comparison with ulcer control group and showed significant difference. There was no significant difference between the rats treated with hydroalcoholic extract and ulcer control group (Figure 5).

**Table 2. The mean  $\pm$  SD of parameters value in the gastric tissue homogenate of rats**

| Group | $\text{PGE}_2$ (ng/mL) | HSP70 (ng/mL)        | MDA (nmol/mL)     | TAC (mmol/L)         |
|-------|------------------------|----------------------|-------------------|----------------------|
| NC    | $0.26 \pm 0.01^a$      | $0.53 \pm 0.06^a$    | $2.96 \pm 0.11^b$ | $2.96 \pm 0.11^a$    |
| UC    | $0.13 \pm 0.01^b$      | $0.39 \pm 0.05^b$    | $3.66 \pm 0.18^a$ | $1.73 \pm 0.25^b$    |
| SC    | $0.22 \pm 0.01^c$      | $0.46 \pm 0.04^{ab}$ | $3.27 \pm 0.05^c$ | $2.01 \pm 0.03^{ce}$ |
| AE100 | $0.23 \pm 0.01^c$      | $0.50 \pm 0.04^a$    | $3.08 \pm 0.05^b$ | $2.28 \pm 0.05^d$    |
| AE400 | $0.23 \pm 0.01^c$      | $0.51 \pm 0.04^a$    | $3.01 \pm 0.04^b$ | $2.28 \pm 0.03^d$    |
| HA100 | $0.14 \pm 0.01^b$      | $0.40 \pm 0.06^b$    | $3.43 \pm 0.10^d$ | $1.82 \pm 0.11^b$    |
| HA400 | $0.13 \pm 0.01^b$      | $0.40 \pm 0.10^b$    | $3.46 \pm 0.08^d$ | $1.84 \pm 0.09^{bc}$ |
| EE100 | $0.22 \pm 0.01^c$      | $0.48 \pm 0.04^{ab}$ | $3.28 \pm 0.05^c$ | $2.01 \pm 0.03^{ce}$ |
| EE400 | $0.21 \pm 0.00^c$      | $0.47 \pm 0.03^{ab}$ | $3.28 \pm 0.05^c$ | $2.03 \pm 0.04^e$    |
| VC    | $0.21 \pm 0.01^c$      | $0.45 \pm 0.03^{ab}$ | $3.07 \pm 0.03^b$ | $2.24 \pm 0.03^d$    |

Different letters indicate statistically significant differences ( $p < 0.05$ )

No statistically significant difference was observed in the HSP70 level between the rats treated with aqueous extract and the normal control group, but there was a significant difference between these groups with ulcer control group ( $p < 0.05$ ). Although this parameter increased in the rats treated with etheric extract and omeprazole, no significant difference was seen with other groups (Figure 6).

In general, the most curative effect of *S. striata* was related to aqueous extract. Although the hydroalcoholic extract resulted in improvement of some parameters, it did not have curative effects.

## DISCUSSION

A key experimental model for evaluation of agents with potential anti-ulcer effect is ethanol-induced gastric injury as ethanol has been considered as a cause of gastric ulcer in humans (13). Oral administration of ethanol results in gastric mucosal damage and alterations in vascular (14). In fact, ethanol causes gastric mucosal injury via direct effects including dehydration, disruption of cellular membranes and cytotoxic effects and also indirect effects via the recruitment of leukocytes (15). In addition, ethanol leads to stasis of blood flow and disruption of gastric microvessels, which in turn inflict hemorrhage and necrosis (15,16). Ethanol-induced

necrotic lesions decrease defensive factors such as the mucus production and secretion of bicarbonate (17).

The mentioned effects are probably because of biological actions like lipid peroxidation, intracellular oxidative stress, formation of free radicals, changes in permeability and depolarization of the mitochondrial membrane (18). The defensive mechanism against free radicals is weakened by ethanol (19). Ethanol consumption results in hemorrhagic damage, severe submucosal edema and epithelial cell injury. In the present study, similar lesions were also observed in gastric tissue sections of ulcer control group. The results of this study revealed aqueous extract of *S. striata*, especially at 400 mg/kg have therapeutic effects on ethanol-induced gastric ulcer as tissue structure was near-normal architecture and no edema and leukocyte infiltration were seen in tissue sections.

With both short and long-term use, omeprazole is effective in the treatment of gastroesophageal reflux and peptic ulcer disease (20). Omeprazole functions as an acid inhibitor and offers a protective role such as gastric mucosa (21). Also, this agent, as mucosal protection, is effective in the treatment of nonacid dependent models like ethanol-induced ulcer (22). In the rats treated with omeprazole, mild mucosal damage

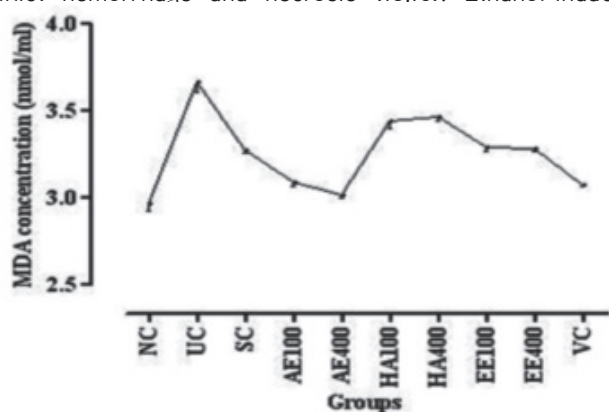


Figure 3. The effect of *S. striata* extracts on MDA levels in the gastric tissue homogenate of rats

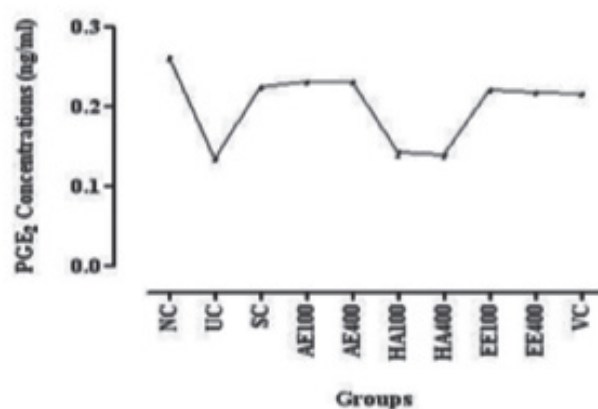


Figure 5. The effect of *S. striata* extracts on PGE<sub>2</sub> levels in the gastric tissue homogenate of rats

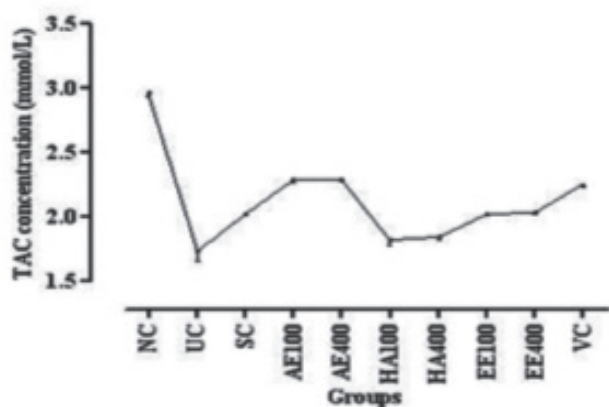


Figure 4. The effect of *S. striata* extracts on TAC levels in the gastric tissue homogenate of rats

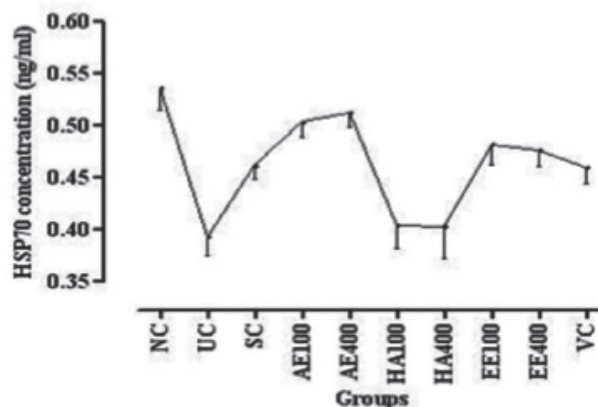


Figure 6. The effect of *S. striata* extracts on HSP70 levels in the gastric tissue homogenate of rats

with moderate edema and infiltration of leukocytes in the submucosal layer were seen which were similar to those observed in the rats treated etheric extract.

Treatment of the rats with *S. striata* extract showed significant antioxidant activity by reduction of MDA and increase of the TAC level in response to ethanol-induced oxidative stress. Oxidative stress plays a key role in the pathogenesis of different diseases including gastric ulcer, and antioxidants perform a significant role in protection of gastric mucosa against necrotic injury (23). Several studies have reported the involvement of oxidative stress in the pathogenesis of ethanol-induced gastric injury (13,15). Ethanol causes damage to the gastric mucosal microcirculation, resulting in hypoxia, formation of free radicals and lipid peroxidation (24). It seems free radicals formation plays an important role in the production of lipid peroxides together with interference in antioxidant activity (25). Free radicals can reduce enzyme activity such as antioxidant enzymes (26).

MDA is the final product of lipid peroxidation (27). An important pathophysiologic event in various diseases containing gastric ulcer is lipid peroxidation (28) which leads to impaired membrane integrity and ion transport, and finally, loss of cellular function. Oxygen free radicals produced in the gastric tissue might damage cell membranes and increase MDA level (29).

One mechanism involved in the healing of ulcer is removal of oxygen free radicals (Mei 30). Antioxidants like catalase and SOD are the first defensive barrier against free radicals by removing them and preventing their harmful effects (15). Various studies have reported the cytoprotective effect of some antioxidants in the healing of gastric lesions. For instance, melatonin prevented ethanol-induced gastric damage, probably because of its antioxidant effect (31).

Previous studies have shown prostaglandins affect various components of the mucosal defense, including maintaining blood flow, stimulating bicarbonate and mucus secretion, increasing the resistance of epithelial cells to injury and preventing leukocyte recruitment (32). In the present study, ethanol consumption led to decreasing PGE<sub>2</sub> that was consistent with previous studies (13,22). PGE<sub>2</sub> is the most abundant prostaglandin of alimentary system which performs an important role in the regulation of gastric mucus secretion (33) and its mucosal content can be reduced by ethanol (34). PGE<sub>2</sub> improves blood flow, actively maintaining the cellular integrity in the mucosa (34). In addition, this prostaglandin increases mucus secretion and bicarbonate to reinforce the resistance of mucosal cells to the necrotizing effect of irritants (33). The findings of the present study revealed that the gastroprotective effect of *S. striata* is related to PGE<sub>2</sub> because its mucosal level was elevated by compound.

HSPs play an important role in both normal and pathologic situations (35). HSP70 is a 70 kDa protein and a member of heat shock proteins family (36) which is expressed

by mammalian cells. These proteins act as a molecular chaperone and protect the cellular homeostatic processes from various injurious agents through preservation of the structure of normal proteins and repair or removal of damaged proteins (37,38). HSP70 protects mitochondria and interferes with the stress-induced apoptosis, resulting in cytoprotection (38). Ethanol-induced oxygen reactive species inhibit the expression of HSP70 and increase the expression of BAX, while HSP70 protects cells from oxidative stress or heat shock (39).

Jin et al. (40) reported that when the PGE<sub>2</sub> level is decreased by NSAID, HSP70 can play an important role in gastric mucosal adaptation. In the present study, the HSP70 levels increased in various groups, which was observed as a significant difference between the rats treated with aqueous extract and ulcer control group. Oyake et al. (41) showed the overexpression of Hsp70 protects gastric mucosal against monochloramine-induced damage. Also, HSP70 induction protected rats from ethanol-induced gastric mucosal injury (15).

In conclusion, the findings of the present study indicated the efficiency of *S. striata* extract in improving free radicals-induced injury and exerting the protective effects in ethanol-induced gastric ulcer model. These effects were related to increased PGE<sub>2</sub> secretion, prevention of consumption of antioxidant sources and maintaining MDA at normal level.

## ACKNOWLEDGMENTS

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# Voltammetric Determination of Nimesulide Using Multiwalled Carbon Nanotubes Modified Carbon Paste Electrode

## Çok Duvarlı Karbon Nanotüp ile Modifiye Edilmiş Karbon Pasta Elektrot Kullanılarak Nimesulidin Voltametrik Miktar Tayini

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### ABSTRACT

A multiwalled carbon nanotubes (MWCNTs) modified carbon paste electrode (CPE) was prepared for voltammetric determination of anti-inflammatory drug nimesulide (NIM). The electro-oxidation of NIM was exhibited irreversible and diffusion controlled process with MWCNTs modified CPE. The linear response between peak current and concentration in the quantitative determination of NIM by differential pulse voltammetry in 0.1 M phosphate buffer solution (PBS) at pH 5.0 obtained in the range of the concentration from  $6 \times 10^{-8}$  -  $1 \times 10^{-5}$  M with limit of detection (LOD)  $1.07 \times 10^{-9}$  M and limit of quantification (LOQ)  $3.24 \times 10^{-9}$  M. Differential pulse voltammetry was developed according to linear response of NIM with high selectivity, precision, accuracy using modified electrode was successfully applied to the determination of NIM in pharmaceuticals and human serum samples.

**Key words:** Carbon paste electrode, Determination, Multiwalled carbon nanotubes, Nimesulide, Voltammetry

### ÖZ

Antienflamatuvar ilaç etken maddesi nimesulidin (NIM) voltametrik miktar tayini için çok duvarlı karbon nanotüp ile modifiye edilmiş karbon pasta elektrot hazırlanmıştır. NIM'in çok duvarlı karbon nanotüp ile modifiye edilmiş karbon pasta elektrot ile elektro-oksidasyonu tersinmez ve difüzyon kontrollü bir özellik göstermiştir. NIM'in diferansiyel puls voltametrisi ile 0.1 M fosfat tampon çözeltisinde pH 5.0 de miktar tayininde derişim ve pik akımı arasındaki doğrusallık  $6 \times 10^{-8}$  -  $1 \times 10^{-5}$  M derişim aralığında, saptama sınırı  $1.07 \times 10^{-9}$  M ve tayin alt limiti  $3.24 \times 10^{-9}$  M olarak bulunmuştur. Çok duvarlı karbon nanotüp ile modifiye edilmiş karbon pasta elektrodu kullanarak NIM'nin doğrusal cevabına göre yüksek seçicilik, kesinlik, ve doğrulukla geliştirilen diferansiyel puls voltametrisi, NIM'nin farmasötik preparatlardan ve insan serum numunelerinden miktar tayinine başarılı bir şekilde uygulanmıştır.

**Anahtar kelimeler:** Karbon pasta elektrot, Tayin, Çok duvarlı karbon nanotüp, Nimesulid, Voltametri

### INTRODUCTION

Nimesulide, N-(4-nitro-2-methanesulfonamide) (Figure 1), is a new non-steroidal anti-inflammatory drug that is selective for cyclooxygenase-2 and effective in reducing the pain which is associated with rheumatoid arthritis and osteoarthritis (1).  $pK_a$  value of NIM is 6.46 that is very important for gastric tolerability, and this avoids the back diffusion of the hydrogen ions that are liable for tissue damage. NIM is nearly completely biotransformed to 4-hydroxynimesulide in free and conjugated forms and it provides to promote to the anti-inflammatory activity of NIM (2,3).

There are several reports on the determination of NIM in the literature for example HPLC (high performance

liquid chromatography) (4), with spectrophotometry UV (ultraviolet) (5,6), capillary electrophoresis (7). Most of these methods appear as time-consuming, expensive, complicated and lengthy procedures. Electrooxidation of NIM at gold electrode (8), electroreduction of NIM using glassy carbon electrode modified with SiC (silicon carbide) (9) and MWCNTs (multiwalled carbon nanotubes) modified glassy carbon electrode (10) have been reported in the literature. Electrochemical methods have certain advantages for example fast response, low cost and high sensitivity compare to other analytical methods (11,12).

Carbon nanotubes (CNTs) can be used as electrode material for electrochemical and bioelectrochemical applications

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because of their unique characteristics and useful properties for example high chemical stability, conductivity, aspect ratio, and extremely high mechanical strength and modules (13-16). CNTs are largely used as working electrode modification material for drug analysis due to they have the capability for promoting electron transfer reactions and developing sensitivity in electrochemistry (17).

In this study, a sensitive MWCNTs modified CPE was prepared for electroanalytical determination and it used to investigate electro-oxidative behavior of NIM with cyclic and differential pulse voltammetry. The prepared MWCNTs modified CPE was exhibited rapid response, high selectivity, sensitivity, low detection limit, and good reproducibility and successfully used electroanalytical determination of NIM.

## EXPERIMENTAL

### Instrumentation

Voltammetric measurements were carried out with a computer-controlled Autolab Pgstat128n potentiostat/galvanostat with Nova 10.0 software (Metrohm-Autolab, The Netherlands). A three-electrode electrochemical cell analyzer contains a carbon paste electrode with modification

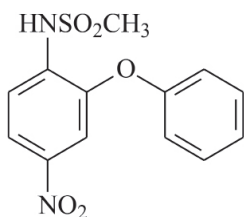


Figure 1. Molecular structure of nimesulide

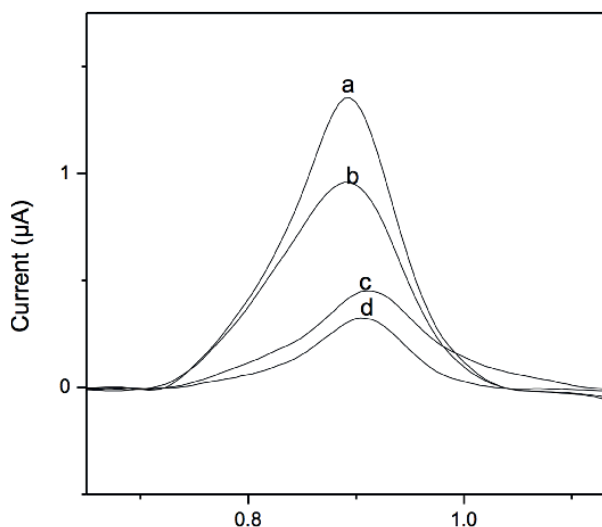


Figure 2. Differential pulse voltammograms 10  $\mu\text{M}$  of NIM in 0.04 M Britton-Robinson buffer at pH 7.0, (a) 0.2%, 2.5  $\mu\text{L}$ ; (b) 0.5%, 1  $\mu\text{L}$ ; (c) 0.5%, 3  $\mu\text{L}$ , (d) 0.2%, 5  $\mu\text{L}$  of MWCNTs

MWCNTs as working electrode, a platinum wire as the counter electrode and Ag/AgCl electrode as the reference electrode. The pH measurements were carried out using model Hanna HI2211 pH meter (Romania) with an accuracy of  $\pm 0.05$  pH at room temperature.

### Reagents

Nimesulide and its pharmaceutical dosage form tablet (100 mg per tablet) were supplied by Sanovel-Turkey. They were used without further purification. Stock solutions of NIM ( $1 \times 10^{-3}$  M) were prepared in methanol and stored at  $+4^\circ\text{C}$  away from light. NIM working solutions for voltammetric investigation were prepared by the direct dilution of the stock solution with selected supporting electrolyte containing a constant amount of methanol (20% (v/v)). Graphite powder ( $d=2.2$  g/mL, Merck, Germany) and paraffin oil ( $d=0.84$  g/mL, Aldrich, U.S.A) as the binding agent were used for preparing the pastes. MWCNTs was purchased from NanoLab, U.S.A, with purity 95%,  $30 \pm 10$  nm diameter, and 1-5  $\mu\text{m}$  lengths.

Phosphate buffer solutions (PBS) (0.1 M) were prepared from phosphoric acid (Merck, Germany) for pH 4.0 and disodium hydrogen phosphate (Aldrich, U.S.A.), sodium dihydrogen phosphate (Merck, Germany) for pH 5.0-8.0. Britton-Robinson (BR) buffer solutions (0.04 M) were prepared at pH 3.0-9.0 from 0.04 M phosphoric acid (Merck, Germany), 0.04 M boric acid (Aldrich, U.S.A.) and 0.04 M acetic acid (Merck, Germany). Acetate (AT) buffer solutions (1 M) at pH 3.5, 4.5 were prepared from 1 M acetic acid (Merck, Germany). pH was adjusted with 5 M sodium hydroxide (Aldrich, U.S.A.) solution.

Sartorius Arium proUV nanopure water (resistivity  $\geq 18$  M $\Omega$  cm), and analytical reagents were used for the preparation

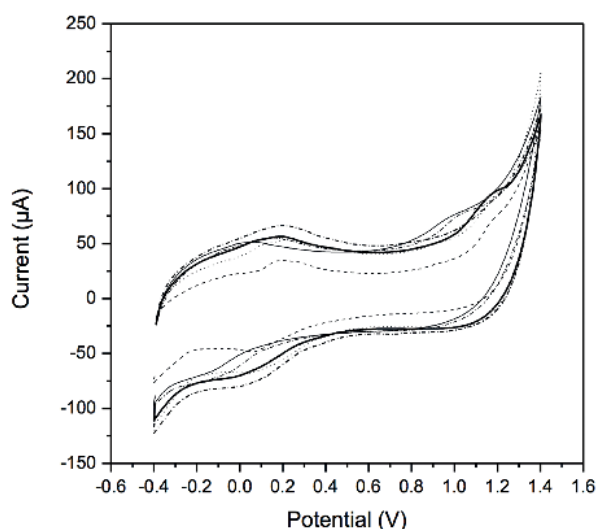


Figure 3. Cyclic voltammograms of NIM in 1 M acetate buffer at pH 3.5 (-----), 0.1 M phosphate buffer at pH 4.0 (.....), pH 5.0 (—), 0.04 M Britton-Robinson buffer pH 7.0 (-.-.-), pH 9.0 (—) with MWCNTs modified CPE. Short dash dot line 0.1 M phosphate buffer at pH 5.0; NIM concentration: 100  $\mu\text{M}$ ; scan rate 100 mV/s

of solutions. All of the experiments were performed at room temperature ( $25\pm 1^\circ\text{C}$ ).

#### Preparation of bare and MWCNTs modified carbon paste electrodes

The ratio of graphite powder and paraffin oil to binder were optimized for NIM, and then the carbon paste electrode was prepared homogeneous paste by thoroughly hand-mixing the from optimized graphite powder and paraffin oil in the ratio of 75:25 (w/w). A portion of the homogeneous paste was packed into the cave of the teflon tube. A copper wire inserted into the carbon paste provided the electrical contact. When necessary, a new surface was obtained by pushing an excess of the paste out of the tube and polishing with a weighing paper.

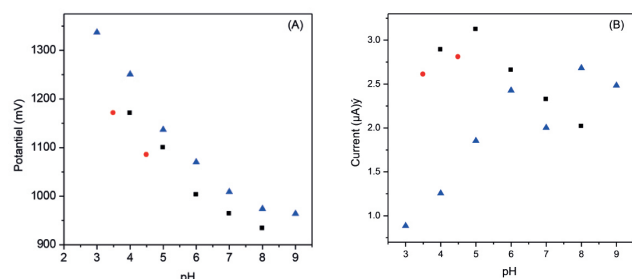
The MWCNTs were dispersed in DMF with loading 0.2% (w/v) and sonicated for 4h to obtain a homogeneous mixture. A selected 2.5  $\mu\text{L}$  of the dispersion was dropped directly on the surface of CPE. The resulting modified electrode was named as MWCNTs modified CPE. The MWCNTs modified CPE electrode dried for overnight at room temperature.

#### Pharmaceutical assay

Ten tablets (each tablet contains 100 mg NIM) were first weighed and then finely powdered. The required amount of powder equivalent to  $10^{-3}$  M of NIM was diluted to 100 mL with methanol and sonicated for 15 min. The analyzed solutions were prepared by taking suitable aliquots of the clear supernatant liquor and diluting with the selected supporting electrolyte. NIM working solutions for voltammetric investigation were prepared by the direct dilution of the stock solution with selected supporting electrolyte containing a constant amount of methanol (20% v/v).

#### Analysis of serum

Drug-free human serum samples were obtained from healthy people and stored frozen in the dark until assay. An aliquot volume of serum sample was fortified with NIM dissolved in methanol to achieve final concentration of  $1 \times 10^{-3}$  M and treated with acetonitrile to removing serum proteins effectively. Blank and stock solution of NIM were transported to ultrasonic bath and agitated for 15 min and subsequently centrifuged for 15



**Figure 4.** Plots of peak potential, ( $E_p$ ), versus pH (A) and peak current ( $I_p$ ), versus pH (B) from cyclic voltammetry voltammograms of 100  $\mu\text{M}$  of NIM with MWCNTs modified CPE. Squares indicate 0.1 M phosphate buffer solution, triangle 0.04 M Britton-Robinson buffer solution and circles 1 M acetate buffer solution

min at 5000 rpm to separate serum protein residues and supernatant. Appropriate volumes of this supernatant were taken carefully and transferred into the volumetric flask and diluted up to the required volume with the selected supporting electrolyte containing a constant amount of methanol (20% v/v).

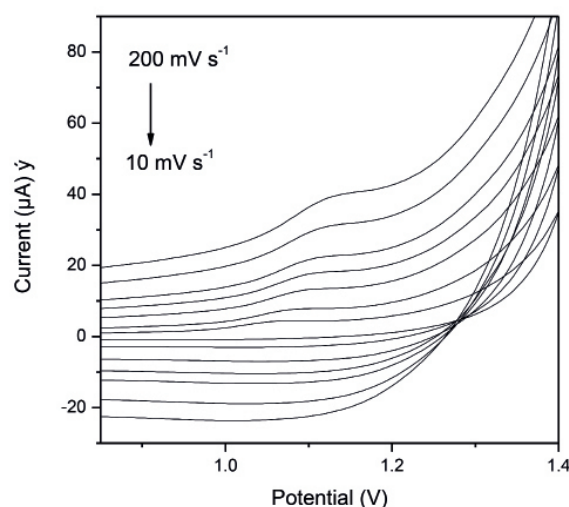
#### Validation of the analytical methods

The ruggedness, precision, and accuracy of the studied methods, were checked by assaying five replicate samples on the same day and on different days over a week. Relative standard deviations (%) were also calculated to check the ruggedness and precision of the method. The accuracy of the methods was expressed as bias (%) (18,19). Each of the solutions was freshly prepared just before the experiments and protected from the light. All of the measurements were carried out at room temperature ( $25\pm 1^\circ\text{C}$ ). The calibration equation for differential pulse voltammetry method was constructed by plotting the peak current against NIM concentration.

## RESULTS AND DISCUSSION

Effect of volume variations on the peak current was investigated at two different concentrations of MWCNTs suspension (0.2% and 0.5% in DMF) to optimize MWCNTs volume for determination of NIM (Figure 2). The suspension amounts of 2.5  $\mu\text{L}$  and 5  $\mu\text{L}$  for 0.2% of MWCNTs, 1  $\mu\text{L}$  and 3  $\mu\text{L}$  for 0.5% of MWCNTs suspension were studied for 10  $\mu\text{M}$  NIM with CV and DPV. As shown in Figure 1, the peak current reaches its maximum value when the suspension amount is 2.5  $\mu\text{L}$  for 0.2% MWCNTs. So, 2.5  $\mu\text{L}$  was chosen as the optimized amount for 0.2% MWCNTs suspension.

The electro-oxidation behavior of NIM on MWCNTs modified CPE was studied by CV at a scan rate of 100 mV/s between

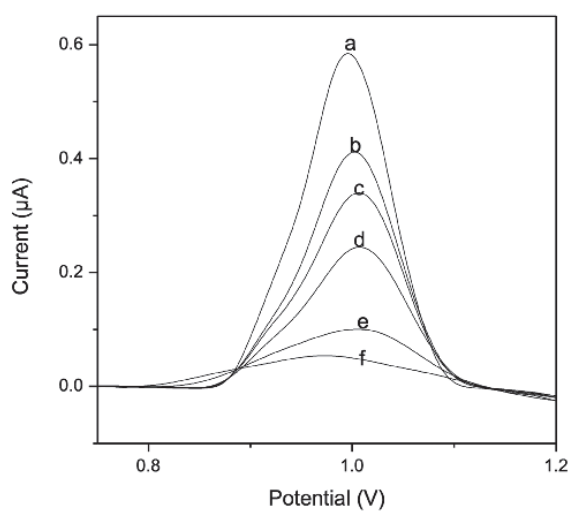


**Figure 5.** Cyclic voltammograms of 100  $\mu\text{M}$  of NIM in 0.1 M phosphate buffer solution at pH 5.0 at scan rates of 10, 25, 50, 75, 100, 150 and 200 mV/s with MWCNTs modified carbon paste electrode



pH 3.0 and 9.0 in different buffer solutions. The cyclic voltammetric measurements (Figure 3) performed for 100  $\mu\text{M}$  NIM solution exhibit that NIM has irreversible electrochemical oxidation behavior on MWCNTs modified CPE.

Effect of pH on the anodic peak current and peak potential of 100  $\mu\text{M}$  NIM were analyzed with cyclic and differential pulse voltammetry in different buffer solutions between pH 3.0 and 9.0 using MWCNTs modified CPE electrode. Due to detection responses for CV and DPV are similar, only cyclic voltammetry



**Figure 6.** Differential pulse voltammograms (a) 10, (b) 8, (c) 6, (d) 4, (e) 1, (f) 0.4  $\mu\text{M}$  of NIM solution in 0.1 M phosphate buffer solution at pH 5.0 with MWCNTs modified carbon paste electrode

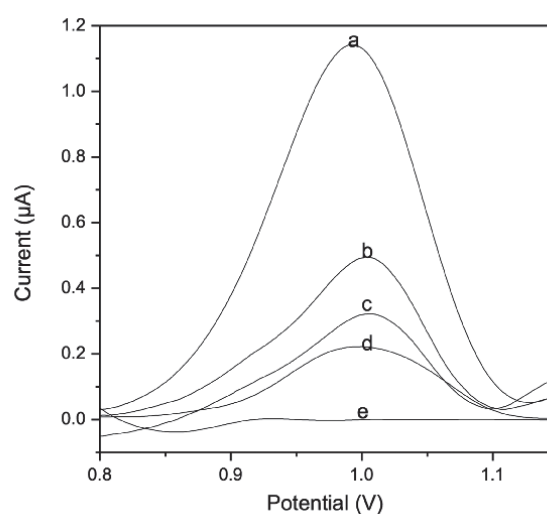
**Table 1. Validation data of calibration lines for the quantitative determination of NIM by DPV for MWCNTs modified carbon paste electrode in 0.1 M PBS at pH 5.0 and serum samples**

|   | MWCNTs modified carbon paste electrode |                     |
|---|--|---------------------|
|   | Supporting electrolyte<br>DPV          | Spiked serum<br>DPV |
| Peak potential (V)  | 1.006                                  | 1.012               |
| Linearity range ( $\mu\text{M}$ )                                 | 0.06-10                                | 0.4-40              |
| Slope ( $\mu\text{A}/\mu\text{M}^{-1}$ )                          | 89055                                  | 29531               |
| Intercept ( $\mu\text{A}$ )                                       | +0.1008                                | +0.0672             |
| Correlation coefficient   | 0.9922                                 | 0.9977              |
| Limit of detection ( $\mu\text{M}$ )                              | 0.00107                                | 0.0363              |
| Limit of quantification ( $\mu\text{M}$ )                         | 0.00324                                | 0.1101              |
| Repeatability of peak current (R.S.D.%)                           | 0.581                                  | 0.424               |
| Repeatability of peak potential (Relative standard deviation %)   | 0.418                                  | 0.255               |
| Reproducibility of peak current (Relative standard deviation %)   | 0.808                                  | 0.669               |
| Reproducibility of peak potential (Relative standard deviation %) | 0.961                                  | 0.566               |

responses of NIM were exhibited in Figure 4. As shown in Figure 4, NIM show irreversible anodic peak in the studied all pH values. The peak potential in the oxidation process of NIM shifted to less positive potentials (Figure 4A) with increasing pH. Anodic peak of NIM exhibited a pH dependent behavior between pH 3.0 and 7.0 with linear relationship (equation 1).  $E_p$  (mV) = 1298 - 64.1pH;  $r=0.992$  (between pH 3.0 and 7.0) (equation 1)

The observed pH dependence in the electro-oxidation behavior of NIM indicated that the methylsulfonamide group (electroactive group) corresponding to the NIM main oxidation peak was in acid-base equilibrium with  $\text{pK}_a$  of about 7.0. The breaking point of the curve was close to the  $\text{pK}_a$  value of NIM, at about 6.56 (20). The obtained slope value for plot of peak potential versus pH was close to theoretical value of 59 mV/pH in the Nerst equation. This corresponds to the oxidation process of NIM involves equal number of electrons and protons (21,22). The peak potential of NIM nearly was pH independent (Figures 4A), above pH 7.0. This attributed because of a change in the protonation-deprotonation process of the methylsulfonamide and the oxidation potential of NIM remains pH independent and before the electron transfer rate-determining step there are no proton transfer steps. The conjugate base must be formed by rapid dissociation of the protonated form at  $\text{pH} < \text{pK}_a$ . The plot of peak current versus pH is shown in Figure 4B. The maximum peak current and well peak shape of oxidation NIM were obtained in 0.1 M PBS at pH 5.0. Thus, electroanalytical determination of NIM and further studies were studied in 0.1 M PBS at pH 5.0.

The effect of scan rate over the range of 10.0-200.0 mV/s on the peak potential and peak current was studied by CV in PBS at pH 5.0. The peak potential of 100  $\mu\text{M}$  NIM in 0.1 M PBS at pH 5.0 is moved to the anodic direction with the scan rate increasing (Figure 5). The plot of logarithm of peak



**Figure 7.** Differential pulse voltammograms (a) 40, (b) 20, (c) 8, (d) 6  $\mu\text{M}$  of NIM solution (f) blank solution in spiked serum 0.1 M phosphate buffer solution at pH 5.0 with MWCNTs modified carbon paste electrode

current versus the logarithm of scan rate showed a linearity with a slope value of 0.341 (equation 2) and the plot of peak current versus square root scan rate exhibited a straight line with a slope value of 0.166 (equation 3). These values of the slopes were found to be close to the theoretical value of 0.5. This contributes that the electro-oxidation process of NIM is diffusion controlled process on the MWCNTs modified CPE (23). Related equations are noted below:

$$\log I_p = 0.341 \log v - 0.339 \quad r=0.991 \quad n=8 \quad (\text{equation 2})$$

$$I_p = 0.166 v^{1/2} + 0.519 \quad r=0.991 \quad n=8 \quad (\text{equation 3})$$

#### Validation of the analytical procedure

Voltammetric studies for determination of the NIM were carried out by DPV. DPV was selected due to the peaks are sharper and better determined at lower concentration of NIM than the peaks derived by CV. The anodic peak current increased linearly with increasing concentration of NIM in the DP voltammograms, as shown in Figure 6. The MWCNTs modified CPE showed linearity in the range from 0.06 and 10  $\mu\text{M}$  of NIM for DPV.

The related validation parameters for DPV and characteristics of the calibration equation are reported in Table 1. The developed DPV was validated according to standard validation procedures (24,25). Limit of detection and limit of quantification were calculated according to 3 s/m and 10 s/m, respectively, by using the standard deviation of the anodic peak response (s) and the slope value of the calibration curve (m) (26). The limit of detection value that was obtained in this study was the lowest value than the reported value in the literature for the electroanalytical determination of NIM.

We have investigated repeatability, reproducibility, precision, recovery, bias%, and selectivity for validation NIM with MWCNTs modified CPE. All validation results for NIM with MWCNTs modified CPE were repetitive, selective, reproducibility measurements, as shown in Table 1. The validation results demonstrate good precision, accuracy, repeatability and reproducibility (Table 1).

**Table 2. The results for the determination of NIM from tablet dosage forms and recovery experiments in 0.1 M PBS buffer at pH 5.0 by DPV for MWCNTs modified carbon paste electrode**

|   | Tablet (mg) |
|---|-------------|
|   | DPV         |
| Labeled claim (mg)                        | 100         |
| Amount found (mg)*                        | 100.41      |
| Relative standard deviation %             | 0.278       |
| Bias %                                    | +0.406      |
| Added (mg)                                | 33.3        |
| Found (mg)*                               | 34.16       |
| Average recovered (%)                     | 100.64      |
| Relative standard deviation % of recovery | 0.817       |
| Bias %                                    | +0.627      |

#### Determination of NIM in pharmaceutical dosage forms

The MWCNTs modified CPE was applied for the determination of NIM in Nimes® tablet dosage form. Each NIM tablet in pharmaceutical dosage form contains 100 mg NIM and inactive ingredients. The developed DPV was carried out to direct determination of NIM in pharmaceutical dosage form, using the related calibration straight line. Pretreatment such as evaporation, extraction was not required for tablet dosage form. The results obtained from the tablet dosage form are listed in Table 2. The proposed method could be successfully applied for NIM assay in tablet dosage form without any interference.

#### Determination of NIM in spiked human serum samples

The differential pulse voltammetry optimized was successfully carried out to the voltammetric determination of NIM in protein-free spiked human serum samples. Acetonitrile was used as a serum precipitating agent. No evaporation or extraction other than centrifugal protein separation at 5000 rpm was required before analyse for the drug. The calibration equation parameters and validation parameters were shown in Table 1. Obtained recovery results of human serum samples were given in Table 3.

Differential pulse voltammograms of 40 and 6  $\mu\text{M}$  of NIM obtained serum spiked were exhibited in Figure 7. As shown in Figure 7, no oxidation or noise peaks were present in

**Table 3. Results of obtained for NIM determination from spiked serum**

|   | DPV   |
|---|-------|
| Added concentration ( $\mu\text{M}$ )     | 8.00  |
| Obtained concentration ( $\mu\text{M}$ )  | 8.02  |
| Number of experiments                     | 5     |
| Average recovered (%)                     | 99.83 |
| Relative standard deviation % of recovery | 0.692 |
| Bias %                                    | +0.25 |

**Table 4. Electrochemical detection of NIM at different modified electrodes**

| Electrode  | Method | Linear range ( $\mu\text{M}$ ) | Limit of detection ( $\mu\text{M}$ ) | Reference |
|--|--------|--------------------------------|--------------------------------------|-----------|
| Glassy carbon electrode modified by cysteic acid/CNTs                  | DPV    | 0.1-10                         | 0.05                                 | 27        |
| Barium doped zinc oxide nanoparticles modified glassy carbon electrode | DPV    | 0.1-10                         | 0.0018                               | 28        |
| Gold electrode   | DPV    | 0.2-1.2                        | 0.0011                               | 8         |
| Multiwalled carbon nanotubes modified carbon paste electrode           | DPV    | 0.06-10                        | 0.00107                              | This work |

the potential range where the analytical peak was formed analytical peak and determination of NIM was successfully applied in human serum samples.

Serum samples was kept in +4°C in darkness and the stability of serum samples was studied by five consecutive analyses of the serum samples over a period of, approximately, five hours. The peak currents and peak potentials of NIM was not shown significant changes between the first and last measurements.

As it is shown in Table 4, the MWCNTs modified CPE was compared to other modified electrode in the literature according to their linear range and limit of detection. The linear range for determination of NIM with MWCNTs modified CPE at this method are better than other electrochemical methods reported in literature (27, 28, 8). When compared to limit of detection values for NIM with gold electrode (10), barium doped zinc oxide nanoparticles modified electrode glassy carbon electrode (28) and MWCNTs modified CPE (this study), this study has lowest limit of detection value.

## CONCLUSION

Carbon paste electrode was modified with multiwalled carbon nanotubes and optimized for NIM. The prepared MWCNTs modified CPE was used electroanalytical determination of NIM CV and DPV. The MWCNTs modified CPE for electroanalytical determination of NIM using DPV was carried out highly selectively, simply and stably from pharmaceutical dosage forms and human serum samples. Additionally, simplicity of the electrode preparation is very practical. Thus, multiwall carbon nanotubes modified carbon paste electrode is a practical sensor and very useful for the voltammetric determination of NIM.

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# Cytotoxic Effect of *Conyza canadensis* (L.) Cronquist on Human Lung Cancer Cell Lines

## *Conyza canadensis* (L.) Cronquist'in İnsan Akciğer Kanseri Hücre Hatları Üzerine Sitotoksik Aktivitesi

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### ABSTRACT

Asteraceae family plants are receiving great attention because they have potential anticancer activity. Therefore, in this study, the cytotoxic effect of *Conyza canadensis* (L.) Cronquist (Asteraceae) were tested against human lung adenocarcinoma cell lines (A549 and H1299) for the first time. Cytotoxic effect of the *n*-hexane, chloroform, *n*-butanol and remaining water (R-H<sub>2</sub>O) extracts fractionated from the methanol extracts of the aerial parts and roots of *C. canadensis* was investigated using Sulforhodamine B (SRB) assay and percent (%) viability was measured. The results indicate that the extracts of *C. canadensis* have cytotoxic activities on these cells in a dose-dependent manner. The root extracts exhibited relatively higher cytotoxic effects than the aerial parts of the plant. The most active extract was found to be *n*-hexane extract of the roots with IC<sub>50</sub> values 94.73 and 84.85 µg/mL on A549 and H1299 cell lines, respectively. These results suggest that *C. canadensis* exhibits moderate cytotoxic effect in lung cancer cells. This might be taken into account in its use for therapeutic purposes.

**Key words:** Asteraceae, *Conyza canadensis*, Cytotoxic activity, Lung cancer, Sulforhodamine B assay

### ÖZ

Asteraceae familyası bitkileri antikanser aktivite potansiyeline sahip olmalarından dolayı büyük ilgi görmektedir. Bundan dolayı, bu çalışmada, *Conyza canadensis* (L.) Cronquist'in (Asteraceae) insan akciğer kanseri hücre hatlarına karşı (A549 ve H1299) sitotoksik etkisi ilk kez test edilmiştir. *C. canadensis*'in kökleri ve toprak üstü kısımlarının metanol ekstresinden fraksiyonlanan *n*-hekzan, kloroform, *n*-butanol ve kalan sulu (R-H<sub>2</sub>O) ekstraktlarının sitotoksik etkisi Sülfürdamini B (SRB) yöntemiyle incelenmiş ve % canlılık oranı hesaplanmıştır. Sonuçlar *C. canadensis* ekstraktlarının kanser hücreleri üzerinde doza bağlı olarak sitotoksik aktiviteye sahip olduğunu göstermiştir. Kök ekstraktları nispeten toprak üstü kısımlarından daha yüksek aktivitede bulunmuştur. En aktif ekstre A549 ve H1299 hücre hatları üzerine IC<sub>50</sub> değerleri sırasıyla 94.73 ve 84.85 µg/mL olan köklerin *n*-hekzan ekstresi olarak tespit edilmiştir. Bu sonuçlar *C. canadensis*'in akciğer kanseri hücrelerinde orta derecede sitotoksikite gösterdiğini belirtmektedir. Bu durum bitkinin terapötik amaçlar için değerlendirilebileceğini göstermektedir.

**Anahtar kelimeler:** Asteraceae, *Conyza canadensis*, Sitotoksik aktivite, Akciğer kanseri, Sülfürdamini B testi

### INTRODUCTION

Cancer is a major group of diseases and is still among the leading of death in the world (1). Natural products are important sources of anticancer lead molecules. Many success plant-derived anticancer drugs such as paclitaxel, docetaxel, vincristin, etoposide, camptothecin, irinotecan are clinically available; however due to the rapid development of resistance to chemotherapeutic drugs and their side effects,

novel anticancer drugs of natural sources are under an extensive search for cancer therapy (2,3).

The family Asteraceae which contains over 1600 genera and more than 23000 species is the largest family with rich chemical constituents (4). It has a broad medicinal utilization worldwide in which over 300 species of the family are known to have ethnomedicinal uses for cancer related diseases (5). The number of studies related to the cytotoxicity of Asteraceae plants and their secondary metabolites have

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intensively investigated in recent years. In these studies, sesquiterpene lactones and flavonoids have been principally stated to be responsible compounds for the cytotoxic effects (6-10). These compounds induce apoptosis after disrupting cell cycle of cancer cells *in vitro* and *in vivo* and also inhibit angiogenesis and metastasis (11,12).

The genus *Conyza* Less. belongs to the family Asteraceae and consists of about fifty species all over the world. In Turkey, the genus *Conyza* is represented by three species in the flora of Turkey, namely *C. canadensis* (L.) Cronquist, *C. bonariensis* (L.) Cronquist, and *C. albida* Willd. ex. Sprengel (13,14). *C. canadensis* (syn. *Erigeron canadensis* L.), known as "Canadian fleabane" or "horseweed", is native throughout of North America and is also widespread in Europe. It is an annual plant erecting 10 to 180 cm high, with sparsely hairy stems (15). *C. canadensis* was reported to be used for its diuretic, antibacterial, anti-inflammatory, tonic, astringent, antihemorrhagic properties as well as for the treatment of diarrhea and dysentery in folk medicines (16-18). In addition, a decoction of the plant is used anticancer purposes in North America (19).

The phytochemical studies on *C. canadensis* have so far pointed out to presence of terpenes, acetylene derivatives, flavonoids, benzoic acid derivatives, alkaloids, essential oils, sphingolipids, fatty acids and sterols (20-26). Among them, C<sub>10</sub> acetylenes such as diyn-ene (e.g. *E*-lactonophyllum methyl ester) and ene-diyn-ene (e.g. matricaria methyl ester isomers), and C<sub>10</sub> lactones (e.g. 8*Z*-matricaria- $\gamma$ -lactone) are typical constituents of the genus (20,26).

*C. canadensis* have been demonstrated to exert several biological activities such as cytotoxic, antifungal, antibacterial, antiviral, anti-inflammatory, antioxidant, and antiagregant (26-33). In previous studies, the extracts from *C. canadensis* were reported to have prominent cytotoxic effects on various cancer cell lines (26,34-37). To the best of our knowledge, there is no scientific report available in support of the cytotoxic effect of *C. canadensis* on A549 and H1299 human lung cancer cells. The aim of the present study was to investigate the possible *in vitro* cytotoxic effect of the extracts from the aerial parts and roots of *C. canadensis* in human lung adenocarcinoma cell lines (A549 and H1299).

## EXPERIMENTAL

### Chemicals

In the extraction procedure, methanol, *n*-hexane, chloroform and *n*-butanol were of analytical grade and were purchased from Merck Co. (Darmstadt, Germany). Analytical thin layer chromatography (TLC) was performed on precoated Kieselgel 60 F<sub>254</sub> plates (Art. 5554, Merck). The plates sprayed with anisaldehyde reagent [76% methanol (Merck) and 19% ortho-phosphoric acid (Riedel-De Haën, Buchs, SG Switzerland), 5% *p*-anisaldehyde (Merck)], 30% H<sub>2</sub>SO<sub>4</sub> (Merck) solution in MeOH (Merck) and 1% vanillin-H<sub>2</sub>SO<sub>4</sub> solution [vanillin

(Boehringer Mannheim, Mannheim, Germany) and H<sub>2</sub>SO<sub>4</sub> (Merck)].

### Plant material

*Conyza canadensis* (L.) Cronquist was collected from Balcova, Izmir, Turkey, in the flowering-fruit stage, in November 2013. The plant was identified by Prof. Dr. Mecit Vural from the Department of Botany, Faculty of Science, Gazi University. A voucher specimen (F. Ayaz 29) has been deposited at Herbarium of Gazi University (GAZI), Ankara, Turkey.

### Preparation of extracts

39.52 g powdered aerial parts (CCH) and 40.25 g roots (CCR) of the plant were extracted with 80% methanol by stirring at 40°C for 6 h three times (3×300 mL). Following filtration, the combined methanol extracts were evaporated *in vacuo* at 40°C to dryness. The concentrated MeOH extracts (100 mL) were further fractionated by successive solvent extractions with *n*-hexane (3×100 mL), chloroform (3×100 mL) and *n*-butanol saturated with H<sub>2</sub>O (3×100 mL) in a separatory funnel. Each extract as well as remaining aqueous phase (R-H<sub>2</sub>O) after solvent extractions was evaporated to dryness under reduced pressure to yield "*n*-Hexane extract" (0.04 g for CCH, 0.29 g for CCR), "CHCl<sub>3</sub> extract" (0.14 g for CCH, 0.30 g for CCR), "*n*-BuOH extract" (2.63 g for CCH, 1.86 g for CCR) and "R-H<sub>2</sub>O extract" (1.03 g for CCH, 2.83 for CCR), respectively.

### Phytochemical analysis

1 mg/mL of *C. canadensis* extracts were applied to silica gel plates. The *n*-hexane and CHCl<sub>3</sub> extracts were developed with the mixture of *n*-hexane:acetone (7:3) as a mobile phase. TLC plates were evaluated under UV light at 254 and 366 nm for the determination of fluorescent compounds. Anisaldehyde reagent and 30% H<sub>2</sub>SO<sub>4</sub> were sprayed to the plates to visualize the separated compounds and then plates were heated for 5 min at 100°C. Terpenes were appeared in pink, purple and green coloration with anisaldehyde reagent. In addition, terpenes also showed red coloration under UV 254 nm sprayed with 30% H<sub>2</sub>SO<sub>4</sub> after heating for 5 min at 100°C. The *n*-butanol and R-H<sub>2</sub>O extracts were developed in a mixture of solvent system CHCl<sub>3</sub>:MeOH:H<sub>2</sub>O (61:32:7) and then sprayed with 1% vanillin-H<sub>2</sub>SO<sub>4</sub> solution. The plates were heated for 5 min at 100°C before examined under UV light. Flavonoids detected as yellow and orange zones on the plates (38).

### Cell culture

A549 and H1299 human lung adenocarcinoma cells were kindly provided by Prof. Hakan Akça (Pamukkale University, Faculty of Medicine, Denizli, Turkey). A549 and H1299 cells were cultured in RPMI 1640 medium supplemented with penicillin G (100 U/mL), streptomycin (100 mg/mL), L-glutamine, and 10% fetal bovine serum at 37°C in a humidified atmosphere containing 5% CO<sub>2</sub>.

### Cytotoxicity assay

For the Sulforhodamine B (SRB) assay, which is routinely performed by National Cancer Institute (NCI) for *in vitro* drug screening, *n*-hexane, chloroform, *n*-butanol and remaining

water extracts were added to 96-well plates to make up a final concentration range of 1.56  $\mu\text{g/mL}$  to 100  $\mu\text{g/mL}$  by serial dilutions (six two-fold dilutions). Then, A549 and H1299 cells were seeded at a density of  $5 \times 10^3$  cells per well of 96-well plates. Subsequently, cells were incubated with various concentrations of the aerial parts and roots extracts for 48 h. The assay was terminated by the addition of ice-cold 50% (w/v) trichloroacetic acid. SRB 0.4% (w/v) in 1% (v/v) acetic acid staining was then performed. The bound dye was extracted using 10 mM unbuffered Tris and optical density was measured at 564 nm with an ELISA plate reader (FLASH Scan S12, Analytik Jena, Germany). Viability of treated cells was calculated in reference to the untreated control cells by using the following formula:

$$\text{Cell viability (\%)} = [100 \times (\text{Sample Abs}) / (\text{Control Abs})].$$

## RESULTS AND DISCUSSION

The cytotoxic effect of the extracts from the aerial parts and roots of *C. canadensis* on lung cancer cell lines (A549 and H1299) were first investigated by the SRB assay after treating cells with increasing doses of extracts (1.56  $\mu\text{g/mL}$ -100  $\mu\text{g/mL}$ ) for 48 h. It was found that the extracts inhibited

growth of cells in a dose-dependent manner and prominently reduced the cell viability at the 100  $\mu\text{g/mL}$ . The cytotoxic effects after the treatment with the extracts against human lung adenocarcinoma cell lines were shown in Figures 1a, 1b, 2a and 2b.

In order to compare antigrowth effects of various concentrations of the aerial parts and roots extracts of *C. canadensis*, the root extracts exhibited relatively higher antigrowth effects than the aerial parts extracts. The strongest cytotoxic activity was detected for the *n*-hexane extract of the roots with  $\text{IC}_{50}$  values 94.73 and 84.85  $\mu\text{g/mL}$  on A549 and H1299 cell lines, respectively. Overall, the antigrowth effects of the extracts were not dependent on the cell line. Considering the fact that A549 expresses wild type p53 and H1299 is p53 null, it can be stated that the resulting antigrowth effects of the extracts are p53 independent (39). This may actually be favorable because majority of cancers have mutated p53, thereby these extracts would still be active against even p53-mutated ones.

To the best of our knowledge, this is the first attempt rationalizing the cytotoxic effect of *C. canadensis* on A549 and H1299 human lung cancer cells. Nevertheless, a few

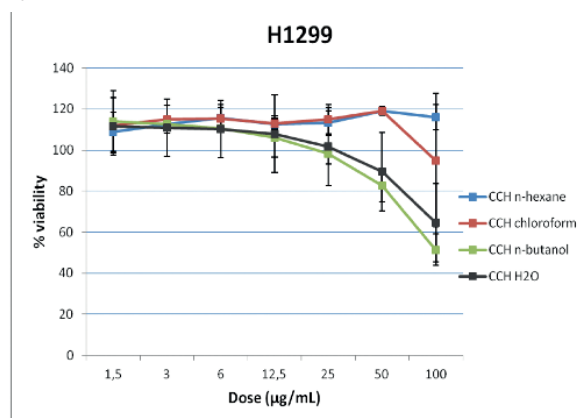


Figure 1a. The cytotoxic effects with different concentrations of the aerial parts extracts of *C. canadensis* on H1299 cell lines

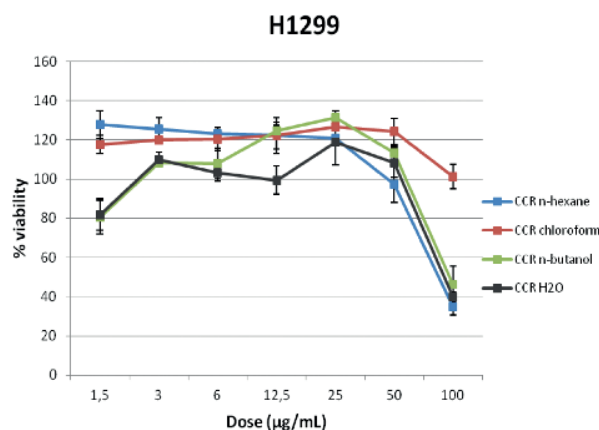


Figure 2a. The cytotoxic effects with different concentrations of the root extracts of *C. canadensis* on H1299 cell lines

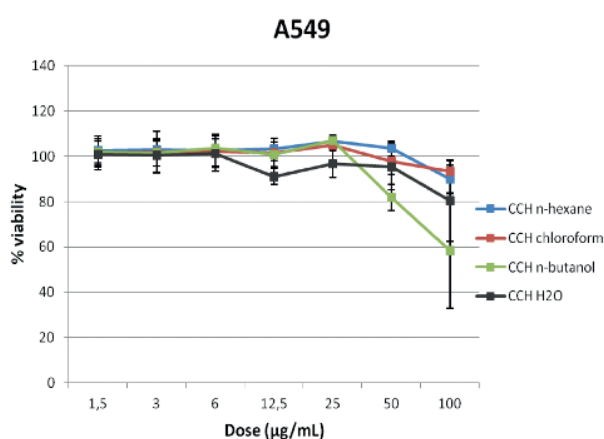


Figure 1b. The cytotoxic effects with different concentrations of the aerial parts extracts of *C. canadensis* on A549 cell lines

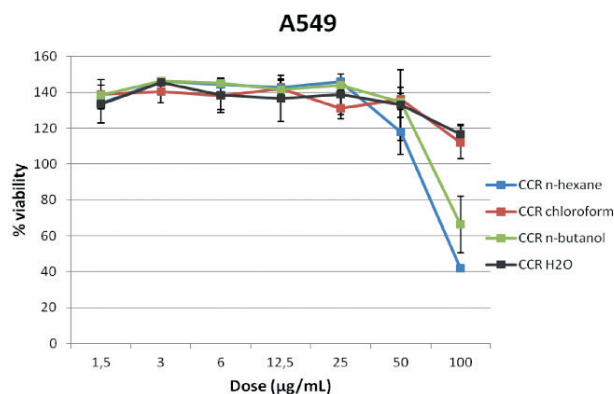


Figure 2b. The cytotoxic effects with different concentrations of the root extracts of *C. canadensis* on A549 cell lines

studies demonstrated earlier cytotoxic effect of the extracts of *C. canadensis* of different origin having varying levels of inhibition on various cancer cell lines (26,34-37). For instance; *n*-hexane, chloroform and aqueous MeOH extracts partitioned from MeOH extract of the aerial parts, flowers and roots of *C. canadensis* and the H<sub>2</sub>O extracts prepared from the residual plant materials were investigated for their cytotoxic properties on HeLa, MCF-7 and A431 cell lines using the MTT assay. The *n*-hexane phase of the roots exhibited markedly antigrowth effects on the cell lines (62.4-70.1%) at 10 µg/mL, and the CHCl<sub>3</sub> phase of the roots demonstrated moderate antiproliferative activity (39.3-47.9%) at the same concentration (34). According to the bioactivity-guided fractionation of the *n*-hexane and chloroform phases of the methanol extract from the roots of *C. canadensis*, two new unusual C<sub>10</sub> γ-dihydropyranone derivatives (conyzapyranone A and conyzapyranone B), as well as 2 γ-lactone acetylene derivatives (e.g. 4E,8Z-matricaria-γ-lactone), triterpenes, sterols (e.g. spinasterol), a hydroxy fatty acid and a flavonoid were isolated. Among them, conyzapyranone B, 4E,8Z-matricaria-γ-lactone and spinasterol were found to have remarkable antiproliferative activity against HeLa, MCF-7 and A431 cell lines (26). In other study, cytotoxic activities of petroleum ether, ethyl acetate and methanol extracts of the aerial parts of *C. canadensis* were investigated on Hep-2 using methylene blue assay at 24, 48 and 72 h of incubation. At 72 h of incubation, the most active extracts were found to be ethyl acetate and petroleum ether extracts with IC<sub>50</sub> values 45 and 50 µg/mL, respectively (36). In another study, erigeronol, a new triterpene derivative, was isolated from *C. canadensis* as a potent cytotoxic compound with IC<sub>50</sub> value of 7.77±0.47 µg/mL on melanoma B16 cell line by the MTT method (37). In these studies, triterpenes, C<sub>10</sub> acetylene derivatives and dihydropyranones have mainly found as effective cytotoxic constituents in *C. canadensis* (26,37). In our study, terpenes and flavonoids were principally detected in the root extracts according to the preliminary phytochemical analysis.

The present investigation represents a preliminary screen for the cytotoxic effect of *C. canadensis* in human lung cancer cell lines. In accordance with the National Cancer Institute Guidelines, extracts with IC<sub>50</sub> values <20 µg/mL were accepted as active (40). This study resulted in moderate cytotoxic activity compared to the previous studies against the selected cell lines which might be attributed to usage of different cell lines as well as the diverse phytochemical composition in the extracts. The present results are also in accordance with ethnomedicinal uses of the plant reported by Hartwell (Hartwell, 1968).

In conclusion, we provided the first evidence for cytotoxic effects of *C. canadensis* against A549 and H1299 cancer cell lines although it shows the cytotoxic activity at relatively higher doses.

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# Antioxidant and Anticholinesterase Activities of Essential Oil of *Alseodaphne peduncularis* Meisn

## *Alseodaphne peduncularis* Meisn. Uçucu Yağının Antioksidan ve Antikolinesteraz Aktivitesi

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### ABSTRACT

This study was designed to investigate the antioxidant and anticholinesterase activities of the essential oils from *Alseodaphne peduncularis* Meisn. GC and GC/MS analysis of the leaves oil showed thirty one components representing 72.6% of the oil. The most abundant components were  $\beta$ -caryophyllene (24.0%),  $\delta$ -cadinene (15.9%) and germacrene B (12.2%). The antioxidant activity was determined by DPPH (1,1-diphenyl-2-picrylhydrazyl) radical scavenging and total phenolic content assays, while anticholinesterase activity assessed by measuring inhibition of acetylcholinesterase (AChE) and butyrylcholinesterase (BChE) enzymes. The essential oil showed weak activity in the DPPH radical scavenging (IC<sub>50</sub> of 253.2  $\mu$ g/mL) and phenolic content (32.5 mg GA/g), while moderate inhibition activity against AChE (I: 45.2%) and BChE (I: 48.6%).

**Key words:** Essential oil, *Alseodaphne peduncularis*, Antioxidant activity, Anticholinesterase activity, Lauraceae

### ÖZ

Bu çalışma *Alseodaphne peduncularis* Meisn. uçucu yağının antioksidan ve antikolinesteraz aktivitelerini araştırmak için tasarlanmıştır. GC ve GC/MS analizleri yaprak yağında %72,6 oranında otuz bir bileşen bulunduğunu göstermiştir. En çok bulunan bileşenler  $\beta$ -karyofilen (%24,0),  $\delta$ -kadinen (15,9) ve germakren B (%12,2) dir. Antioksidan aktivite DPPH (1,1- difenil-2-pikrilhidrazil) radikal süpürme ve toplam fenolik içerik yöntemleriyle değerlendirilirken, antikolinesteraz aktivite asetilkolinesteraz (AChE) ve bütirikolinesteraz (BChE) enzim inhibisyonu ölçülerek değerlendirilmiştir. Uçucu yağ, zayıf DPPH radikal süpürme aktivitesi (IC<sub>50</sub> değeri 253,2 mg/mL) ve fenolik içerik (32,5 mg GA/g) gösterirken, AChE (I: %45,2) ve BChE (I: %48,6) inhibisyon aktivitesi orta düzeydedir.

**Anahtar kelimeler:** Uçucu yağ, *Alseodaphne peduncularis*, Antioksidan aktivite, Antikolinesteraz aktivite, Lauraceae

### INTRODUCTION

*Alseodaphne* is a genus of small to medium sized trees of the wet evergreen tropical forests. The genus is having more than fifty species and distributed in tropical belt of Cambodia, China, Indonesia, Laos, Malaysia, Myanmar, Philippines, Sri Lanka, Thailand, Vietnam and India. There is no information in the literature regarding medicinal uses of genus *Alseodaphne*. Most of the species of genus are unexplored, both, pharmacologically and phytochemically (1). Previous phytochemical investigations of *Alseodaphne* have resulted in the isolation of alkaloids (aporphines, bisbenzylisoquinoline, morphinandienones) (2), lactones (3), neolignans (4) and phenanthrenes (5). The isolated constituents of genus *Alseodaphne* might become useful as a source of pharmacologically interesting molecules.

*A. peduncularis* is a shrub or sometimes a small tree of up to 6-12 m height. The twigs colour is whitish. The leaves/stalk is slender and around 0.51 cm long in size. The blade is membranous and lanceolate with drying greenish brown colour. The midrib rise above and secondary nerves is about 610 pairs, curving and joining near margin and rise on both surfaces. The tertiary nerves are reticulate and visible on both the surfaces. The colour of the flowers is greenish. The shape of fruits is ellipsoid or globose with a dark purple colour and on enlarged red perianth tube (1).

To the best of our knowledge there is no report on the chemical compositions of the essential oil of *A. peduncularis*, therefore we would like to report their chemical composition and their antioxidant and anticholinesterase activities.

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## MATERIALS AND METHODS

### *Plant materials*

A sample of *A. peduncularis* was collected from Hutan Simpan Bangi, Selangor in September 2015, and identified by Dr. Shamsul Khamis from Institute of Bioscience (IBS), Universiti Putra Malaysia (UPM). The voucher specimen (SK2955/16) were deposited at the Herbarium of IBS, UPM.

### *Solvents and chemicals*

Antioxidant: 1,1-Diphenyl-2-picrylhydrazyl (DPPH), gallic acid and butylatedhydroxytoluene (BHT) were obtained from Sigma-Aldrich (Germany). Analytical grade methanol, ethanol and dimethylsulfoxide (DMSO), HPLC grade chloroform, Folin-Ciocalteu's reagent, anhydrous sodium sulphate, and sodium carbonate were purchased from Merck (Germany). Anticholinesterase: AChE enzyme (Type-VI-S, EC3.1.1.7), butyrylcholinesterase enzyme (BChE; EC3.1.1.8), acetylthiocholine iodide, butyrylthiocholine chloride, 5,5'-dithio-bis(2-nitrobenzoic)acid (DTNB), and galantamine were purchased from Sigma-Aldrich (Germany).

### *Extraction of essential oils*

The fresh leaf (300 g) was subjected to hydrodistillation in an all glass Dean-stark apparatus for 6 hours. The oils obtained were dried over anhydrous magnesium sulfate and stored at 4-6°C. The oil yield (w/w) was 1.55 g (0.52%) based on the fresh weight.

### *Gas chromatography (GC)*

GC analysis were performed on a Hewlett Packard 6890 series II A gas chromatograph equipped with an Ultra-1 column (100% polymethylsiloxanes) (25 m long, 0.33 µm thickness and 0.20 mm inner diameter). Helium was used as a carrier gas at flow rate of 0.7 mL/min. Injector and detector temperature were set at 250 and 280°C, respectively. Oven temperature was kept at 50°C, then gradually raised to 280°C at 5°C/min and finally held isothermally for 15 min. Diluted samples (1/100 in diethyl ether, v/v) of 1.0 µL were injected manually (split ratio 50:1). The injection was repeated three times and the peak area percents were reported as means ± SD of triplicates. Calculation of peak area percentage was carried out by using the GC HP Chemstation software (Agilent Technologies).

### *Gas chromatography-mass spectrometry (GC-MS)*

GC-MS chromatograms were recorded using a Hewlett Packard Model 5890A gas chromatography and a Hewlett Packard Model 5989A mass spectrometer. The GC was equipped with Ultra-1 column (25 m long, 0.33 µm thickness and 0.20 mm inner diameter). Helium was used as a carrier gas at a flow rate of 1 mL/min. Injector temperature was 250°C. Oven temperature was programmed from 50°C (5 min hold) at 10°C/min to 250°C and finally held isothermally for 15 min. For GC-MS detection, an electron ionization system, with ionization energy of 70 eV was used. A scan rate of 0.5

s (cycle time: 0.2 s) was applied, covering a mass range from 50-400 amu.

### *Identification of components*

The constituents of the oil were identified by comparison of their mass spectra with reference spectra in the computer library (Wiley) and also by comparing their retention indices, with data in the literature (6). The quantitative data were obtained electronically from FID area percentage without the use of correction factor.

### *Antioxidant activity*

#### *DPPH radical scavenging*

The free radical scavenging activity was measured by the DPPH method with minor modifications (7). Each sample of stock solution (1.0 mg/mL) was diluted to final concentration of 1000-7.8 µg/mL. Then, a total of 3.8 mL of 50 µM DPPH methanolic solution (1 mg/50 mL) was added to 0.2 mL of each sample solution and allowed to react at room temperature for 30 min. The absorbance of the mixtures was measured at 517 nm. A control was prepared without sample or standard and measured immediately at 0 min. Lower absorbance of the reaction mixture indicates higher free radical scavenging activity, and vice versa. The percent inhibitions (I%) of DPPH radical were calculated as follow:

$$I\% = [A_{\text{blank}} - A_{\text{sample}} / A_{\text{blank}}] \times 100$$

where  $A_{\text{blank}}$  is the absorbance value of the control reaction and  $A_{\text{sample}}$  is the absorbance values of the test samples. The sample concentration providing 50% inhibition ( $IC_{50}$ ) was calculated by plotting inhibition percentages against concentrations of the sample. All tests were carried out in triplicate and  $IC_{50}$  values were reported as means ± SD of triplicates.

#### *Total phenolic content (TPC)*

Total phenolic contents of the essential oils were determined as described previously (8). A sample of stock solution (1.0 mg/mL) was diluted in methanol to final concentrations of 1000 µg/mL. A 0.1 mL aliquot of sample was pipetted into a test tube containing 0.9 mL of methanol, then 0.05 mL Folin-Ciocalteu's reagent was added, and the flask was thoroughly shaken. After 3 min, 0.5 mL of 5%  $Na_2CO_3$  solution was added and the mixture was allowed to stand for 2 h with intermittent shaking. Then, 2.5 mL of methanol was added and left to stand in the dark for 1 h. The absorbance measurements were recorded at 765 nm. The same procedure was repeated for the standard gallic acid solutions. The concentration of total phenolic compounds in the oils was expressed as mg of gallic acid equivalent per gram of sample. Tests were carried out in triplicate and the gallic acid equivalent value was reported as mean ± SD of triplicate.

#### *Anticholinesterase activity*

AChE/BChE inhibitory activity of the essential oils was measured by slightly modifying the spectrophotometric

**Table 1. Constituents identified from the essential oil of *A. peduncularis***

| No                                | Components               | KI <sup>a</sup> | Percentage (%) |
|-----------------------------------|--------------------------|-----------------|----------------|
| 1                                 | δ-Elemene                | 1335            | 1.5            |
| 2                                 | α-Cubebene               | 1345            | 0.5            |
| 3                                 | α-Ylangene               | 1373            | 0.1            |
| 4                                 | α-Copaene                | 1374            | 2.5            |
| 5                                 | β-Copaene                | 1374            | 0.8            |
| 6                                 | Isodene                  | 1375            | 0.3            |
| 7                                 | β-Cubebene               | 1387            | 0.4            |
| 8                                 | α-cis-Bergamotene        | 1411            | 0.4            |
| 9                                 | <b>β-Caryophyllene</b>   | 1417            | <b>24.0</b>    |
| 10                                | α-trans-Bergamotene      | 1432            | 0.1            |
| 11                                | γ-Elemene                | 1434            | 5.2            |
| 12                                | α-Guaiene                | 1437            | 0.0            |
| 13                                | 6,9-Guaiadiene           | 1442            | 0.6            |
| 14                                | cis-Cadina-1(6)4-diene   | 1461            | 0.2            |
| 15                                | trans-Cadina-1(6)4-diene | 1475            | 0.4            |
| 16                                | γ-Murolene               | 1478            | 0.9            |
| 17                                | β-Selinene               | 1489            | 0.6            |
| 18                                | β-Guaiene                | 1492            | 0.1            |
| 19                                | γ-Amorphene              | 1495            | 0.3            |
| 20                                | Valencene                | 1496            | 0.3            |
| 21                                | Cubebol                  | 1514            | 0.3            |
| 22                                | <b>δ-Cadinene</b>        | 1522            | <b>15.9</b>    |
| 23                                | α-Cadinene               | 1537            | 0.1            |
| 24                                | γ-Cadinene               | 1543            | 0.2            |
| 25                                | Germacrene B             | 1559            | <b>12.2</b>    |
| 26                                | (E)-Nerolidol            | 1561            | 0.1            |
| 27                                | Spathulenol              | 1577            | 0.2            |
| 28                                | Caryophyllene oxide      | 1582            | 0.8            |
| 29                                | Globulol                 | 1590            | 0.2            |
| 30                                | Epicubenol               | 1627            | 1.0            |
| 31                                | trans-Longipinocarveol   | 1634            | 0.2            |
| 32                                | t-Murolol                | 1644            | 1.9            |
| 33                                | β-Eudesmol               | 1649            | 0.1            |
| 34                                | α-Bisabolol              | 1685            | 0.1            |
| 35                                | Phytol                   | 1942            | 0.1            |
| <b>Group components</b>           |                          |                 |                |
| <b>Sesquiterpene hydrocarbons</b> |                          |                 | 67.6           |
| <b>Oxygenated sesquiterpenes</b>  |                          |                 | 5.0            |
| <b>Identified components (%)</b>  |                          |                 | <b>72.6</b>    |

<sup>a</sup>Kovat's indices (KI) experimental: n-alkanes (C<sub>9</sub>-C<sub>30</sub>) were used as reference points in the calculation of KI

method developed by Ellman et al. (9) and Orhan et al. (10). Electric eel (*Electrophorus electricus*) AChE and horse serum BChE were used, while acetylthiocholine iodide (AChI) and butyrylthiocholine chloride (BChI) were employed as substrates of the reaction. DTNB acid was used for the measurement of the anticholinesterase activity. Briefly, in this method, 140 µL of sodium phosphate buffer (pH 8.0), 20 µL of DTNB, 20 µL of essential oils and 20 µL of AChE/BChE solution were added by multichannel automatic pipette in a 96-well microplate and incubated for 15 min at 25°C. The reaction was then initiated with the addition of 10 µL of AChI/BChI. Hydrolysis of AChI/BChI was monitored by the formation of the yellow 5-thio-2-nitrobenzoate anion as a result of the reaction of DTNB with thiocholines, catalyzed by enzymes at 412 nm utilizing a 96-well microplate reader (Epoch Micro-Volume Spectrophotometer). Percentage inhibition (I%) of AChE/BChE was determined by comparison of rates of reaction of samples relative to blank sample (EtOH in phosphate buffer pH 8) using the formula:

$$I\% = [E - S/E] \times 100$$

where E is the activity of enzyme without test sample and S is the activity of enzyme with test sample. The experiments were done in triplicate. Galantamine was used as reference.

#### Statistical analysis

Data obtained from essential oil analysis and their bioactivities were expressed as mean values. The statistical analyses were carried out by employing one way ANOVA ( $p < 0.05$ ). A statistical package (SPSS version 11.0) was used for the data analysis.

## RESULTS AND DISCUSSION

Hydrodistillation of the fresh leaves of *A. peduncularis* gave pale yellow oil in 0.52% (w/w). The chemical compositions of the leaves oil of *A. peduncularis* are listed in Table 1. GC and GC-MS analysis of the essential oil had successfully found thirty five components, which accounted for 72.6% of the chromatographical components. Sesquiterpene hydrocarbons were the major components in the essential oil (67.6%) with β-caryophyllene (24.0%) being the most substantial component, followed by δ-cadinene (15.9%) and germacrene B (12.2%). Furthermore, oxygenated sesquiterpenes made

**Table 2. Antioxidant and anticholinesterase activities of the essential oil of *A. peduncularis***

| Samples       | DPPH IC <sub>50</sub> (µg/mL) <sup>a</sup> | TPC (mg GA/g) <sup>b</sup> | AChE (I%) <sup>c</sup> | BChE (I%) <sup>c</sup> |
|---------------|--|----------------------------|------------------------|------------------------|
| Essential oil | 253.2                                      | 32.5 ± 0.1                 | 45.2 ± 0.2             | 48.6 ± 0.2             |
| BHT           | 18.5                                       | -                          | -                      | -                      |
| Galantamine   | -  | -                          | 95.9 ± 0.2             | 88.7 ± 0.2             |

<sup>a</sup>IC<sub>50</sub> value at concentrations of 1000-7.8 µg/mL; <sup>b</sup>TPC at a concentration of 1 mg/mL; <sup>c</sup>Percentage inhibition at a concentration of 1 mg/mL; ± represents SD of three independent experiments ( $p < 0.05$ )

up a minor fraction which gave 5.0%, while monoterpenoids were not found in this essential oil.

The antioxidant activities were investigated by DPPH free radical scavenging assay together with the Folin-Ciocalteu assay which evaluated the total phenolic content of the essential oil. The results are displayed in Table 2. The antioxidant activity on DPPH radical scavenging is due to their hydrogen donating ability. The capability of substances to donate hydrogen is able to convert DPPH into their non-radical form DPPH and the reaction can be followed spectrophotometrically (11). The essential oil exhibited weak DPPH radical scavenging activity ( $IC_{50}$  of 253.2  $\mu\text{g/mL}$ ) compared to standard antioxidant, BHT ( $IC_{50}$  of 18.5  $\mu\text{g/mL}$ ). The low activity was attributed to the low phenolic content of the essential oil which is responsible for antioxidant activity. This was supported by the results of the Folin-Ciocalteu assay on the essential oils, which showed low amount of phenolic (32.5%). AChE plays an important role in the central nervous system. It is one of the fastest known enzymes and catalyses the cleavage of acetylcholine in the synaptic cleft after depolarisation. Inhibitors of AChE, such as galanthamine, are used frequently in the pharmacotherapy of Alzheimer disease (12). The essential oil indicated moderate AChE and BChE activity at 1000  $\mu\text{g/mL}$  concentration, which gave 45.2% and 48.6% for AChE and BChE activity, respectively. In previous reports, AChE inhibition can be explained by the high content of monoterpenes. It has been mentioned that 1,8-cineole, camphor,  $\alpha$ -pinene,  $\beta$ -pinene, borneol, linalool, menthone, carvone, anetole, anisole, have anticholinesterase activity (13-15). This current oil lacked the presence of monoterpenes, hence contributed to the low AChE/BChE inhibition.

## CONCLUSION

In conclusion, essential oil and their components generally displayed significant bioactivity properties, which are useful as preventive agents from various diseases. In the case of *A. peduncularis* oil, although there is no striking on their bioactivity in the oil, it is still worthwhile to investigate the other parts of the plant as a natural source for essential oil composition or their phytochemical studies.

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# Nano Chemotherapeutics in Bone Metastasis and Targeting

## Kemik Metastazlarında Nano Kemoterapötikler ve Hedeflendirilmeleri

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### ABSTRACT

Due to the increase of cancer incidence in recent years, contemporary studies have focused on the identification, diagnosis and treatment of this disease. It is rather difficult to treat cancer without harming normal cells. Thus, studies exploring nano systems that are targeted to the cancerous cells are drawing attention. The frequent formation of metastasis in the bone tissue of certain cancers, such as the breast and prostate, has increased the mortality risk from these cancers. For this reason, preventing metastasis by targeting metastatic bone tissue with anti-neoplastic agents without damaging healthy soft tissue has become a subject of interest for many researchers. In addition to the therapeutic properties of bisphosphonates for bone metastases that form osteoclastic and osteoblastic, its affinity to hydroxyapatite, the main mineral of bone tissue, has been extensively studied as a targeting agent for bone diseases. With the conjugation of bisphosphonates to the nanoparticulate systems, nanoparticles loaded with the active ingredient are transmitted solely to the diseased bone tissue. Then, with the help of a nanostructured system, the bisphosphonates are localized only in the tumor containing diseased bone tissue, and the chemotherapeutic agent is released in a controlled manner. In this review, current approaches targeting bone metastases, and especially studies conducted with nanostructured drug carrier systems, are assessed.

**Key words:** Nano chemotherapeutics, Bone metastasis, Bisphosphonates, Nanoparticles, Drug targeting

### ÖZ

Kanser hastalığının son yıllarda artan insidansı nedeniyle güncel çalışmalar bu hastalığın tanı, teşhis ve tedavisi üzerine yoğunlaşmıştır. Normal hücrelere zarar vermeden hastalığın tedavisi oldukça güçtür. Bu nedenle sadece kanserli hücreye hedeflendirilmiş nano sistemler ile yapılan çalışmalar dikkat çekmektedir. Yaygın olan meme ve prostat kanserleri gibi hastalıkların metastazının sıklıkla kemik dokusunda meydana gelmesi hastanın yaşam kalitesini azaltmasından öte bu kanser türlerinin ölüm riskini artırmaktadır. Bu amaçla sağlıklı yumuşak dokuya zarar vermeden anti-neoplastik ajanların doğrudan metastaz meydana gelmiş kemik dokusuna hedeflendirilmesi ile metastazın önlenmesi veya tedavisi çoğu araştırmacının üzerinde çalıştığı konular arasına girmiştir. Osteoklastik veya osteoblastik şekilde oluşan kemik metastazları için bifosfonatların tedavi edici özelliklerine ek olarak kemik dokusunun ana minerali olan hidroksiapatite olan afinitesi nedeni ile kemik hastalıkları için hedefleme ajanı olarak da oldukça yaygın şekilde çalışma konusu olmuştur. Bifosfonatların nanopartiküler sistemlere konjugasyonu ile de etkin madde yüklü nanopartiküller sadece hasta kemik dokusuna iletilmekte, nano yapı sayesinde de sistemin tümörlü dokuda lokalize olup kemoterapötik ajanı kontrollü olarak salması sağlanmaktadır. Bu derleme ile kemik metastazlarına hedeflemedeki güncel olan yaklaşımlar, özellikle nano yapıli ilaç taşıyıcı sistemler ile yapılan çalışmalar tartışılarak değerlendirilmiştir.

**Anahtar kelimeler:** Nano kemoterapötikler, Kemik metastazı, Bifosfonatlar, Nanopartiküller, İlaç hedefleme

### INTRODUCTION

Recent developments in nanotechnology have advanced the direction of biomedical applications and the optimization of therapies. Nano chemotherapeutics have continued to be the center of interest due to several distinguishing attributes, including: the surface area of their delivery systems, which is greater than a micrometer; several unique structural properties; high residency time in the circulatory system; configurability for the diagnosis and treatment of many diseases; ability to target various bioactive agents and organs;

ability to travel in the tissue through capillaries; and their relatively easy uptake by the cells. Especially the production of nanoparticles that can deliver chemotherapeutic agents to cancerous tissues and thus, blocking the inadequate selectivity of conventional cancer drugs are very promising (1-3). Aside from their targeting ability, another property that attracts attention of nanoparticles is their ability to facilitate the delivery of different chemotherapeutic agents together, thereby increasing the efficiency of the treatment (4).

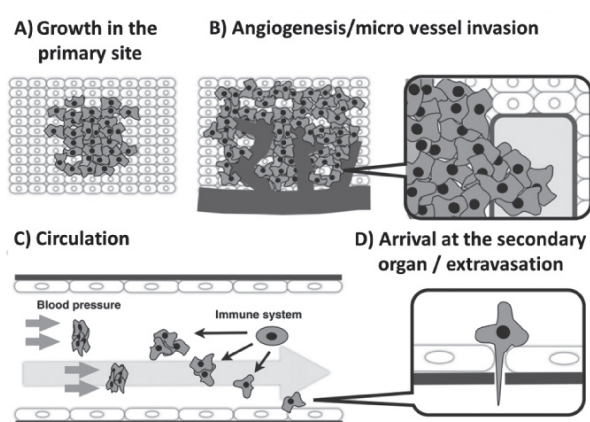
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Bone diseases are disorders related to the skeletal system and include disorders that restrict basic movements. These disorders disrupt the balance between bone resorption and formation (5). Since there is no effective treatment for multiple bone disorders, such as arthritis, osteoarthritis, osteosarcoma, and metastatic bone cancer, new drugs and new drug delivery systems that performed efficiently and safely in clinical treatments must be developed immediately. However, for most of bone diseases that cause bone disturbances, such as bone cancer, the main concern is maintaining a balance between the efficiency and the adverse effects of the treatment (6). To address this, conducted studies have revealed that a potential strategy in order to increase treatment efficiency is the preparation and application of targeted nano technological systems. Although nano systems have not yet been clinically used for the treatment of bone cancer, studies about the design of multifunctional nanoparticles in the treatment of multiple bone diseases are encouraging.

## RESULTS AND DISCUSSION

### *Bone metastasis and treatment approaches*

The spread of cancer from its initiation site to other parts of the body is called as metastasis. Metastasis occurs when cells separate from the tumor and spread to other parts of the body, either through blood circulation or the lymph ducts, as shown in Figure 1. With this movement, cancer cells can spread to distant tissues and organs. While many cancer cells that separate from the main tumor disappear without causing any harmful effects, some of them settle in new areas and start to proliferate and form new tumors (7). Most of the cancer patients dies from tumor cells that have caused metastasis in other parts of the body (8-10). However, different types of cancer have different dispositions toward different body areas, with the most extensive metastasis observed in the liver, lung, brain and bone (7,11).



**Figure 1.** Metastasis stages of cancerous cells A) Proliferation of the tumor in primary area B) Increase of vascularization and transfer of tumor cells into capillaries C) Participation of tumor cells into blood circulation D) Arrival of tumor cells in secondary area where vascularization and extravasation will occur (figure was used with permission of researchers) (22)

When the cancer spreads to the bone, this is referred to as “bone metastasis”. Some cancers begin in the bone and are called primary bone cancers, such as osteosarcoma, chondrosarcoma, and Ewing’s sarcoma. Especially in adults, bone metastasis is observed more often than bone cancer. Furthermore, tumors that cause breast, lung, and prostate cancer usually metastasize in the bones (6,7,12-15).

The skeletal system is the basic framework of the body and is composed of bone tissue, which consists of a fibrous collagen structure, and calcium phosphate minerals. Hydroxyapatite is the main mineral component of bones. These minerals confer to the bone rigidity and strength for carrying the body by attaching the cells (16-18). Bone tissue harbors two basic cells in its structure-osteoblasts that allow the formation of new bone and osteoclasts that degrade aged bone. The degradation of old cells and the formation of new bone tissue are synchronous processes. Through these two mechanisms, bone tissue preserves its strength (19-21).

It is generally agreed that due to its rich nourishment and stable environment, tumors migrate to bone (23). Additionally, adhesive molecules produced by tumor cells facilitate their binding to the bone. As a result of this adhesion, the production of angiogenic and bone resorbing factors increase, thereby accelerating the proliferation of the tumor in bone tissue. Furthermore, the bones fulfill the storage function for several growth factors. These growth factors are actively released during bone resorption, forming an environment where tumor cells can develop and proliferate. This phenomenon is explained by the “seed and soil” theory (21,24).

The aim of treatment is to prevent tumor growth and eliminate symptoms. Treatment is applied either systemically or locally, with the former preferred in cases where the cancer has spread to more than one bone. However, local treatment is

**Table 1. Advantages and disadvantages of nanosystems used in targeting (28,30,32)**

| Advantages   | Disadvantages  |
|--|--|
| <ul style="list-style-type: none"> <li>- Longer residence time in blood circulation</li> <li>- Ability to localize in the cancerous area because of higher permeation and retention properties of the neoplastic tissue</li> <li>- Reconstructability</li> <li>- Allowance of polymeric nanoparticles to multifaceted modification (such as surface modification with polyethylene glycol [PEG])</li> <li>- Ability to directly adhere to the tumor by employing the ligands, which have different affinities to biological targets in active targeting</li> </ul> | <ul style="list-style-type: none"> <li>- Although nanoparticles have an affinity to accumulate in the tumor area, their efficiency is dependent upon their complete release of the active substance</li> <li>- Weak tumor penetration of nanopharmaceuticals</li> <li>- Complexity of nanoparticle design and active targeting that can be beneficial in a potential application for a specific therapy</li> <li>- Lack of combined nanoformulation applications that comply with the diet combinations that are applied in clinical usage (difficulty preparing a patient-specific combination)</li> <li>- Passage requirement of some biological membranes, muscle tissue and fibroblast-based cell layers between cancer cells and endothelial cells</li> </ul> |

preferred when the cancer has spread to only one bone or a small area. Systemic treatments can be provided through hormone therapy, chemotherapy and the administration of drugs orally or via injection. While radiotherapy and surgical procedures are mostly utilized for local therapy (25), in cancer types that have specifically spread to the bone, radiopharmaceuticals and bisphosphonates are used, and at times, systemic and local therapies are used in conjunction. Targeted drugs follow the cancer cells directly; therefore, healthy normal cells are not damaged during treatment. These types of drugs can also be combined with chemotherapy and hormone therapy (25-28).

#### *Nanosystems and tumor targeting*

Nanotechnological targeting can be performed actively or passively. Some advantages and disadvantages of using nanosystems to target tumor tissue are shown in Table 1. Some of the passively targeted nanopharmaceuticals that have been approved for clinical usage include Myocet®, Doxil®, Daunoxome®, Abraxane® and Genexol-PM® (1,29). Because of being largely impossible to facilitate the accumulation of active substances in the impact area with passive targeting, researches tend to be directed toward actively targeted nano-formulations. With active targeting in cancer therapy, transmitting toxic drugs to healthy tissue is minimized or eliminated, and the therapeutic efficacy of actively targeted nanoparticles is higher than non-targeted nanoparticles (30). Some external stimulants, such as light, heat, ultrasound waves, and magnetic areas are applied in order to release of drug to the areas desired to be affected to eliminate disadvantages of targeted nanosystems. These systems are called triggered release systems (1,31).

#### *Targeting of nano chemotherapeutics to the bone*

Bisphosphonates are the predominant group used in the treatment of bone metastasis and bone targeting (33-35).

Bisphosphonates have high bone affinity and are released into bone tissue by instilling or grafting them over the nanoparticles that are loaded with other active substances (6,36). Table 2 lists some examples of active substances, ligands, and production methods which are used in different nanosystems (nanoparticle, dendrimer, liposome, and nanocomplex) for bone targeting.

Biological activity of bisphosphonates was first discovered in 1968, and they were soon prescribed to treat osteoporosis (21). These drugs effect by slowing down the activity of the osteoclast cells and, therefore, are especially effective in treating osteoclastic metastasis (7,33,37). They are also effective in terms of reducing apoptosis related to cell permeability, and they reduce the capillary vascularization of endothelial tissue (21). Bisphosphonates can reduce cancer related bone pain, bone damage, and high blood calcium concentrations; they are also capable of averting bone fractures (10). Because they show high affinity towards bone, bisphosphonates are generally utilized as imaging agents conjugated with radiopharmaceuticals. Furthermore, they can be conjugated with active substances such as estradiol, prostaglandin, diclofenac, fluoroquinolone, cisplatin, methotrexate, technetium hydroxyethylene diphosphonate, technetium methylene diphosphonate, samarium, leixidronam. Peptides and proteins have also been conjugated with bisphosphonates for bone targeting (10,14,21,43-45).

Utilizing bisphosphonates for the targeting of nanoparticulate systems has many advantages. Nitrogenous bisphosphonates contain primer amines and carboxylic acid. If they are conjugated with the nanoparticle via degradable bonds, they become pharmaceutically active and will even show synergy with a suitable active substance (21).

**Table 2. Bone targeted nano systems**

| Formulations                                  | Ligands                              | Active ingredients               | Polymer/Lipid  | Production method        | Particle size       | Ref. |
|---|--------------------------------------|----------------------------------|--|--------------------------|---------------------|------|
| Nano structured ceramic implants              | Calcium phosphate and hydroxyapatite | -                                | -  | Precipitation            | 8.84 µm and 5.21 µm | (38) |
| Nano complex                                  | Hydroxyapatite                       | BMP-2<br>Titanium                | -  | Ionic interaction        | 150-250 nm          | (39) |
| Nanoparticle                                  | Hydroxyapatite                       | -                                | -  | Precipitation            | 20-40 nm            | (18) |
| Nanoparticle                                  | Alendronate                          | Estrogen                         | Polyethylene glycol- Poly(lactic-co-glycolic acid) block copolymer   | Dialysis method          | 43.5-57.3 nm        | (40) |
| Dendrimer                                     | Alendronate                          | Paclitaxel                       | Hydroxypropyl methacrylamide (HPMA)  | Conjugation              | 95 nm               | (41) |
| Dendrimer                                     | Alendronate                          | Paclitaxel                       | Polyethylene glycol  | Conjugation              | 190 nm              | (23) |
| Radiopharmaceutics and bisphosphonate complex | Hydroxyethylidene-1,1-diphosphonate  | Rhenium-186 ( <sup>186</sup> Re) | -  | Conjugation              | -                   | (42) |
| Liposome                                      | RGD peptide                          | Cisplatin                        | Phosphatidylcholine, cholesterol, 1,2-distearoyl-sn-glycero-3-phosphoethanolamine-N-[maleimide (polyethylene glycol)-2000] | Extrusion of suspensions | 214.7 nm            | (9)  |



Zoledronic acid (ZOL) is a nitrogen-containing bisphosphonate with anticancer activity. ZOL increases apoptosis, prevents the formation of new veins that will ensure the blood supply to the tumor, decreases the level of vascular endothelial growth factor (VEGF), and decreases substance adhesion activity towards the tumor and osteoclast cells. Chaudhari et al. developed targeted drug release systems for bone metastasis containing docetaxel loaded nanoparticles formed from PLGA (poly(lactic-co-glycolic acid)), which have been approved by the FDA for use with ZOL on humans. In this study, nanoparticles were prepared by forming a PLGA-PEG-ZOL conjugate. According to the results obtained from the *in vitro* drug release experiments, the release model is biphasic. The initial burst release was due to the release of drug located on and near the surface of the nanoparticles. In the second phase, the release was slower and controlled by diffusion rate of drug across the polymer matrix. In addition, as a result of the bone affinity tests, no significant differences were observed in bone adhesion between the conjugate nanoparticles and the solutions containing only ZOL. The distribution of radiolabeled particles among the body and bone affinity of these particles were compared with the blood concentrations of the PLGA-PEG-ZOL nanoparticles. At the end of 24 hours, it was observed that 50 times more PLGA-PEG-ZOL conjugate was present in the bone compared to PLGA-PEG nanoparticles. Furthermore, the bone adhesion ratio of the PLGA-PEG-ZOL nanoparticles was found 3.5 times increased compared to normal bone. The authors postulated that the reason for this phenomenon was the loss of lining wherein surrounds the healthy bone in diseased bone tissue. This lining prevents zoledronate from binding, thus its affinity to diseased tissue (osteoporosis, bone metastasis, etc.) is much higher (10).

Li et al. stated that prostate cancer, believed to be osteoblastic, shows osteolytic character at the initial stage of bone metastasis. Therefore, they argued that by targeting to the bone metastasis area that contains both osteolytic and osteoblastic drug molecules, the bone tumor related alterations will be prevented (14). In another study, researchers produced nanoparticles conjugated with alendronate (ALE) for calcification, which is a pathological bone metastasis *in vivo*. Calcification forms as a result of the deposition of calcium in normal tissues, such as veins, nerves and breast tissues, which is similar to embryonic osteogenesis. Generally, lesions formed as a result of calcification are benign; however, in prostate and breast cancer, this situation is reversed. It has been observed that surface of the nanoparticles formed for this purpose is primarily covered with polydopamine, which provides adhesive properties, and conjugation of the particle surface is facilitated. Then, ALE can easily be conjugated with the particle surface, ensuring that it will act as both an active substance and a targeting agent (46).

In a study conducted by Clementi et al. (2011), ALE and paclitaxel (PTX) were prepared in dendrimer form together

and separately, and conjugates were obtained. The purpose of this study was to investigate the effect of PEG when used as a carrier on cytotoxic PTX and ALE. Active targeting to the bone by virtue of ALE and passive targeting with an endothelial permeation retention effect were successful. In addition, by increasing the dimensions of the PEG conjugation relative to the free drug, the deposition in the tumor tissue increased (23). In another study by the same research group, bone metastasis caused by breast cancer was targeted using micelles with a dendrimer structure. Cell culture experiments were performed using preparation of the same conjugates over MDA-MB-231 human breast cancer cells and 4T1 mouse breast cancer cells. For both cell types, conjugate formation did not cause any loss of effect. In addition, *in vivo* antitumor activity determination study showed that after 15 days, the highest tumor healing could be provided with the PTX-PEG-ALE conjugate. Deposition in the tumor was monitored by using a non invasive monitoring system, and at the end of 8 hours, the PTX-PEG-ALE conjugate showed the highest deposition. When the distribution in the body was investigated, it was found that the PTX-PEG-ALE conjugate bound to the tumor with a higher ratio than the conjugate formed between PTX and ALE or PTX and PEG (12).

In another study, H40-star-PEG/ALE micelles were obtained and targeted delivery of doxorubicin to the bone was successful. In this micelle, H40 polyester formed the hydrophobic nucleus part and served as storage for the hydrophobic active substance. ALE was the bone targeted fragment. PEG formed the hydrophobic arms of the star shaped micelle. Release profiles of the micelles loaded with doxorubicin were investigated at pH 4 and pH 7.4. For both pH, the burst effect was quiet fast. The release rate observed in pH 5 was faster than the physiological pH. The reason for these differences was protonation of the amino group of doxorubicin in an acidic environment and low speed degradation of the micelle nucleus at low pH. The authors stated that this drug release rate was promising for pH dependent release systems to bone tumors (43).

Miller et al. aimed to facilitate the targeting of PTX to bone by utilizing ALE. Owing to conjugation with the N-(2-Hydroxypropyl) methacrylamide (HPMA) copolymer, PTX was targeted to metastatic bone regions. It was determined that the passage of the conjugate from the leaking tumor area and its exit from the exterior of the vein were passive. Furthermore, a normal blood vein would not allow the conjugate to pass due to its size; thus, it was ensured that the conjugate would only target the tumor area. HPMA can be dissolved in water and is among the carriers that are biocompatible and non-toxic, and it does not stimulate the immune system. These macromolecules do not diffuse in normal blood veins, but they selectively accumulate in tumor areas because of the increased permeability and retention (IPR) effect. In addition, since the blood brain barrier cannot be overcome with this conjugation, the neurotoxic side

effects formed by PTX are prevented, and residence time in the circulation is extended (41).

Currently, the most popular nano systems for local substance release are magnetic nanoparticles. The disadvantage of these methods is low target cell selectivity, and the requirement to place the patient in a magnetic area for long periods of time (23). One study observed the effect of magnetic nanoparticles on *in vitro* osteoblasts, and according to the cell culture results collected on the fifth and eighth days, it was found that  $\gamma\text{-Fe}_2\text{O}_3$  magnetic nanoparticles significantly increased osteoblast density compared to the control group. Then, these nanoparticles were coated with calcium phosphate for the treatment of bone disorders, and in order to prevent agglomeration, were distributed in bovine serum albumin or citric acid. With the presence of this coating inside the bovine serum albumin, by day one, osteoblast density had significantly increased compared to the control group (47).

The performed studies show that modifying the polymeric structure does not significantly increase the targeting binding ratio of processes to the bone tissue, but the presence of the higher binding affinity ligands increased the adsorption. Moreover, it was stated that multiple types of ligand conjugations give better bindings. However, increasing number of the ligands in the nanostructure and increasing the ligand size are not advantageous, once the size of nanoparticles is increased then nanoparticles cannot be used in every formulation (21,48).

When metastasis is generated in bone, acid base balance of the tissue where metastasis is formed is disturbed, due to the protons and hydrolases that are given to the medium by the osteoclast cells. Here, the organic and mineral structure of the bone matrix disintegrates, and the pH decreases to 4.5 in the medium where bone resorption occurs. Because of this, pH sensitive bone specific drug release systems have drawn

considerable attention. Ye et al. obtained ALE-monoethyl adipate-(hydrazone)-doxorubicin conjugate by conjugating monoethyl adipate, ALE, and doxorubicin with an amide bond and a hydrazone bond, respectively. It was determined that with conjugation, binding of doxorubicin to the bone increased significantly. In addition, the authors found that in *in vitro* release tests conducted under different pH conditions, a pH dependent release was observed in conjugates with the hydrazone bond, but not in conjugates with the amide bond (49).

#### Bone binding and drug release

In bone targeted systems, *in vitro* binding tests are performed by determining the binding of particles to hydroxyapatite. These tests are generally based on the addition of a nanoparticle suspension over bone powder or hydroxyapatite in a test tube. Then, the mixture is incubated, and after a predetermined period of time, centrifuged. The particular system is then analyzed inside the supernatant using a suitable analysis method (48). A binding test was performed using the above mentioned method. Figure 2 shows a graph comparing the bone binding capacities of micelles that were conjugated with ALE and those which were not conjugated with ALE. It shows that the bone binding ratio of the nano sized conjugate formed with ALE was 70% after 5 minutes, while the conjugates not bound with ALE showed no significant binding (43).

In another study bone binding kinetics of nanoparticles conjugated with ZOL and ZOL solution was investigated and it was determined that the utilization of ZOL as nanoparticle surface ligand did not cause any effect loss in terms of bone binding (10).

*In vitro* drug release studies are generally performed in buffer mediums that mimic 37°C, lysosomal (pH 5), and physiological mediums (pH 7.4) with the incubation of a sufficient amount of conjugate in a proper way (23,43). Chen et al. investigated the

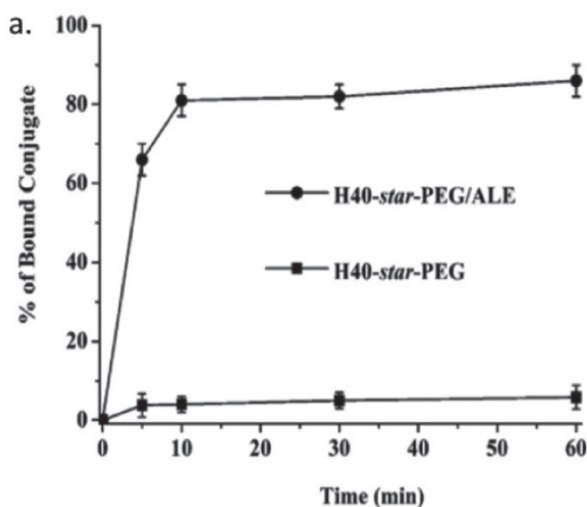


Figure 2. H-40-star-PEG/ALE and H-40-star-PEG micelles *in vitro* bone binding kinetics (figure was used with permission of researchers) (43)

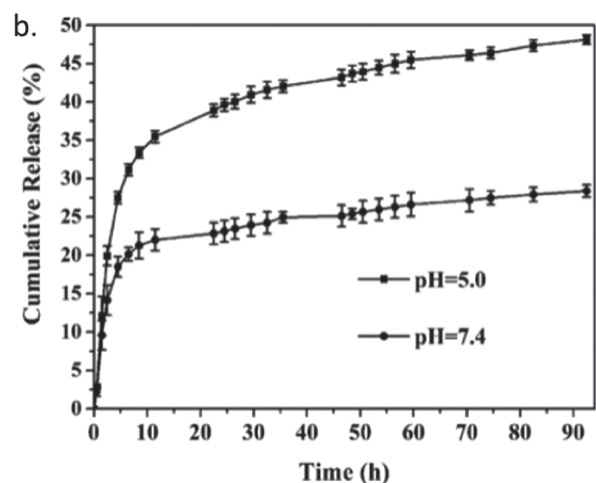


Figure 3. *In vitro* active substance release graphic of doxorubicin loaded Boltorn H40-PEG conjugate, which was conjugated with ALE in pH 5.0 vs. pH 7.4 buffer mediums (figure was used with permission of researchers) (43)

*in vitro* drug release rate of doxorubicin loaded Boltorn-H40-PEG conjugate, which was conjugated with ALE by incubating it in pH 5 and pH 7.4 buffer mediums. According to the obtained results (Figure 3), pH-dependent drug release was observed in the established particulate. These results showed that the obtained micelles could provide more efficient treatment at the tumor containing area of the bone (43).

Although the *in vitro* drug release studies will give information about release rate, the antineoplastic effect in the metastatic tissue cannot be analyzed in these tests. In addition, even though a reduction in the tissue containing tumor can be observed in the *in vivo* experiments, whether the anticancer agent harms the healthy cells cannot be determined or not. Therefore, conducting cell culture experiments is important to show that targeting can be performed. It is important to prove that the nanoparticles do not show a non-proliferative effect on healthy cells, with regard to showing that the prepared system does not harm healthy tissue. For non healthy cells that develop bone metastasis, generally sarcoma, breast or prostate cancer cells are used (9,23,46,50).

## CONCLUSION

As a result of the review we performed, it is determined that the most frequently used approach for targeted delivery of anticancer agents to areas containing tumors in the bone is conjugation of bisphosphonates to nano particulate systems. Bisphosphonates' affinity to hydroxyapatite shows that these molecules are ideal molecules for bone targeting. This conjugation was generally formed by conjugating bisphosphonates to the particle surface. The usability of bisphosphonates as active substances, especially in osteoclastic metastasis, make these molecules advantageous. Concurrently, it was shown in many studies that both the targeting and the anticancer effect of the active substance of bisphosphonates, such as zoledronate, ZOL and ALE, which were bound to nanoparticles loaded with active substances used for the treatment of cancer, can be increased with a synergic effect. Nano particulate systems that are conjugated with bisphosphonates and/or other ligands with bone affinity are promising therapies in the treatment of bone cancer or bone metastasis, as well as other bone diseases.

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## 2016 Author Index - 2016 Yazar Dizini

|                               |        |                                   |             |
|-------------------------------|--------|-----------------------------------|-------------|
| A. OLATUNJI Gabriel.....      | 41     | E AL-DHUBIAB Bandar.....          | 213         |
| AĞIN Fatma.....               | 335    | ECE Abdulilah.....                | 241         |
| AHMAD Farediah.....           | 347    | ERGENE ÖZ Burçin.....             | 115         |
| AKTAY Göknur.....             | 135    | ERYILMAZ Müjde.....               | 35          |
| AKTÜMSEK Abdurrahman.....     | 323    | G.R Godge.....                    | 91          |
| ALAN Sevim.....               | 300    | GANESH Mani.....                  | 189         |
| ALEMDAR Mihriban.....         | 57     | GHAFFARI Muhammad Abuzar.....     | 259         |
| ALTINYAY Çiğdem.....          | 115    | GHORI Muhammad Khurm.....         | 259         |
| ALTUN Mehmet Levent.....      | 115    | GÖGER Gamze.....                  | 173         |
| ANLAR Hatice Gül.....         | 27     | GÖKBULUT Alper.....               | 159         |
| ARISOY Gülsüm Gül.....        | 292    | GÖRGÖZ Miray.....                 | 351         |
| ARISOY Mustafa.....           | 292    | GÜRSOY Şule.....                  | 135         |
| ARMAGAN Güliz.....            | 103    | GÜVENDİK Gülin.....               | 292         |
| ASAKAWA Yoshinori.....        | 285    | HAMEED Hasan.....                 | 225         |
| ATAKCI Sumeyra.....           | 167    | HONARMAND Masoumeh.....           | 233         |
| ATİLA KARACA Sakine.....      | 17     | HUSSAIN Sajid Nawaz.....          | 259         |
| ATOLANI Olubunmi.....         | 41     | ILGÜN Selen.....                  | 201         |
| AYAZ Fatma.....               | 342    | IZADI Arezoo.....                 | 233         |
| AYDIN Ahmet.....              | 167    | İLHAN Mert.....                   | 323         |
| AYDIN Sevtap.....             | 225    | İNCESU Zerrin.....                | 318         |
| AZIZ Ashima.....              | 81     | İŞCAN Gökcalp.....                | 318         |
| AZIZ Mukhayar.....            | 259    | JANG Hyun Tae.....                | 189         |
| BACANLI Merve.....            | 27     | K.A Rupnar.....                   | 91          |
| BASKAK Bora.....              | 145    | K.K Pawa.....                     | 91          |
| BAŞARAN Arif Ahmet.....       | 27,225 | KANIT Lutfiye.....                | 103         |
| BAŞARAN Nurşen.....           | 27,225 | KARAKAYA Songül.....              | 173         |
| BHUIYAN Mohiuddin Ahmed.....  | 81     | KARATOPRAK Gökçe Şeker.....       | 201         |
| BNS Sailakshmi.....           | 304    | KARUNA Dasari Subramaniyam.....   | 189         |
| BÜYÜKKÖROĞLU Gülay.....       | 249    | KELEŞ Hikmet.....                 | 135,285     |
| CHAHARDAHCHERIK Marjan.....   | 233    | KILIÇ C. Sibel.....               | 173         |
| CHAREHSAZ Mohammad.....       | 167    | KILIÇARSLAN Müge.....             | 351         |
| CHISTY Sharmin Jahan.....     | 81     | KOŞAR Müberra.....                | 201         |
| DAMINENI Saritha.....         | 304    | KUMAR Jayaraja.....               | 9           |
| DEĞİM Zelihagül.....          | 69     | KURANEL Esin.....                 | 135         |
| DELİORMAN ORHAN Didem.....    | 181    | KUTLUK Berkin.....                | 27          |
| DEMİRCİ Betül.....            | 173    | KÜÇÜK Sevim.....                  | 318         |
| DEVİRİM Burcu.....            | 57     | KÜÇÜKBOYACI Nurgün.....           | 342         |
| DEVİRİMCİ OZGUVEN Halise..... | 145    | KÜPELİ AKKOL Esra.....            | 135,285,323 |
| DICKO Amadou.....             | 47     | KÜRKÇÜOĞLU Mine.....              | 300         |
| DJILANI Abdelouaheb.....      | 47     | LITVINOVA Elena Vyacheslavna..... | 121         |
| DJILANI Salah Eddine.....     | 47     | MANZOOR Majid.....                | 259         |
| DURAL Emrah.....              | 292    | MERGEN Görkem.....                | 292         |

## 2016 Author Index - 2016 Yazar Dizini

|                                       |         |
|---------------------------------------|---------|
| MURALIDHARAN Selvadurai.....          | 9       |
| NAMAZI Fatemeh.....                   | 328     |
| NAZIFI Saeed.....                     | 233,328 |
| ORHAN Nilüfer.....                    | 181     |
| ÖZDEMİR Hatice.....                   | 145     |
| ÖLÇEK Rıza Can.....                   | 167     |
| ÖZDEMİR Filiz.....                    | 318     |
| ÖZEL KIZIL Erguvan Tuğba.....         | 145     |
| ÖZENOĞLU KİREMİT Hatice.....          | 285     |
| PARASURAMAN Subramani .....           | 9       |
| PENJURI Subhash Chandra Bose .....    | 304     |
| POREDDY Srikanth Reddy .....          | 304     |
| POSYLKINA Olga Viktorovna.....        | 121     |
| RAFIEI Maryam.....                    | 328     |
| RAJAIAN Hamid .....                   | 328     |
| RATHNAM Grace .....                   | 189     |
| RAVOURU Nagaraju .....                | 304     |
| S.Z Chemate.....                      | 91      |
| SAJEDIANFARD Javad .....              | 233     |
| SALLEH Wan Mohd Nuzul Hakimi Wan..... | 347     |
| SARIMAHMUT Mehmet .....               | 342     |
| SARWER Wajid.....                     | 259     |
| SERDAROĞLU Vildan .....               | 335     |
| SEVGİLİ Elvin .....                   | 103     |

|                             |             |
|-----------------------------|-------------|
| SİPAHİ Hande .....          | 167         |
| SÖYLEMEZOĞLU Tülin.....     | 292         |
| SUZEN Halit Sinan.....      | 145         |
| SÜNTAR İpek .....           | 135,285,323 |
| ŞENER Erol .....            | 17          |
| ŞENER Görkem .....          | 300         |
| TOSHA Sumaiya Mehjabin..... | 81          |
| TOSUN Alev .....            | 35,285      |
| TURUNC Ezgi .....           | 103         |
| TÜMEN İbrahim .....         | 35          |
| UCKUN Zuhal.....            | 145         |
| UDHUMANSHA Ubaidulla.....   | 189         |
| ULUKAYA Engin.....          | 342         |
| ÜSTÜNDAĞ OKUR Neslihan..... | 311         |
| VENUGOPAL Vijayan .....     | 9           |
| YALCIN Ayfer .....          | 103         |
| YENİCELİ UĞUR Duygu.....    | 17          |
| YILMAZ Işıl Tan.....        | 17          |
| YILMAZ Şükran.....          | 69          |
| YOLTAŞ Ayşegül.....         | 311         |
| YOZGATLI Vildan .....       | 311         |
| YÜCEL Çiğdem.....           | 69          |
| ZEGHAD Farida.....          | 47          |
| ZENGİN Gökhan.....          | 323         |

## 2016 Subject Index

|   |               |   |                  |
|---|---------------|---|------------------|
| 5-HTTLPR polymorphism.....                  | 145           | <i>Conyza Canadensis</i> .....              | 342              |
| 7-Hydroxy Quetiapine.....                   | 17            | <i>Copaene</i> .....                        | 41               |
| Acyclovir.....                              | 213           | Cross-linked hard gelatin capsule.....      | 81               |
| <i>Allium porrum</i> .....                  | 181           | <i>Cupressaceae</i> .....                   | 35               |
| <i>Alnus</i> .....                          | 115           | <i>Cydonia oblonga</i> .....                | 181              |
| <i>Alseodaphne peduncularis</i> .....       | 347           | Cytotoxic activity.....                     | 342              |
| Amlodipine besylate.....                    | 81            | Cytotoxicity.....                           | 27,41,69,225,318 |
| Anticandidal activities.....                | 318           | D-serine.....                               | 103              |
| Anticholinesterase activity.....            | 347           | <i>Daniellia oliveri</i> .....              | 41               |
| Antimicrobial.....                          | 35,47         | Determination.....                          | 335              |
| Antioxidant activity.....                   | 201,347       | Diclofenac sodium.....                      | 292              |
| Antioxidant.....                            | 41,47,181,328 | Diterpenoids.....                           | 41               |
| Apiaceae.....                               | 173           | DNA damage.....                             | 103              |
| Asphodeline.....                            | 323           | Docking.....                                | 241              |
| Asteraceae.....                             | 342           | DPPH.....                                   | 47               |
| Base excision repair.....                   | 103           | Drug targeting.....                         | 351              |
| Betulaceae.....                             | 115           | Drug.....                                   | 121              |
| Bioautography method.....                   | 173           | E-pharmacophore.....                        | 241              |
| Biological activities.....                  | 259           | Elastase.....                               | 323              |
| Bisphosphanates.....                        | 351           | Emulsification methods.....                 | 57               |
| Bone metastasis.....                        | 351           | Essential oil.....                          | 300,347          |
| Boraginaceae.....                           | 259           | Ethanol.....                                | 328              |
| Breast cancer.....                          | 27            | Ethylcellulose.....                         | 304              |
| Bryophytes.....                             | 285           | Eudragit.....                               | 213              |
| Buccal delivery.....                        | 213           | Euphorbia.....                              | 47               |
| Buoyancy.....                               | 91            | Ex vivo.....                                | 213              |
| Busulfan.....                               | 233           | Extract.....                                | 35               |
| Caco-2 cell line.....                       | 69            | Extrusion- spheronization.....              | 189              |
| Cadinene.....                               | 41            | <i>Ferulago blancheana</i> .....            | 173              |
| Calcium pectinate beads.....                | 91            | Fickian release.....                        | 304              |
| Cancer.....                                 | 241           | Flavonoids.....                             | 47               |
| Carbon paste electrode.....                 | 335           | Floating-pulsatile drug delivery.....       | 91               |
| Cationic carrier system.....                | 249           | Gastric ulcer.....                          | 328              |
| Cervical cancer.....                        | 27            | Gastroprotective.....                       | 328              |
| Characterization.....                       | 311           | GC/MS.....                                  | 173,300          |
| Chinese hamster lung fibroblasts cells..... | 225           | Gene delivery system.....                   | 249              |
| Chitosan succinate pellets.....             | 189           | Genus <i>Heliotropium</i> .....             | 259              |
| Chitosan.....                               | 189           | Glutathione.....                            | 135              |
| Chronotherapy.....                          | 91            | <i>Helianthus tuberosus</i> .....           | 181              |
| Cisplatin.....                              | 69            | High performance liquid chromatography..... | 292              |
| Citalopram.....                             | 145           | <i>Hispidulin</i> .....                     | 159              |
| Collagenase.....                            | 323           | Histopathology.....                         | 328              |

## 2016 Subject Index

|                                     |                  |                                     |         |
|-------------------------------------|------------------|-------------------------------------|---------|
| HPLC.....                           | 9,17,115,159,201 | Plasma.....                         | 292     |
| HPTLC.....                          | 159              | PLGA microparticles.....            | 57      |
| Human cervical carcinoma cells..... | 225              | Pluronic F-68.....                  | 304     |
| Human plasma.....                   | 9                | Polyphenols.....                    | 47      |
| Hyaluronidase.....                  | 323              | Pulmonary delivery.....             | 57      |
| Hydrogel plug.....                  | 81               | Pulsatile drug delivery.....        | 81      |
| Hypoglycemic.....                   | 181              | Quantitative analysis.....          | 115     |
| In situ gel.....                    | 311              | Quetiapine Sulfoxide.....           | 17      |
| In vivo.....                        | 213              | Quetiapine.....                     | 17      |
| Inula viscosa.....                  | 159              | Rat plasma.....                     | 17      |
| Juniperus oxycedrus.....            | 300              | Respiratory tract infections.....   | 57      |
| Lansoprazole.....                   | 304              | Resveratrol.....                    | 27      |
| Lauraceae.....                      | 347              | RSK.....                            | 241     |
| Levofloxacin hemihydrate.....       | 57               | Safety.....                         | 167     |
| Liposome.....                       | 69               | Salvia virgata.....                 | 201     |
| Liverworts.....                     | 285              | Scrophularia striata.....           | 328     |
| Lung cancer.....                    | 342              | Scrophulariaceae.....               | 318     |
| Marchantiophyta.....                | 285              | Secondary metabolites.....          | 259     |
| Mefenamic acid.....                 | 103              | Shikimic acid.....                  | 115     |
| Microbiological study.....          | 311              | Side effects.....                   | 145     |
| MTT assay.....                      | 27               | Silymarin.....                      | 233     |
| Multiwalled carbon nanotubes.....   | 335              | Sinapic acid.....                   | 225     |
| Nano chemotherapeutics.....         | 351              | Solid lipid nanoparticle (SLN)..... | 249     |
| Nanoparticles.....                  | 351              | Sulforhodamine B assay.....         | 342     |
| Nanosponges.....                    | 304              | Sustained release.....              | 189     |
| Nanotechnology.....                 | 121              | Tap water.....                      | 167     |
| Neutral red uptake assay.....       | 225              | Tiobarbituric acid.....             | 135     |
| Nimesulide.....                     | 335              | Treatment response.....             | 145     |
| Nitrate.....                        | 167              | Triamcinolone.....                  | 9       |
| Nitrite.....                        | 167              | Ultraviolet detection.....          | 292     |
| NRU assay.....                      | 27               | Validation.....                     | 9,292   |
| Ocular drug delivery.....           | 311              | Verbascum.....                      | 318     |
| Optimization.....                   | 9                | Virtual screening.....              | 241     |
| Oxidative stress.....               | 233              | Volatile oil.....                   | 173     |
| Patent research.....                | 121              | Voltammetry.....                    | 335     |
| pH sensitive drug release.....      | 189              | Voriconazole.....                   | 311     |
| Pharmacognostic characters.....     | 259              | Wound healing.....                  | 135,285 |
| Phenolic compounds.....             | 225              | Xanthorrhoeaceae.....               | 323     |
| Pinaceae.....                       | 35               | Zero order.....                     | 304     |
| Plantaginaceae.....                 | 135              |                                     |         |
| Plantago lanceolata.....            | 135              |                                     |         |



## 2016 Konu Dizini

|  |               |                                      |         |
|--|---------------|--------------------------------------|---------|
| 5-HTTLPR polimorfizmi .....                    | 145           | Ekstrüzyon-sferonizasyon .....       | 189     |
| 7-Hidroksi Ketiapin .....                      | 17            | Elastaz .....                        | 323     |
| Akciğer kanseri .....                          | 342           | Emülsifikasyon yöntemleri .....      | 57      |
| Allium porrum .....                            | 181           | Etanol .....                         | 328     |
| Alnus .....                                    | 115           | Etilselüloz .....                    | 304     |
| Alseodaphne peduncularis .....                 | 347           | Eudragit .....                       | 213     |
| Amlodipin besilat .....                        | 81            | Euphorbia .....                      | 47      |
| Antikandidal aktivite .....                    | 318           | Ex vivo .....                        | 213     |
| Antikolinesteraz aktivite .....                | 347           | Farmakognozok karakterler .....      | 259     |
| Antimikrobiyal .....                           | 35,47         | Fenolik bileşikler .....             | 225     |
| Antioksidan .....                              | 41,47,181,328 | Ferulago blancheana .....            | 173     |
| Antioksidan aktivite .....                     | 201,347       | Fick salami .....                    | 304     |
| Apiaceae .....                                 | 173           | Flavonoitler .....                   | 47      |
| Asiklovir .....                                | 213           | Gastrik ülser .....                  | 328     |
| Asphodeline .....                              | 323           | Gastroprotektif .....                | 328     |
| Asteraceae .....                               | 342           | GC/MS .....                          | 173,300 |
| Baz eksizyon tamiri .....                      | 103           | Gen taşıyıcı sistem .....            | 249     |
| Betulaceae .....                               | 115           | Glutatyon .....                      | 135     |
| Bifosfonatlar .....                            | 351           | Güvenlik .....                       | 167     |
| Biyolojik aktiviteler .....                    | 259           | Helianthus tuberosus .....           | 181     |
| Biyootografi yöntemi .....                     | 173           | Heliotropium cinsi .....             | 259     |
| Boraginaceae .....                             | 259           | Hidrojel kapak .....                 | 81      |
| Bryofitler .....                               | 285           | Hipoglisemik .....                   | 181     |
| Bukkal salım .....                             | 213           | Hispidulin .....                     | 159     |
| Busulfan .....                                 | 233           | Histopatoloji .....                  | 328     |
| Caco-2 hücre hattı .....                       | 69            | Hyalüronidaz .....                   | 323     |
| Ciğer otları .....                             | 285           | Inula viscosa .....                  | 159     |
| Cisplatin .....                                | 69            | İlaç .....                           | 121     |
| Conyza canadensis .....                        | 342           | İlaç hedefleme .....                 | 351     |
| Cupressaceae .....                             | 35            | İn situ jel .....                    | 311     |
| Cydonia oblonga .....                          | 181           | İn vivo .....                        | 213     |
| Çapraz bağlı sert jelatin kapsül .....         | 81            | İnsan plazması .....                 | 9       |
| Çin hamster akciğer fibroblast hücreleri ..... | 225           | İnsan serviks kanser hücreleri ..... | 225     |
| Çok duvarlı karbon nanotüp .....               | 335           | Juniperus oxycedrus .....            | 300     |
| D-serin .....                                  | 103           | Kadinen .....                        | 41      |
| Daniellia oliveri .....                        | 41            | Kalsiyum pektinat boncuklar .....    | 91      |
| Diklofenak sodium .....                        | 292           | Kanser .....                         | 241     |
| Diterpenoidler .....                           | 41            | Kantitatif analiz .....              | 115     |
| DNA hasarı .....                               | 103           | Karakterizasyon .....                | 311     |
| DPPH .....                                     | 47            | Karbon pasta elektrot .....          | 335     |
| E-farmakofofor .....                           | 241           | Katı lipid nanopartikül (KLN) .....  | 249     |
| Ekstre .....                                   | 35            | Katyonik taşıyıcı sistem .....       | 249     |

## 2016 Konu Dizini

|  |     |
|--|-----|
| Kemik metastazı.....                   | 351 |
| Kenetlenme .....                       | 241 |
| Ketiapin.....                          | 17  |
| Ketiapin Sülfoksit .....               | 17  |
| Kitozan.....                           | 189 |
| Kitozan süksinat pelletler .....       | 189 |
| Kollajenaz.....                        | 323 |
| Kopaen.....                            | 41  |
| Kronoterapi.....                       | 91  |
| Lansoprazol.....                       | 304 |
| Lauraceae.....                         | 347 |
| Levofloksasin hemihidrat .....         | 57  |
| Lipozom.....                           | 69  |
| Marchantiophyta .....                  | 285 |
| Mefenamik asit.....                    | 103 |
| Meme kanseri.....                      | 27  |
| Mikrobiyolojik çalışma .....           | 311 |
| MTT yöntemi .....                      | 27  |
| Musluk suyu.....                       | 167 |
| Nano kemoterapotikler .....            | 351 |
| Nanopartiküller.....                   | 351 |
| Nanosünger .....                       | 304 |
| Nanoteknoloji.....                     | 121 |
| Neutral kırmızı alım yöntemi.....      | 225 |
| Nimesulid .....                        | 335 |
| Nitrat.....                            | 167 |
| Nitrit.....                            | 167 |
| NKA yöntemi .....                      | 27  |
| Oksidatif stres.....                   | 233 |
| Oküler ilaç dağılımı.....              | 311 |
| Optimizasyon.....                      | 9   |
| Patent araştırması .....               | 121 |
| pH'ya duyarlı etken madde salımı ..... | 189 |
| Pinaceae.....                          | 35  |
| Plantaginaceae.....                    | 135 |
| Plantago lanceolata .....              | 135 |
| Plazma .....                           | 292 |
| PLGA mikropartikülleri .....           | 57  |
| Pluronic F68 .....                     | 304 |
| Polifenoller .....                     | 47  |
| Pulmoner veriliş.....                  | 57  |

|   |                  |
|---|------------------|
| Pulsatil ilaç salımı .....                    | 81               |
| Resveratrol .....                             | 27               |
| RSK.....                                      | 241              |
| Salvia virgata.....                           | 201              |
| Sanal tarama.....                             | 241              |
| Scrophularia striata .....                    | 328              |
| Scrophulariaceae .....                        | 318              |
| Sekonder metabolitler.....                    | 259              |
| Servikal kanser .....                         | 27               |
| Sıçan plazması.....                           | 17               |
| Sıfır derece .....                            | 304              |
| Silimarin .....                               | 233              |
| Sinapik asit.....                             | 225              |
| Sitalopram .....                              | 145              |
| Sitotoksik aktivite .....                     | 342              |
| Sitotoksosite .....                           | 27,41,69,225,318 |
| Solunum yolu enfeksiyonları.....              | 57               |
| Sülforodamin B testi .....                    | 342              |
| Süreklı salım .....                           | 189              |
| Şikimik asit.....                             | 115              |
| Tayın .....                                   | 335              |
| Tedavi yanıtı .....                           | 145              |
| Tiyobarbitürik asit .....                     | 135              |
| Triamsinolon.....                             | 9                |
| Uçucu Yağ.....                                | 173,300,347      |
| Ultraviyole deteksiyon .....                  | 292              |
| Validasyon.....                               | 9,292            |
| Verbascum.....                                | 318              |
| Voltametri .....                              | 335              |
| Vorikonazol.....                              | 311              |
| Yan etkiler.....                              | 145              |
| Yara iyileşmesi.....                          | 135,285          |
| YPİTK.....                                    | 159              |
| YPSK.....                                     | 9,17,115,159,201 |
| Yüksek performanslı sıvı kromatografisi ..... | 292              |
| Yüzen-pulsatil ilaç salımı .....              | 91               |
| Yüzme.....                                    | 91               |
| Xanthorrhoeaceae.....                         | 323              |